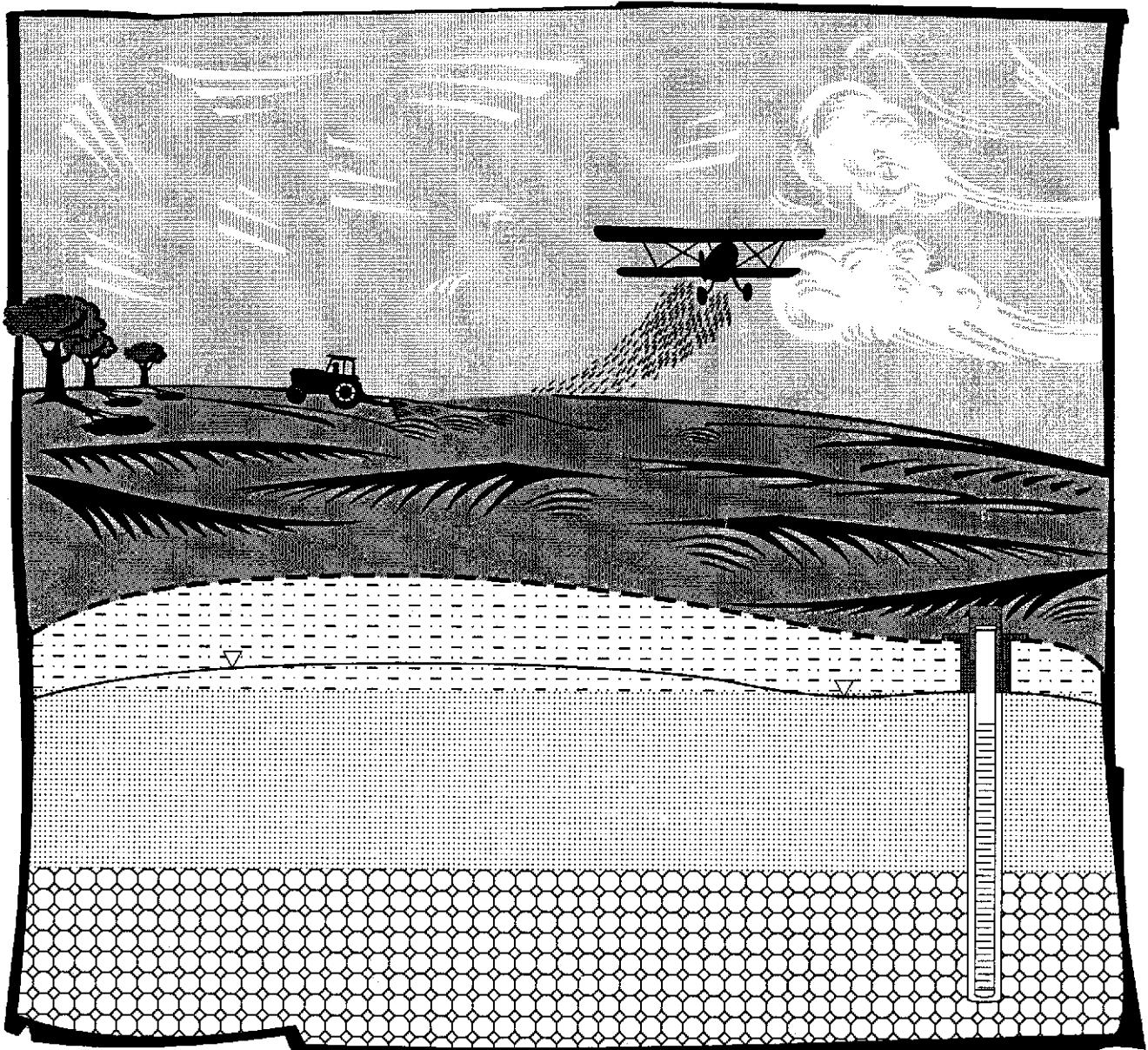


OCCURRENCE OF PESTICIDES IN ALLUVIAL AQUIFER OF EASTERN ARKANSAS



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INTRODUCTION

The monitoring of pesticides has gained increased attention in Arkansas since the early 1990's with the federal mandates for each state to develop a State Management Plan (SMP) for pesticide use (EPA, 1992a; 1992b). In addition, the federal 319 Nonpoint Source Program has provided a source of revenue for assessing both surface and subsurface water sources for nonpoint source impacts, including pesticides. As part of the 319 program, 1996 nonpoint source water-quality assessment, the Department of Pollution Control & Ecology (Department) sampled 77 wells in three areas that are within or near to the borders of three Arkansas counties: Jefferson, Desha and Phillips.

Because pesticide and fertilizer use are the dominant contributors to nonpoint impacts within Arkansas' agricultural community, 61 pesticides and various degradation products were analyzed by the Department laboratory in addition to ammonia, nitrate, major cations and anions, and dissolved metals. The wells were sampled between July 29, 1996, and September 23, 1996. The wells are used exclusively for irrigation and average 80+ feet in depth. Pesticide use in the area is directed primarily to the control of weeds in soybean, rice and cotton fields.

The following report discusses the general water quality of the three study areas and the nonpoint source impacts resulting from agricultural practices in the areas. For additional information on the water quality of both surface and subsurface water and nonpoint source impacts, refer to the 1997 Department report entitled "Arkansas' Nonpoint Source Pollution Assessment Report"

SITE CHARACTERISTICS

LOCATION

The main focus of the present study was to investigate potential nonpoint source impacts to ground water, primarily in the form of pesticide contamination, by sampling irrigation wells in an agricultural area. Because pesticides have been monitored on a continual basis by the Department as part of the state's ambient ground-water monitoring program and also by the Arkansas Water Resource Center (Nichols et al., 1993; Steele et al., 1993; Steele et al., 1994; Nichols et al., 1995; and Nichols et al., 1996) as part of the SMP monitoring program, Department personnel focused on areas of the eastern portion of the state which had little or no previous sampling activities.

In addition to locating the wells in previously non-sampled areas, an effort was made to select wells with ten feet or less of confining layer thickness based on well logs. Although the wells were initially located using this methodology, various reasons including the well being temporarily out of service, dictated that a well in close proximity to the selected well be used as a replacement. Replacement wells, although in close proximity to originally-selected wells, sometimes showed a large local variance of confining layer thickness based on well driller logs.

The sampling sites ultimately are located in three general areas referred to in this report as Area I, Area II and Area III, which generally are associated with portions of Jefferson, Desha and Phillips counties, respectively. Figure 1 depicts the locations of the sampling sites and also the wells from which water samples contained one or more pesticides.

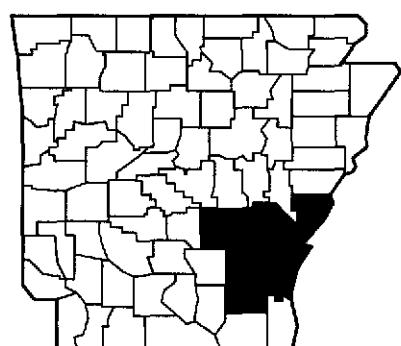
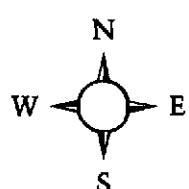
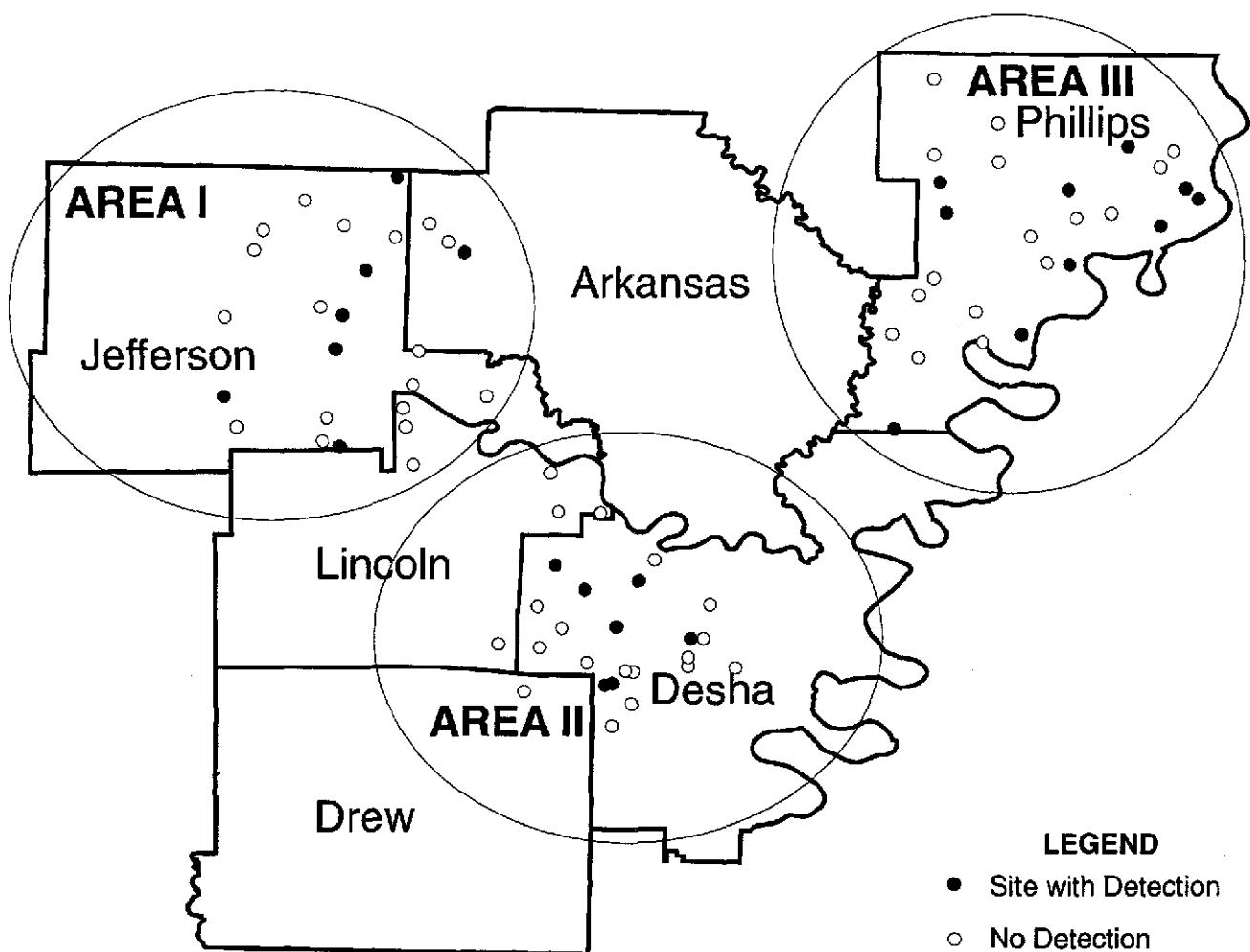


Figure 1. Location of 1996 nonpoint source assessment ground water monitoring sites. Solid circles represent sites with positive pesticide detections.

GEOLOGY

The three study areas are located physiographically in the Gulf Costal Plain region of the state. The deposits underlying this region consist of clay, silt, sand and gravel. The three study areas are underlain entirely by Quaternary-aged alluvial sediments, which were the product of large-scale erosion and deposition during the Pleistocene and Recent Epochs (Boswell et al., 1968). Although the western 1/3 of Jefferson County is underlain by the Jackson Group of Tertiary age, no wells were sampled in this area as yields are typically low (5 gpm) and most irrigation wells are located in the eastern 2/3 of the county within the Quaternary alluvium. The Quaternary alluvium is characterized by a progressive change from fine-grained material at the surface to a coarse sand or gravel at the base.

The Quaternary deposits consist of both Pleistocene terrace deposits and Recent alluvial deposits, which are undifferentiated in this report and referred to as simply Quaternary alluvium. A detailed map differentiating the Quaternary terrace and alluvial deposits within Desha and Lincoln counties is included in Bedinger and Reed (1961).

SUBSURFACE HYDROLOGY

The 77 wells sampled for the present study are completed entirely in the saturated portion of the Quaternary deposits, which together are referred to as the Mississippi River Valley alluvial aquifer. This aquifer provides most all of the water used for irrigation in the study areas. Well yields in the study areas range from 50 to 3000 gallons per minute but generally average about 1400-1600 gallons per minute (Bedinger and Reed, 1961; Klein et al., 1950).

Average water levels in the study areas range from approximately 18 to 24 feet below the surface. Table 1 lists historical and recent water levels for each study area. Average decreases were noted in each area and ranged from approximately 2-10 feet. Because the number of wells varied among different data sets, the possibility that individual years were represented by atypically high or low water tables, and the fact that measurements were taken during different seasons, care should be taken in interpreting the data presented in Table 1. For example, water levels reported for 1989 (Westerfield and Baxter, 1990) were actually lower than those reported for 1992 (Westerfield and Touschner, 1993), which were used for Table 1.

Table 1. Historical versus recent water levels.

	Report	Number of Wells	Water Level (feet)
AREA I	Klein et al., 1950	59	20.8
	Westerfield and Touschner, 1993	11	22.6
AREA II	Bedinger and Reed, 1961	237	14.6
	Westerfield and Touschner, 1993	15	24.4
AREA III	Westerfield, 1977	31	11.3
	Westerfield and Touschner, 1993	16	17.8

METHODOLOGY

Sampling sites were chosen according to a random grid to provide even coverage across each study area. The sampling plan initially included wells which had ten feet or less of confining layer thickness. Attempts to locate and sample the selected wells were made difficult by the fact that many wells were inoperative for a variety of reasons; most often as a result of the cessation of pumping in anticipation of harvest. Although replacement wells were in close proximity to the originally-selected sites, logs of these wells often showed a wide variation in the confining layer thickness, and some replacement wells had thicknesses > 10 feet. Additionally, comparison of many well logs to USGS maps (Gonthier and Mahon, 1994) of confining layer thickness often demonstrated numerous discrepancies. Without additional information, it is assumed that either the well logs are lacking in accuracy in many cases or that the confining layer is of variable thickness across short distances.

The wells were sampled between July 29, 1996, and September 23, 1996. The extension of the sampling period into and past the harvest season (late August and September, 1996), was one of the reasons that operative wells were scarce. The extended sampling period was the result of the increased time requirements associated with the analyses of bentazon, which resulted in a limit of generally ten samples per week to the laboratory. All wells were sampled as near to the wellhead as possible through available faucets or other ports. Where these devices were not available, samples were taken from openings in the irrigation equipment, which included both disposable (plastic) and metal irrigation lines. In the event that a well was started by the farmer for sampling purposes, the well was allowed to run for a minimum of ten minutes prior to sampling.

All samples were collected in approved containers for the selected parameters. Samples were filtered through 0.45 μm pore-sized membranes in the field for metal analysis and, as such, all metals are reported as dissolved metals. All other parameters including pesticides were unfiltered samples. All samples were stored on ice and delivered to the Department laboratory under chain-of-custody requirements by the sampling team.

All analyses are represented by an numbering system which denotes the project type, water type, county and well number. For example, NSGWAR01 represents a nonpoint source (NS) water sample derived from a ground water (GW) source in Arkansas County (AR) from well # 1. Because all samples have the same first four-letter designation, the samples in many instances will be represented by county and number only.

WATER QUALITY

Interpretation of water quality analyses was performed by evaluating general water quality, geochemistry and the detection of pesticides. Individual parameters were compared to federal drinking water standards in order to evaluate the general water quality for use as a drinking water source. Detections of pesticides were compared to health advisory limits and/or federal drinking water standards. Because nitrate as nitrogen concentrations in all cases were below 0.5 ppm and in most cases below 0.1 ppm, impacts from fertilizer use were not considered problematic and are not addressed in the report.

GENERAL WATER QUALITY

In general, the water quality from all three study areas is very good. The water in all three areas is a calcium-bicarbonate type, although sodium, chloride and sulfate were significant contributors to the total dissolved solids (TDS) in several wells, primarily in Jefferson and Desha counties. Calcium comprised over 50% of the total cation concentration in all but 3 cases and averaged 60% of the total cations. Bicarbonate comprised over 50% of the total anion concentration in all cases and averaged 79% of the total anions.

Tables 2, 3 and 4 list the minimum, maximum and mean concentrations for the dominant cations and anions for each of the study areas. A cursory review of the data reveals differences between the data sets for Area I and II wells versus Area III wells. Although sodium and chloride are considerably elevated in Areas I and II with respect to Area III, the mean TDS values are somewhat similar between all three areas, and the mean TDS concentration for the Area III wells is actually higher than that of the mean TDS for the Area I wells. A detailed inspection reveals that the lower concentrations for sodium and chloride are offset by the higher bicarbonate concentrations for the Area III wells. Also, average magnesium concentrations in Area III wells were twice as high as either Area I or II wells. The higher bicarbonate concentrations are related to the increase in magnesium and together with the lower sodium and chloride concentrations clearly demonstrate a ground-water chemistry for Area III which is considerably different from that of the Area I and II well-water samples.

Figure 2 depicts a set of Piper diagrams for all data. Area I and II wells generally plot in an overlapping area of the diagram, indicating the similarity of the chemistry between the two data sets as noted in Tables 2-4. However, Area III wells plot as a separate and distinct pattern as a result of the higher magnesium and bicarbonate concentrations and lower sodium and chloride concentrations. This distinct pattern for the Area III wells is better defined on the tri-linear diagrams for the cations and anions than on the center diagram which plots both cations and anions.

Mean TDS concentrations from the data for the present study were below 500 mg/L, although individual well samples exceeded this number in 17 wells. Chloride was elevated in several samples, although none exceeded the secondary MCL of 250 mg/L. Iron was consistently high (0.75 to 70.5 mg/L) in all samples; however, high iron concentrations are ubiquitous throughout the extent of the alluvial aquifer and often limit its use for municipal, industrial and domestic supply without treatment (Broom and Lyford, 1981). The MCLs for TDS (500 mg/L), iron (0.3 mg/L) and chloride (250 mg/L) are all secondary drinking water limits, which are unenforceable federal guidelines regarding taste, odor, color and other non-aesthetic effects of drinking water (EPA, 1996). As such, there are no human health concerns associated with these secondary MCLs. Individual analyses for each well are in Appendix 1.

Tables 2-4 also compare the data from each of the areas to that of older data from three USGS reports, respectively: Klein et al. (1950), Bedinger and Reed (1961) and Westerfield (1977). The mean values for each of the data sets are remarkably similar including the data from the 1950 report by Klein et al., which reported on data from samples collected dominantly throughout 1949. Most of the samples collected for the 1961 report by Bedinger and Reed were collected during 1952.

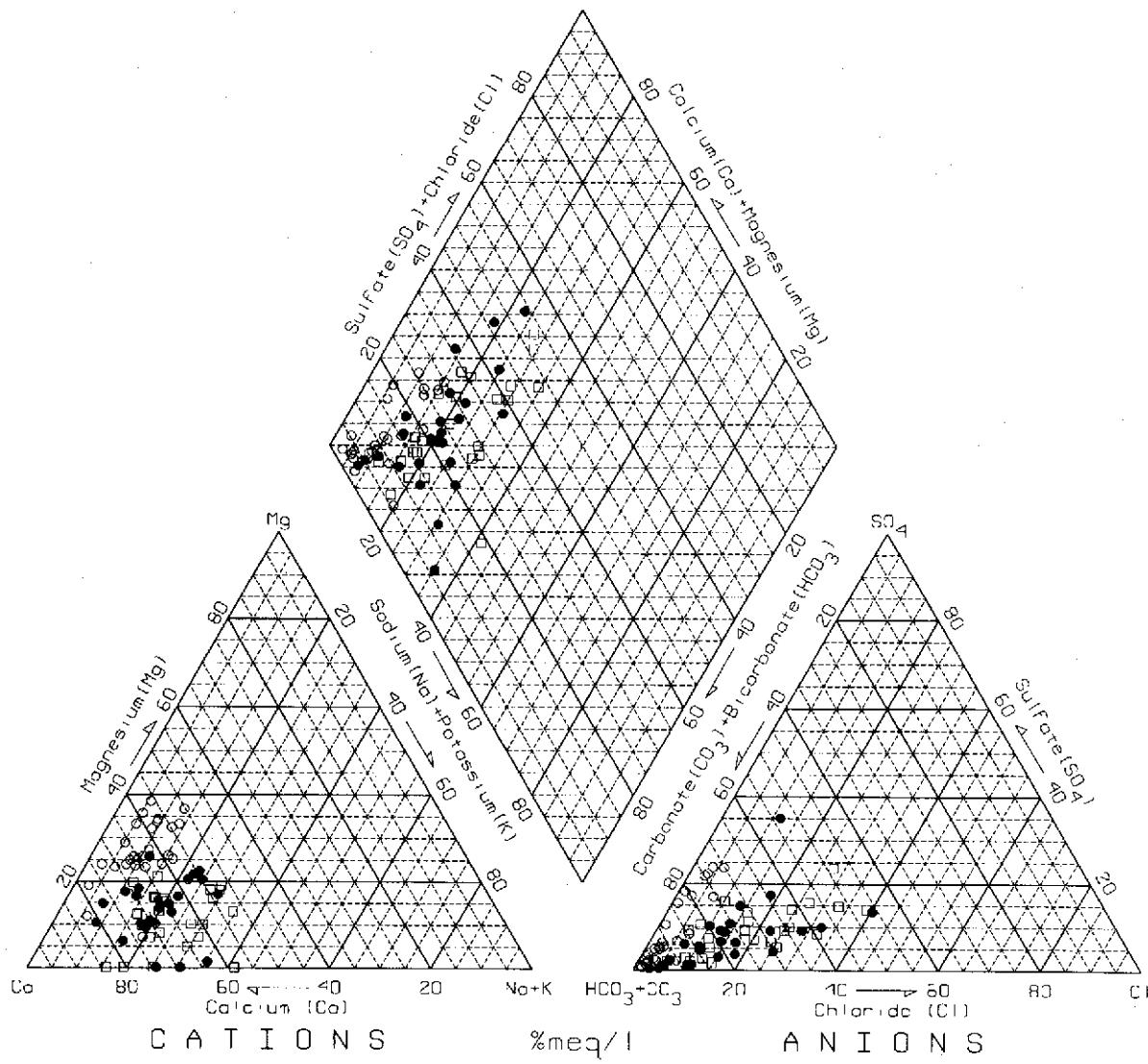


Figure 2 - Piper diagram showing dominant cation/anion relationships. Area I wells are represented by solid circles; Area II wells are represented by squares; and Area III wells are represented by open circles.

Table 2. Statistics for selected parameters for Area I wells.

	Minimum		Maximum		Mean	
	ADPC&E ¹	USGS ²	ADPC&E	USGS	ADPC&E	USGS
Fe ($\mu\text{g/L}$)	4690	30	70500	28000	14426	6800
Ca (mg/L)	30	8	132	162	85	79
Mg (mg/L)	0.03	2.3	33	39	11.8	20.8
K (mg/L)	0.4	---	2.2	---	1.0	---
Na (mg/L)	11.9	8.3	71	206	28.5	36.0
SiO_3 (mg/L)	18.7	11	17.4	53	27.9	25.6
Cl (mg/L)	8	3.8	184	100	35	39
SO_4 (mg/L)	0.5	1.0	174	294	28	29
HCO_3 (mg/L)	165	4	541	602	323	355
TDS (mg/L)	195	195	697	851	395	417

¹ Data from 1996 nonpoint sources assessment wells (77 wells).² Data from Klein et al., 1950 (32 wells).**Table 3.** Statistics for selected parameters for Area II wells.

	Minimum		Maximum		Mean	
	ADPC&E ¹	USGS ²	ADPC&E	USGS	ADPC&E	USGS
Fe ($\mu\text{g/L}$)	1500	20	26700	62000	12118	13060
Ca (mg/L)	39	9.1	127	137	82	49.5
Mg (mg/L)	0.03	5.4	25.6	39	10.7	19.6
K (mg/L)	0.8	---	2.0	---	1.2	---
Na (mg/L)	7.9	12	78.4	54	35	32
SiO_3 (mg/L)	26.0	---	41.1	---	30.4	---
Cl (mg/L)	11	3.2	181	188	56	62
SO_4 (mg/L)	2.1	1	145	172	37	38
HCO_3 (mg/L)	151	53	533	561	299	331
TDS (mg/L)	184	123	707	682	433	464

¹ Data from 1996 nonpoint sources assessment wells.² Data from Bedinger and Reed, 1961 (58 wells for Fe, HCO_3 , SO_4 , and Cl; 13 wells for Ca and Mg; 10 wells for TDS; 7 wells for Na).

Table 4. Statistics for selected parameters for Area III wells.

	Minimum		Maximum		Mean	
	ADPC&E ¹	USGS ²	ADPC&E	USGS	ADPC&E	USGS
Fe ($\mu\text{g/L}$)	753	2400	22200	6100	6964	4457
Ca (mg/L)	57	58	145	120	93	94
Mg (mg/L)	4.5	21	50.5	47	24.4	31
K (mg/L)	0.5	1.2	3.4	3.3	1.5	2.2
Na (mg/L)	3.6	12	37.9	47	15.1	24
SiO_2 (mg/L)	32.0	31	42.6	37	35.5	34
Cl (mg/L)	3	4.9	25	29	8	13
SO_4 (mg/L)	2.1	3.5	135	110	35.7	37
HCO_3 (mg/L)	236	310	749	595	418	440
TDS (mg/L)	227	299	669	625	421	436

¹ Data from 1996 nonpoint sources assessment wells (77 wells).

² Data from Westerfield, 1977 (7 wells).

Another point of interest is that average chloride concentrations virtually have not changed since 1952 in the high chloride areas of Desha county. Elevated chlorides were documented in several places in Desha county and a chloride isoconcentration map was developed from the 1952 data set and is presented in Figure 3. Figure 4 depicts an isoconcentration map developed from data collected for the present study. The areas of high chloride concentrations are similar between both maps and suggest that individual well-water chloride concentrations, similar to that of the average chloride concentrations, have changed little since 1952.

RESULTS OF PESTICIDE ANALYSES

Monitoring for pesticides in ground water is both a challenging and expensive exercise. There are numerous pesticides and pesticide combinations used for control of weeds, fungi and insects. However, insecticide and fungicide use pale in comparison to herbicide use in Arkansas. Herbicides are used previous to and throughout the growing season as preemergents, weed control during the plant cycle, and defoliants prior to harvest. Because of the numerous pesticides used in the state and the exorbitant cost of pesticide analyses, the primary focus of most monitoring plans is on high-use pesticides and those pesticides with a medium to high potential for being transported to the ground water table. One of the objectives of the investigation was to compare the analyses to both pesticide use figures and chemical characteristics in order to evaluate current monitoring practices and modify future practices if required.

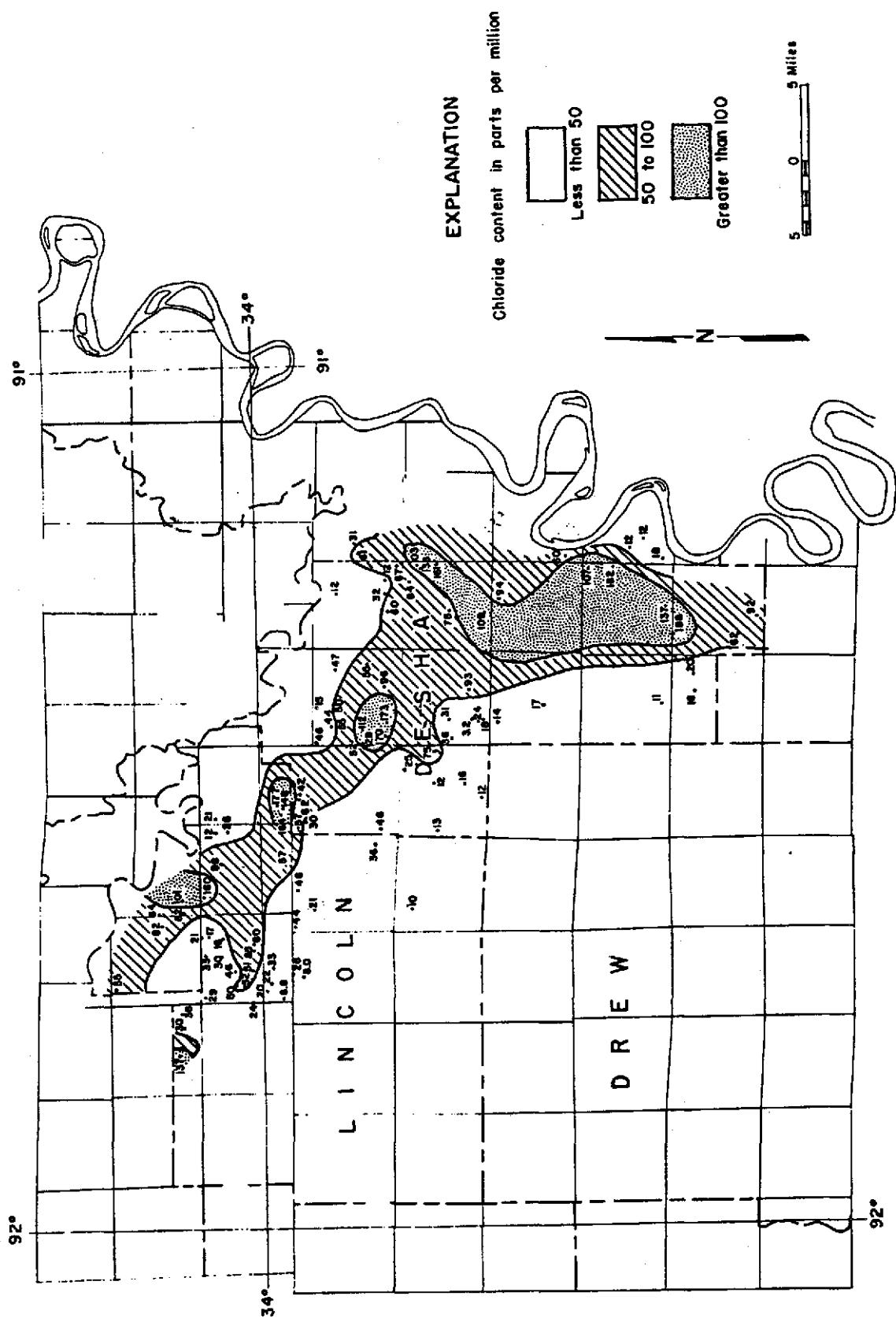
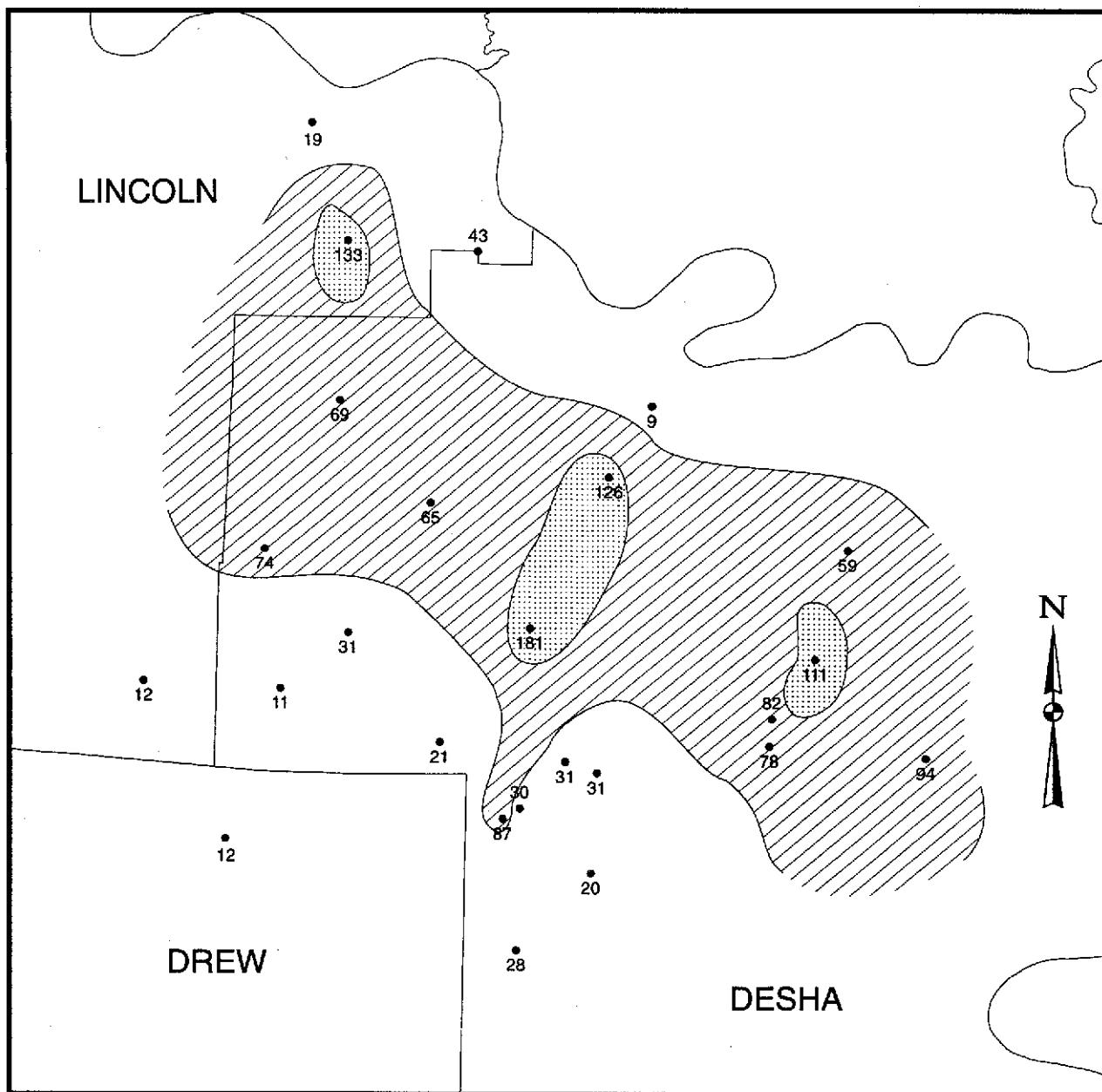


Figure 3. Chloride isoconcentration map of water from the Quaternary alluvium (taken from Bedinger and Reed, 1961).



LEGEND

Chloride content in parts per million

[White box] Less than 50

[Diagonal hatching] 50 to 100

[Dotted box] Greater than 100

5 0 5 miles

Figure 4. Chloride isoconcentration map for Area II wells (1996 data).

Table 5 lists some of the major pesticides used in the state according to the number of acres receiving pesticide application. Because figures for actual use are not kept and/or reported by the farmer, the figures listed in Table 5 are interpolated from three separate studies by the Arkansas Cooperative Extension Service (Spradley, 1991; Spradley, 1992; Johnson and Meyers, 1997). These studies represent a compilation of questionnaire results from participating farmers across the state. As such, the figures should not be interpreted as representing the actual use of any one pesticide, although they provide a good estimate of the more frequently-used pesticides.

Table 6 provides an index of chemical properties and an associated range of values that are used to assess the leaching potential of pesticides. The table was developed by the EPA and represents the range of values for the frequently detected pesticides in the EPA data base up to 1985 (EPA, 1986). Although there are several chemical characteristics that promote or retard the leaching potential of a pesticide including solubility, adsorption, mineralization and microbial degradation, the most important characteristic is probably the solubility of a pesticide. Low-solubility pesticides are not commonly detected in ground water except under special conditions such as an unusually shallow water table, preferred pathways including clay fractures and root zones, and through facilitated transport mechanisms. Also, in most cases low solubility is associated with high adsorption values, which further limits the migration potential of a pesticide.

Table 7 lists some of the more important chemical characteristics affecting leaching potential for the major-use pesticides in Arkansas. Adsorption is represented by the organic carbon partition coefficient (K_{oc}), which is a measure of the tendency of the pesticide to adsorb onto soil organic carbon. The table also cites a leaching potential for each of the pesticides based on three characteristics: solubility, K_{oc} and field half-life ($t_{1/2}$). The leaching potential is provided as a means of estimating the effects of the chemical characteristics on the migration potential of the pesticide and should not be interpreted as a quantitative or exact value. It is the author's opinion that too much weight is given in many cases to the $t_{1/2}$ value in assessing leaching potential. For instance, bentazon is the most frequently detected pesticide in ground water in Arkansas. Bentazon has an effectively infinite solubility in water (2.3 kg/L) and a relatively low organic carbon partition coefficient of 35 mL/g, which would indicate a high leaching potential; however, bentazon is rated as a medium-potential leaching pesticide based on a $t_{1/2}$ of 10 days. Although a low $t_{1/2}$ value is an important characteristic in protecting ground water by destruction of the chemical prior to large-scale migration, field half-lives are based largely on photodegradation and microbial destruction, which are processes that are most effective at the land surface. Once dissolved and leached into the subsurface, field-determined half lives are no longer applicable and the pesticide can remain stable for several months or years, although the half life may be in days. In spite of these considerations, however, the qualitative predictions are very useful to the pesticide investigator. Pesticides termed as low leaching-potential pesticides are rarely present in ground water; however, the medium to high potential leachers are commonly present depending on use and site characteristics.

As stated above, there is an inverse correlation between solubility and K_{oc} ; as solubility increases, K_{oc} decreases. However, there are exceptions as noted in four of the pesticides listed in Table 7: glyphosate, MSMA, paraquat and ethephon, with reported solubilities of 1,000,000 mg/L and K_{oc} values of 10,000 mL/g. These pesticides, although very soluble in water, are retarded by mechanisms other than adsorption to the organic carbon content of soil. They either interact with

Table 5. Most frequently used pesticides in Arkansas

Pesticide	Rice (acres)	Soybean (acres)	Cotton (acres)	Corn (acres)	Total (acres)
Fluometuron			1930500		1930500
Trifluralin		1241000	666900		1907900
Tribufos			1298700		1298700
Imazquin		1224000			1224000
Propanil	1165800				1165800
Pendimethalin	147400	574600	285900		1037900
MSMA			1006200		1006200
Cyanazine			982800		982800
Carboxin		958800			958800
Acifluorfen	53600	894200			947800
Metolachlor		574600	58500	75000	708100
Chlorimuron		703800			703800
Captan		697000			697000
Bentazon		693600			693600
Ethephon			666900		666900
Metribuzin		612000			612000
DSMA			456300		456300
Prometryn			409500		409500
Clomisone			374400		374400
Molinate	361800				361800
Fomesafen		340000			340000
Glyphosate		336600			336600
Benomyl	335000				335000
2,4,D	294800				294800
Fluazifop		272000			272000
Atrazine				270000	270000
Alachlor		176800		75000	251800
Thiram		224400			224400
Norflurazon			222300		222300
Propiconazole	201000				201000
Thidiazuron			198900		198900
Thiobencarb	187600				187600
Quinclorac	160800				160800
Sethoxydim		159800			159800
Bromoxynil	93800				93800
Paraquat Dichloride			93600		93600
Dimethylipin			93600		93600
Quizalofop		91800			91800
2,4,DB		88400			88400
Diuron			81900		81900
Triclopyr	80400				80400
Iprodione	80400				80400
Acetochlor				75000	75000
Dimethenamid				75000	75000
Methyl Parathion	67000				67000
Lactofen			58500		58500
Malathion	53600				53600
Phenoxyprop	40200				40200
Methazole			35100		35100

Table 6. Chemical characteristics of pesticides indicating leaching potential.

CHEMICAL CHARACTERISTIC OF PESTICIDE	VALUE OR RANGE OF VALUES INDICATING LEACHING POTENTIAL
Solubility	> 30 mg/L
Partition Coefficient (Kd)	<5, usually <1
Organic Carbon Partition Coefficient (Koc)	< 300-500
Speciation	Negatively charged, fully or partially at ambient pH
Hydrolysis Half-Life	> 25 weeks
Photolysis Half-Life	> 1 week
Field Dissipation Half-Life	> 3 weeks
Henry's Law Constant	< 10 ⁻² atm·m ³ mol

Adapted from U.S. Environmental Protection Agency (1986)

iron, which is common to most soils, or react with clay minerals in such a way as to retard leaching and, as such, have a leaching potential listed as "small." Apparently, for sake of consistency, the researchers replaced a bulk K_d value with a high K_{oc} to indicate the high retardation factor associated with the pesticides.

Table 7 also suggests that where the runoff potential is high, the leaching potential is often low. Conversely, where the leaching potential is high, the runoff potential is low. Highly sorptive pesticides are generally immobile because of their strong binding characteristics with solids and generally have low solubility values. As such, these pesticides would not be expected to be transported to either surface or subsurface waters. However, the insoluble pesticides commonly are present in surface water bodies as a result of movement with eroded sediment through a process known as facilitated transport. The pesticides are attached to suspended sediment in the water and unfiltered surface water samples commonly demonstrate positive pesticide detections. For pesticides which are highly soluble and would be expected to occur in all phases of the hydrologic system, an inverse relationship exists; surface waters are generally void of positive detections but ground-water systems commonly exhibit positive detections. Although photodegradation is more active in surface water bodies than on the land surface, another phenomena largely accounts for the discrepancy in positive detections in both water systems. It has been demonstrated with conservative tracers, such as bromide, that directly following a rainfall event, the tracer infiltrates the soil zone in the early phase of the rain event. By the time runoff occurs the tracer has largely been lost to migration into the soil zone with little available for transport by runoff water. Similarly, soluble pesticides often are leached to the subsurface before the occurrence of runoff and are not available for transport to surface water bodies (Waughope, 1996).

Table 7. Chemical characteristics of major use pesticides

Pesticide (common name)	Solubility (mg/L)	Koc (mL/g)	Field t _{1/2} (days)	Runoff Potential	Leaching Potential
Quizalofop	1	100000	140	Large	Small
Trifluralin	1	1400	60	Large	Small
Pendimethalin	1	24300	60	Large	Small
Tribufos	1	5000	10	Large	Small
Benzomyl	2	2100	100	Large	Small
Fluazifop	2	3000	20	Large	Small
Methazole	2	10000	14	Large	Small
Iprodione	13	500	20	Medium	Small
Thidiazuron	20	100	10	Medium	Medium
Triclopyr	23	780	46	Large	Medium
Norflurazon	28	248	45	Medium	Medium
Thiram	30	383	20	Medium	Medium
Atrazine	33	160	60	Medium	Large
Diazinon	40	85	30	Medium	Large
Djuron	42	400	60	Large	Medium
Prometryn	48	614	30	Medium	Small
Bromoxynil	50	1000	14	Medium	Small
Methyl Parathion	60	5100	5	Medium	Small
Fluometuron	90	100	14	Medium	Medium
Propiconazole	110	100	20	Medium	Medium
Malathion	145	1780	1	Small	Small
Carboxin	170	264	20	Medium	Medium
Cyanazine	171	168	20	Medium	Medium
Ametryn	185	388	30	Medium	Medium
Alachlor	240	190	14	Medium	Medium
Propanil	500	188	1	Small	Small
Chlorimuron	500	20	50	Small	Large
Metolachlor	530	200	20	Medium	Medium
Propachlor	580	420	7	Medium	Small
Prometon	750	300	120	Large	Large
Molinate	880	110	21	Medium	Medium
Sethoxydim	1000	50	5	Small	Small
Metribuzin	1220	41	30	Medium	Large
Dimethapin	3000	10	10	Small	Large
Endothall	100000	20	2	Small	Small
Trichlorfon	154000	2	27	Small	Large
Imazaquin	160000	20	60	Small	Large
2-4-DB	200000	20	10	Small	Medium
MCPA	270000	20	14	Small	Large
2-4-D	300000	109	10	Medium	Medium
Fomesafen	600000	50	180	Medium	Large
Acifluorfen	900000	139	30	Medium	Medium
Glyphosate	1000000	10000	30	Large	Small
MSMA	1000000	10000	100	Large	Small
Paraquat	1000000	10000	3600	Large	Small
Ethephon	1000000	10000	5	Medium	Small
Bentazon	2300000	35	10	Small	Medium

*adapted from Waughope, 1988

Table 8 lists samples with positive pesticide detections for the present study. A comparison of the available chemical characteristics for the pesticides in Table 8 to the ranges denoting a potential leacher (Table 6) reveals that all of the solubility concentrations except for trifluralin exceed the 30 mg/L concentration denoting a potential leacher; the range of Koc values fall below 300 mL/g except for prometryn and trifluralin, which have Koc values of 614 and 1400 mL/g, respectively; and the half-life values range from 10-60 days with three of the pesticides below the ">3-weeks" value cited as contributing to leaching potential. Because the best correlation of detected pesticides is with solubility, solubility is considered the most important factor in the potential for ground-water contamination as a result of pesticide leaching. All of the detected pesticides are listed in Table 7 as having a medium to large potential for leaching, except for prometryn and trifluralin, which have a small leaching potential. The low leaching potential for prometryn probably is based on the high Koc value, although this propensity for binding to soils is greater in soils with high clay and organic matter content (EPA, 1983), and available data indicates that prometryn is mobile in sandy soils and moderately mobile in sandy loams (EPA, 1987).

The most frequently detected pesticide for the present study was bentazon (14 detections) followed by molinate (7) and metolachlor (3). Figure 5 lists the pesticide detections by the number of detections and the percentage of the total detections for each of the pesticides. Bentazon far exceeded any other pesticide detection and accounted for approximately 37% of the total detections. This information correlates closely with the data gathered by the AWRC (Nichols et al., 1996), which listed bentazon as the most frequently detected pesticide; accounting for 10 of the 23 (43%) detections through 1996. Table 9 compares pesticide data from three sources: the Department, the AWRC and the USGS. The USGS data were generated from a project conducted from 1991-1993 which included sampling and re-sampling of 27 wells in various counties in the Coastal Plain area of the state; the AWRC data are from of an ongoing pesticide monitoring program begun in 1992 and includes 231 wells through the 1996 sampling period; and the Department data are from the present study results. The data vary widely in the type of pesticides monitored and detected by each organization and in the total percentage of detections. Individual detections and the resulting percentage of total detections are dependent on the methodology, instrumentation, and reporting criteria of each agency and strict interpretation of the differences should be avoided. Consideration of these differences have been incorporated into recent agendas on the part of participating agencies and future objectives include evaluating and standardizing certain elements of the analytical and reporting mechanisms.

Figures 6 and 7 depict graphs of solubility versus sorption for some of the major-use pesticides in Arkansas. Figure 6 denotes which pesticides were detected in ground water, whereas Figure 7 depicts the pesticides detected in surface water. The surface water data were generated from samples collected by Department personnel from 1993-1996, and the ground water data were extracted from the totals provided in Table 9. Using the qualitative models proposed in Table 6 and Table 7, the pesticides in the center of the graph dominantly are medium-potential leachers; those in the far left corner below the 10 mg/L solubility line and above the 1000 mL/g line are dominantly small-potential leachers; and those to the right of the 1000 mg/L line and below the 300 mL/g line are dominantly high-potential leachers.

Table 8. List of Sample Locations with Positive Pesticide Detections

Sample Location	Molinate (ug/L)	Ortho-Ethyl Para-Nitrophenol (ug/L)	Atraton (ug/L)	Atrazine (ug/L)	Mefenacet (ug/L)	Ametryn (ug/L)	Prometryn (ug/L)	Metalachlor (ug/L)	Cyanazine (ug/L)	Methoxychlor (ug/L)	Silvex (ug/L)	Bromazone (ug/L)	Achluafen (ug/L)
NSGWAR02	0.03484												
NSGWJF02	0.24612					0.01141							
NSGWJF05	0.08647					0.00769		0.00407					
NSGWJF08		0.01137	0.0164					0.01055					
NSGWJF11				0.00726									
NSGWDE01	0.09336												
NSGWDE06	0.17074												
NSGWDE07	0.01605												
NSGWMJF19	0.09861												
NSGWPHT6				0.02107									
NSGWPHT6					0.00536								
NSGWDE21							0.01808						
NSGWPHT7								0.00536					
NSGWPHT10									0.0578				
NSGWDE10											0.08748		
NSGWDE11													
NSGWPHT4													
NSGWDE02													
NSGWPHT04													
NSGWPHT08													
NSGWPHT09													
NSGWPHT17													
NSGWPHT22													
NSGWPHT23													

Percentage of Detections per Pesticide

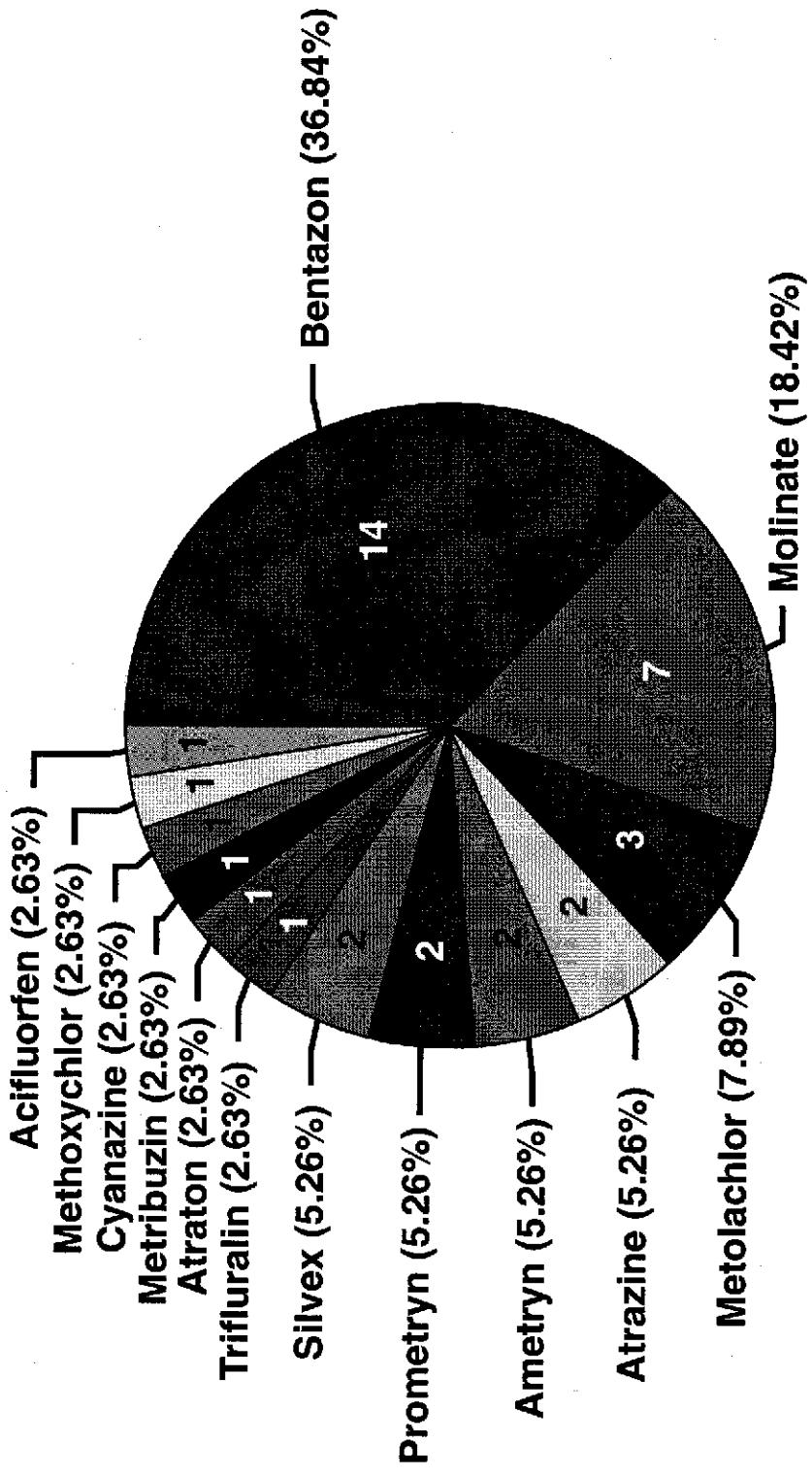


Figure 5. Number and percentage of individual pesticide detections.

PESTICIDE SOLUBILITY VERSUS SORPTION

Pesticide Detections in Ground Water

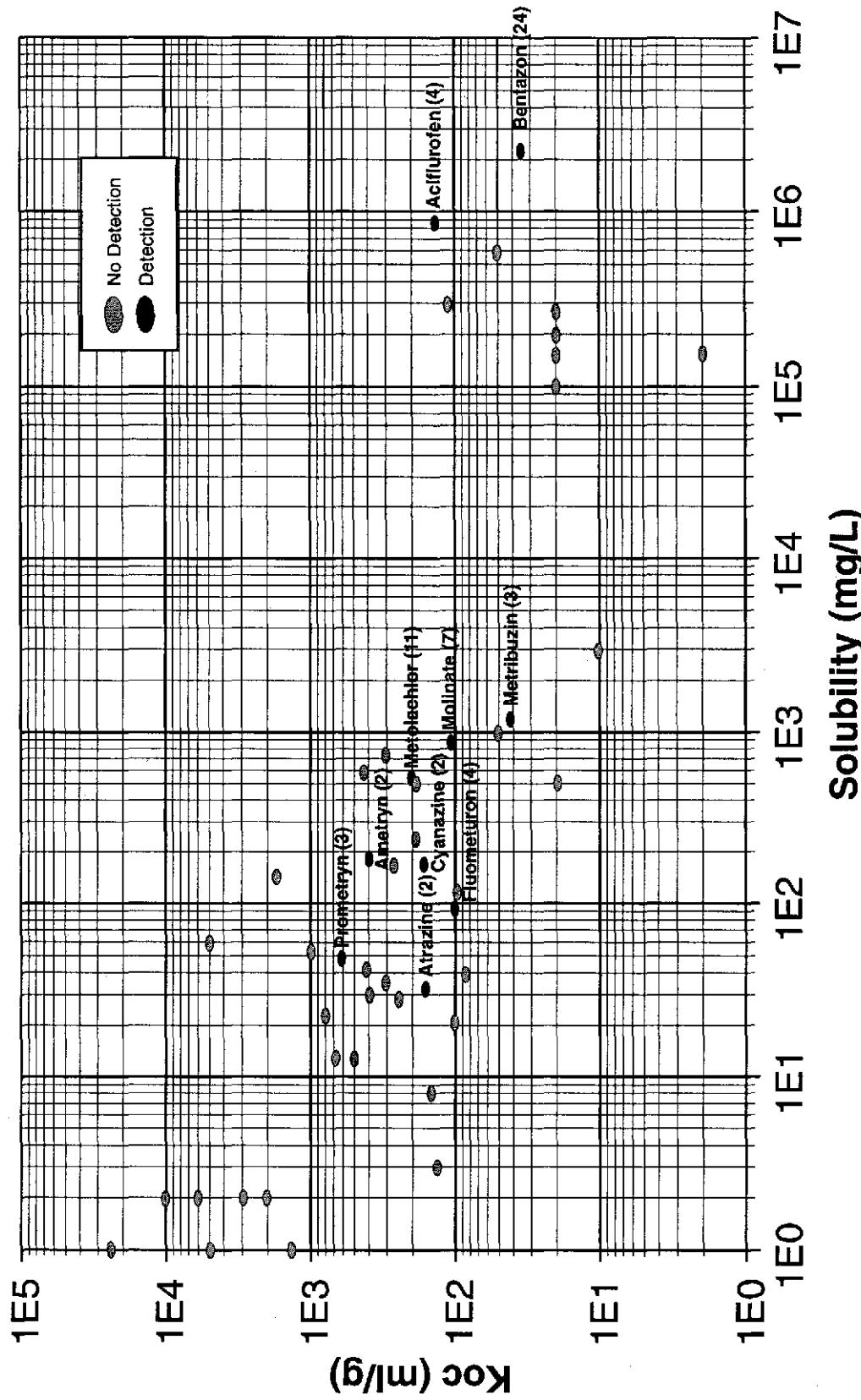


Figure 6. Pesticide detections in ground water as related to solubility and sorption (represented by K_{oc}).

PESTICIDE SOLUBILITY VERSUS SORPTION

Pesticide Detections in Surface Water

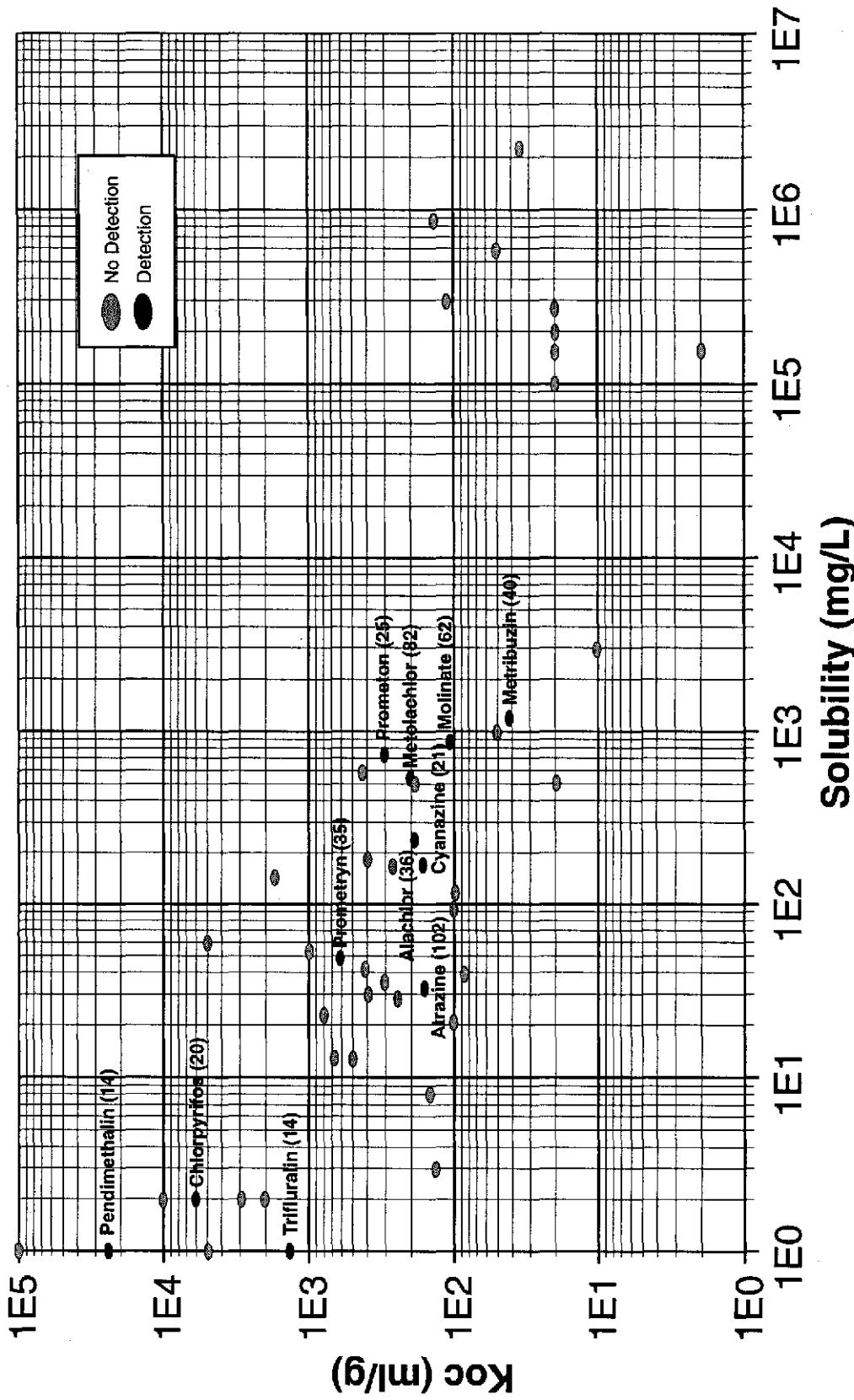


Figure 7. Pesticide detections in surface water as related to solubility and sorption (represented by K_{oc}).

Table 9. Pesticide detections in ground water from major sources*

	AWRC	ADPC&E	USGS	Total
Bentazon	10	14	NA	24
Metolachlor	2	3	6	11
Molinate	0	7	NA	7
Acifluorfen	3	1	NA	4
Fluometuron	4	0	NA	4
Metribuzin	2	1	0	3
Prometryn	NA	2	1	3
Cyanazine	0	1	1	2
Atrazine	0	2	0	2
Ametryn	NA	2	0	2
Silvex	NA	2	0	2
# Wells Sampled	231	77	27	335
# Wells w/ Detection	14	24	9	47
% Detected	6	31	33	14

NA = Not Analyzed

* Detections include multiple hits at some wells; as such, detections outnumber # of wells w/ detections. Table includes only pesticides with two or more total detections.

It is interesting to note that the pesticides in the center of the Figures 6 and 7 were detected in both surface and subsurface waters; those in the upper left-hand corner (pendimethalin, chlorpyrifos, and trifluralin) were detected only in surface waters; and the pesticides in the far right-hand corner (bentazon and acifluorfen) were detected only in ground water. This observation supports the preceding discussion concerning the detection of the highly-sorptive, low-solubility pesticides in surface water only; whereas, the high-solubility pesticides are found in ground water, but generally not detected in surface water. The findings also suggest that further investigations may be needed for determining the flux of pesticides into the water and sediment phases of surface water bodies in Arkansas.

The most frequently detected pesticides for the present study, bentazon (14), molinate (7) and metolachlor (3), all have solubilities greater than 500 mg/L and Koc values less than 200 mL/g. Also, all pesticide detections for the present study and pesticide detections for the combined studies (Table 9) were pesticides in the top 20 of the most frequently used pesticides (Table 5). Four of the seven most frequently-used pesticides from Table 5, including tribufos (3rd), propanil (5th), pendimethalin (6th) and MSMA (7th), were not detected in any of the studies and are all "low-potential" leachers according to Table 7. This finding suggests that the chemical and physical properties of pesticides far exceeds pesticide use as the overall controlling factor for the migration

of pesticides to ground water. However, pesticide use and patterns of use seem to play an important role in the occurrence of pesticides in ground water. Figures 8 and 9 denote the locations of sampling sites with associated bentazon detections and molinate detections, respectively, and also the 1996 number of acres of harvested soybeans and rice, respectively. Because molinate is used on rice, whereas bentazon is used dominantly on soybeans, a clear pattern of occurrence of detections versus use is evident from the figures. For example, Phillips County produced almost the combined total of soybeans harvested in Jefferson and Desha counties and also had seven detections of bentazon; the combined total detections for Jefferson and Desha counties. However, both Jefferson and Desha counties produced twice the rice harvested in Phillips county and had a combined total of seven detections for molinate; whereas, no molinate was detected in Phillips County. This relationship indicates the need for gathering information concerning dominant crops near pesticide sampling locations and also provides an additional cause for the diversity of results from various agencies by introducing another important variable in addition to analytical and reporting criteria.

In summary, pesticides were detected in ground water at percentages ranging from 6 to 33% of the sites sampled by the various investigators. Pesticides were detected at 24 of the 77 (31%) sites sampled for the present study. The most pesticides detected at any one well was 4 pesticides at NSGWJF08. The chemical and physical properties of pesticides strongly control the migration potential and transport to the ground water table. Highly soluble pesticides were found by all investigators in ground water but not in the surface water bodies; whereas, the low-soluble, highly-sorptive pesticides were not detected in ground water, but frequently detected in surface water. Moderately soluble pesticides are found in both water sources. A relationship was established between the harvested acres of rice and soybeans and the total detections for molinate and bentazon, respectively.

GEOCHEMISTRY OF ALLUVIAL AQUIFER IN STUDY AREA

As stated above, the water in all study areas is a calcium-bicarbonate type with varying concentrations of sodium, chloride, sulfate and magnesium as other major contributors to the TDS. Least-squares linear regression analyses was applied to the data using QuattroPro in order to compare the relationships between various chemical parameters. This analysis method tests the variance between a set of independent and dependent variables. The r-squared value explains the variation within the linear model, and, as such, represents the reliability of the regression with a value between zero and unity. The linear relationship is more reliable as the r-squared value approaches unity.

The charge balance of the ground water parameters is evident from Figure 10, which displays the close correlation between the cations and anions. In view of this relationship, a close correlation should exist between the individual ion-pairs, where there is a singular source rock for the dissolved ions and a lack of reactions between the ions in solution and the material through which the water flows. Calcium and bicarbonate ions are commonly derived from the dissolution of calcium carbonate rocks and the sodium and chloride is in most cases from the dissolution of halite. However, inspection of the graphical relationship between both ion pairs (Figures 11 and 12), although depicting a strongly positive relationship, have relatively low r^2 values.

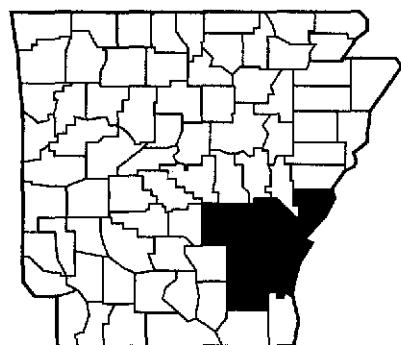
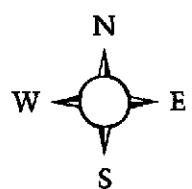
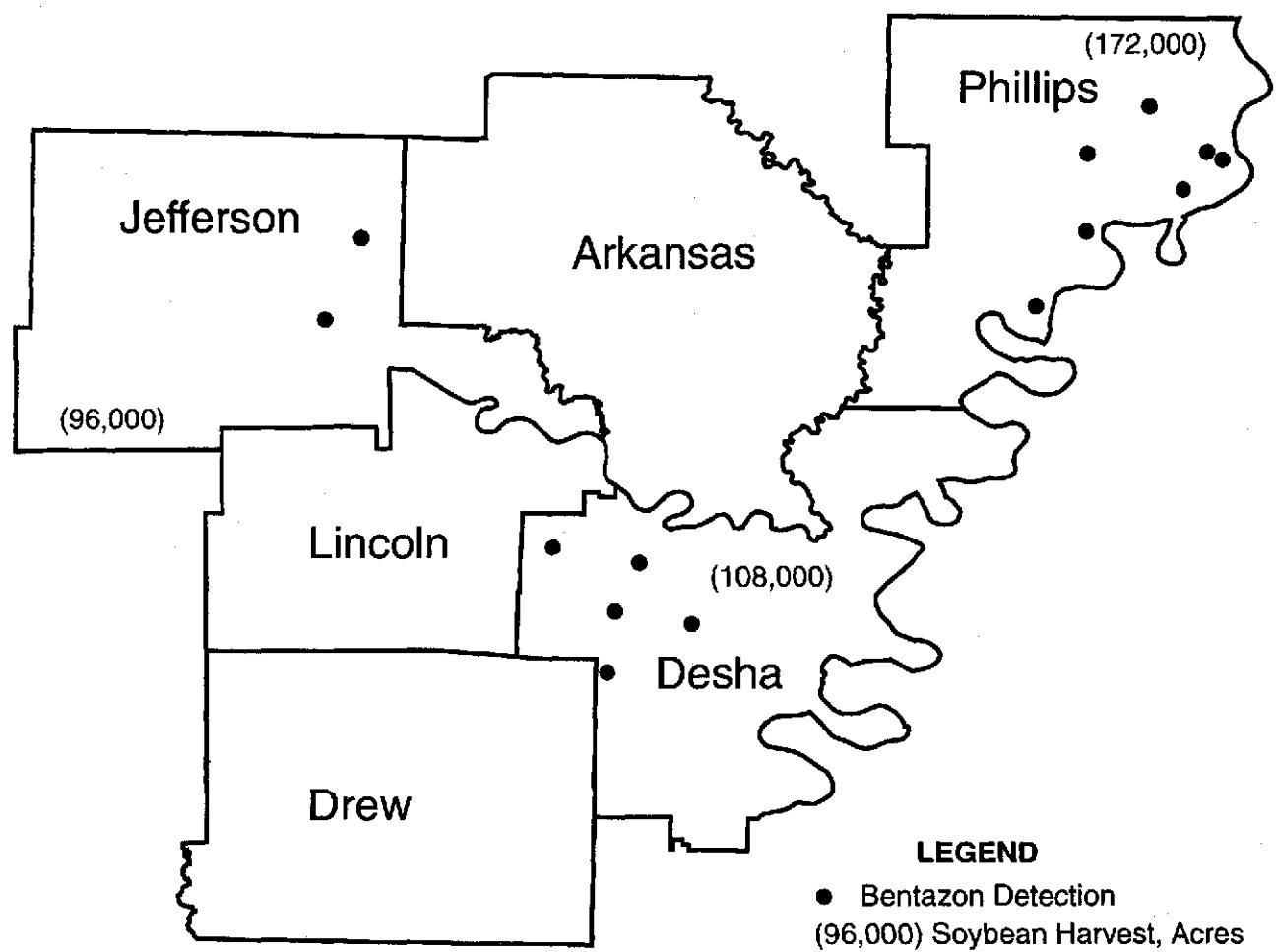


Figure 8. Detections of bentazon as related to soybean production.

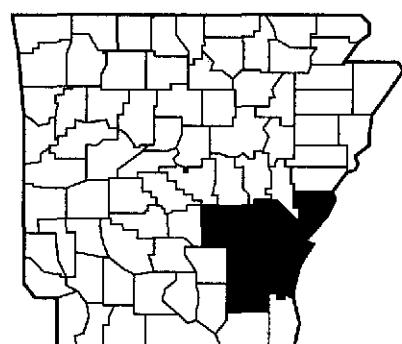
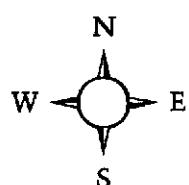
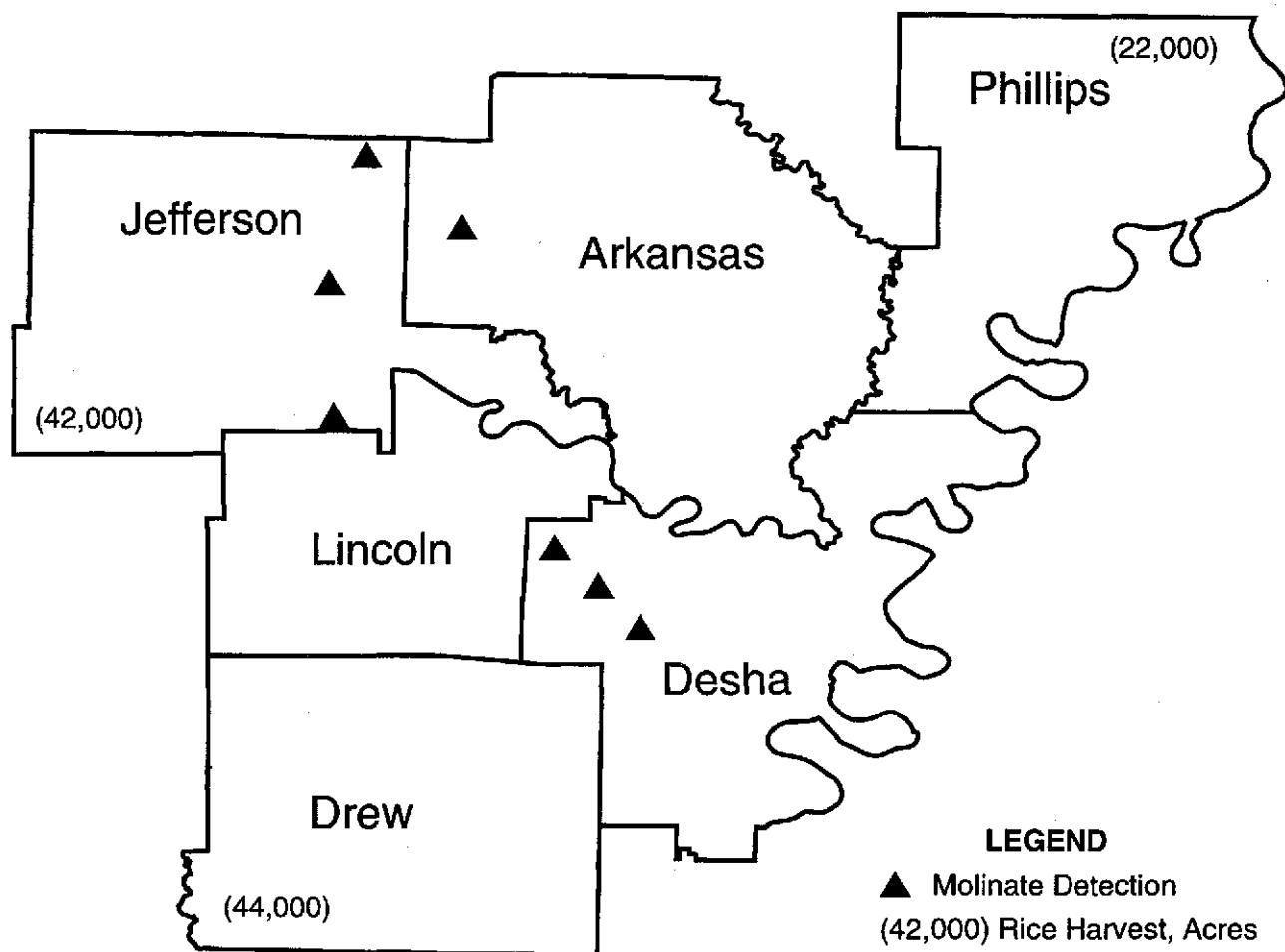


Figure 9. Detections of molinate as related to rice production.

Cations vs. Anions

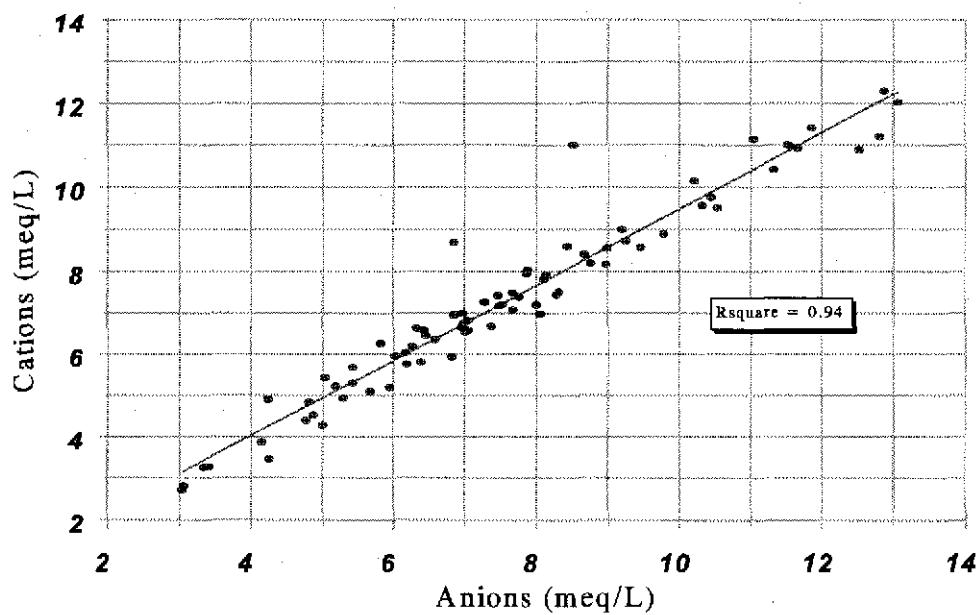


Figure 10. Relationship of Total Cations (meq/L) to Total Anions (meq/L). Goodness of Fit is Represented by Rsquared Value from Regression Analyses.

Calcium + Magnesium versus Bicarbonate

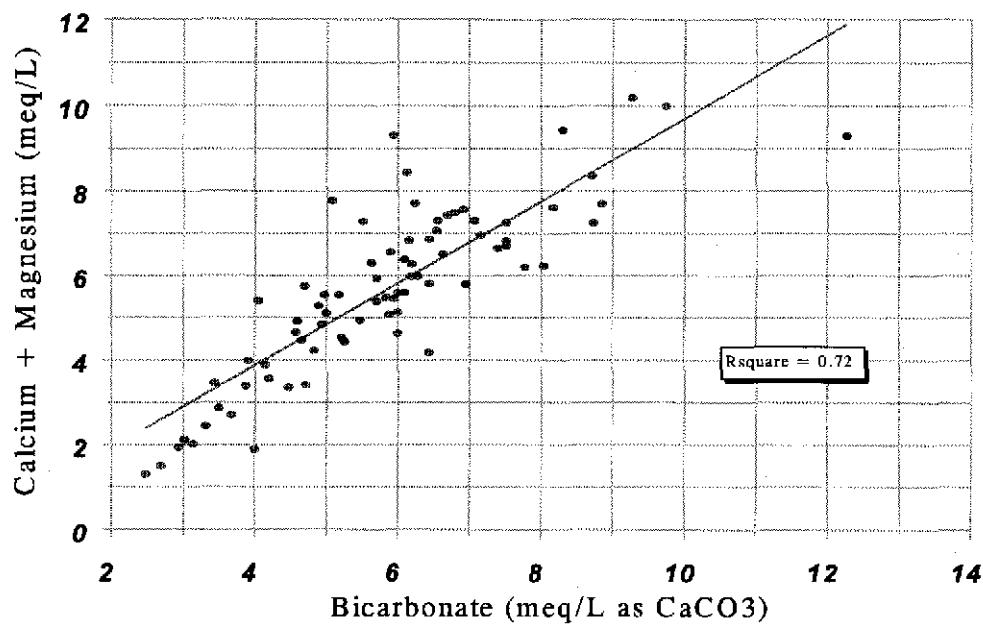


Figure 11. Relationship Between Calcium and Magnesium (meq/L) to Bicarbonate (meq/L). Goodness of Fit is Represented by the Rsquared Value from Regression Analyses.

Sodium versus Chloride

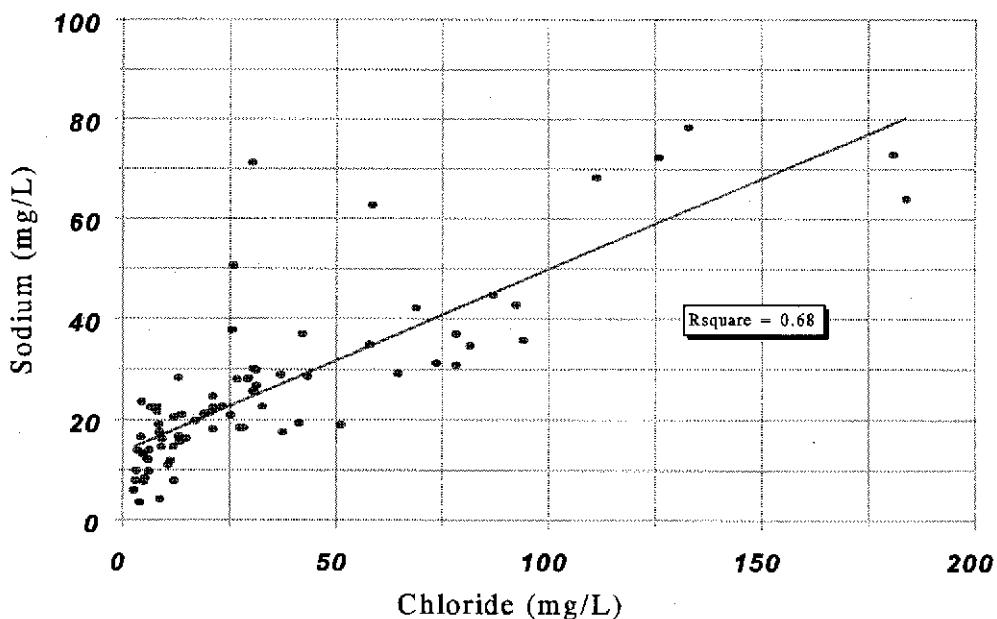


Figure 12. Linear relationship between sodium (mg/L) and chloride (mg/L). Goodness of fit is represented by r^2 , which was calculated by linear regression analysis.

The fact that individual relationships between the major ion pairs did not exhibit a high level of correlation suggests that there are additional mineral sources and/or chemical reactions which result in a change to the overall chemistry of the study area ground water. Ion exchange processes can account for an increase of some ions at the expense of other ions. The dominant cation exchange process within most soils and unconsolidated sediments is the exchange of calcium for sodium, especially within clay minerals. In addition, sulfate concentrations ranged upwards to 174 mg/L, which suggests that some of the calcium may be derived from the dissolution of gypsum. In order to test the influence of sources of calcium other than calcium carbonate and the influence of cation exchange, several relationships were investigated.

Figure 13 depicts the relationship of calcium and magnesium divided by bicarbonate versus sulfate concentrations. The dissolution of calcium carbonate should yield calcium plus magnesium concentrations equal to that of bicarbonate concentrations in meq/L; i.e., a ratio of one. Two phenomena are noted from inspection of Figure 13. First, ratios greater than one, which indicate an enrichment of calcium and/or magnesium, correlate to an increase in sulfate concentrations. The increase in sulfate with corresponding increases in calcium suggests that gypsum is most likely contributing substantial amounts of calcium to the ground water. This theory was further investigated by plotting calcium plus magnesium versus bicarbonate plus sulfate (Figure 14). The r -squared value improved from 0.72 (Figure 11) to 0.86, which supports the theory that gypsum is contributing substantially to the enrichment of calcium. Secondly, ratios less than one extend to the left past the $x=0.5$ line on Figure 13, which denotes a concentration of calcium and magnesium at 50% of the bicarbonate concentration, and suggests that cation exchange processes are contributing to the depletion of calcium.

Ca/HCO₃ versus Sulfate

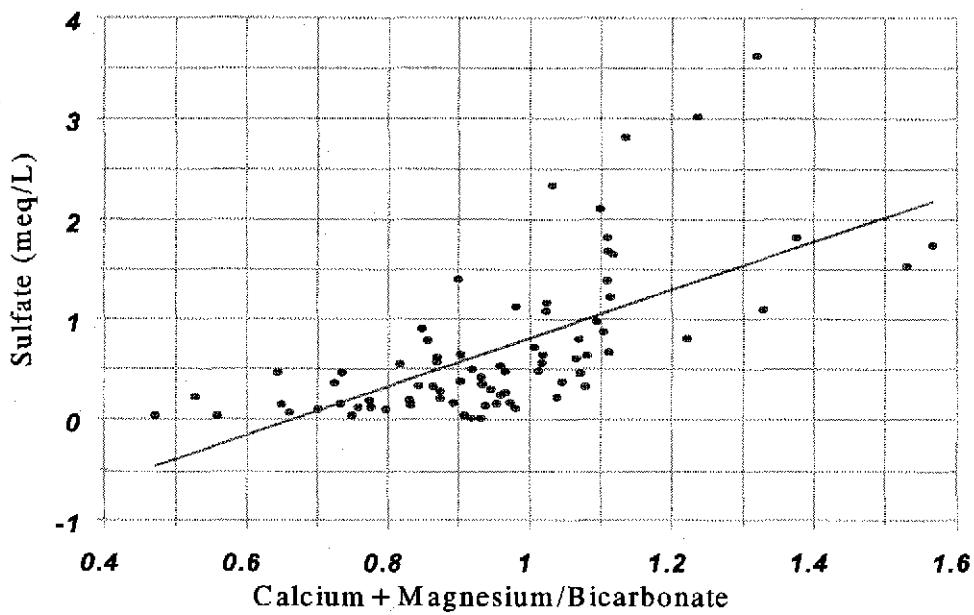


Figure 13. Linear relationship of calcium and magnesium (meq/L) divided by bicarbonate (meq/L) to sulfate (meq/L). Increase of sulfate to the right of x=1 indicates potential contribution of calcium by dissolution of gypsum.

Ca + Mg versus HCO₃ + SO₄

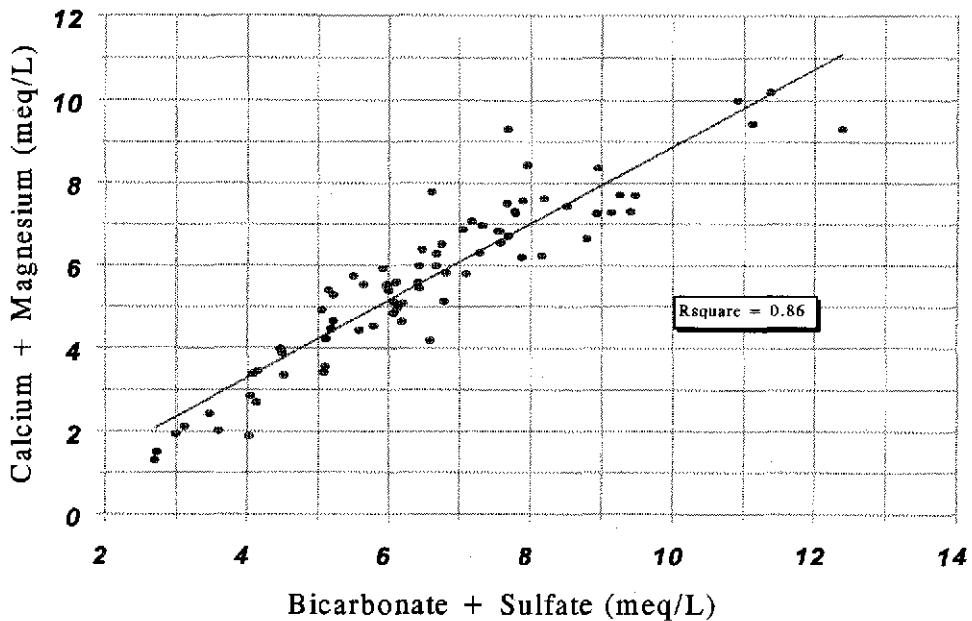


Figure 14. Linear relationship of bicarbonate plus sulfate (meq/L) to calcium plus bicarbonate (meq/L). Goodness of fit is represented by r^2 , which was calculated by regression analyses.

The potential for cation exchange as a cause of calcium depletion was tested by plotting the ratio of calcium plus magnesium divided by bicarbonate plus sulfate to the ratio of sodium divided by chloride (Figure 15). The ratio involving calcium should equal 1.0 in the case of dissolution of limestone and gypsum and in the absence of chemical reactions along the ground-water flow path. It is interesting to note that at the point of $x=1$ for the relationship involving calcium, the sodium to chloride ratio is also approximately one ($y=1$). Extending 15 % in either direction does not result in a substantial change in the sodium to chloride ratio; however, an increase is noted moving left in the area of $x=0.85$ and large increases begin at $x=0.75$. It is evident from the graph that cation exchange is affecting the chemistry in many of the well-water samples, resulting in an increase in sodium at the expense of calcium. Figure 16 is the same graph as Figure 15 with the addition of study area locations and demonstrates that most of the wells to the left of the $x=0.75$ line are wells located in Area III. This relationship suggests that cation exchange processes are more active in Area III. It is possible that the water in Area III has chemically evolved to a greater degree as a result of an increased resident time in the aquifer. Another possibility is an increase in the amount of clay minerals in the soils or aquifer in Area III. In either case, the difference in the water chemistry of Area III to that of Areas I and II, which is noted in the Piper diagram and review of Tables 2 through 4, is further demonstrated by the relationship depicted in Figure 16. To test the theory of cation exchange, an exercise similar to the one used in Figure 14 for testing gypsum as an additional source of calcium was employed by plotting calcium plus magnesium plus sodium versus bicarbonate plus chloride (Figure 17). Again, the r-squared value has improved demonstrating the effect of cation exchange on calcium concentration.

Figures 14 and 17 show that depletion of calcium through cation exchange and enrichment of calcium through dissolution of gypsum both impact calcium concentrations to a similar degree ($r^2=0.83$ and $r^2=0.86$, respectively). Plotting the total cations and anions from both graphs results in the relationship depicted in Figure 18. Because iron and potassium, both present in smaller quantities, are the only other cations missing, and bicarbonate, chloride and sulfate are the only major anions, the graph results in an r-squared value similar to that of Figure 10 for total cations versus anions. All of the multiple-ion graphs merely provide a convenient methodology for assessing the chemical evolution of the ground water within the study areas.

Precision and Accuracy of Water Quality Data

A quality-control program for checking the precision of the laboratory and field analyses was applied to the data for the present study. The data were entered into a spreadsheet program developed by Department personnel according to procedures set forth in Section 1030 of Standard Methods for the Examination of Water and Wastewater, 18th Edition (APHA, 1992). The procedures outlined in Standard Methods require analysis of the major cations and anions contributing to the TDS and uses this information to develop the following ratios: anion-cation balance, measured TDS versus calculated TDS, measured conductance versus calculated conductance, measured conductance versus ion sums, calculated TDS to conductance, and measured TDS to conductance. The procedures outline methods for calculating TDS and conductance and also cite a recommended ranges of values for the above-outlined ratios, which indicate correctness of analyses.

Ca + Mg/(HCO₃ + SO₄) versus Na/Cl

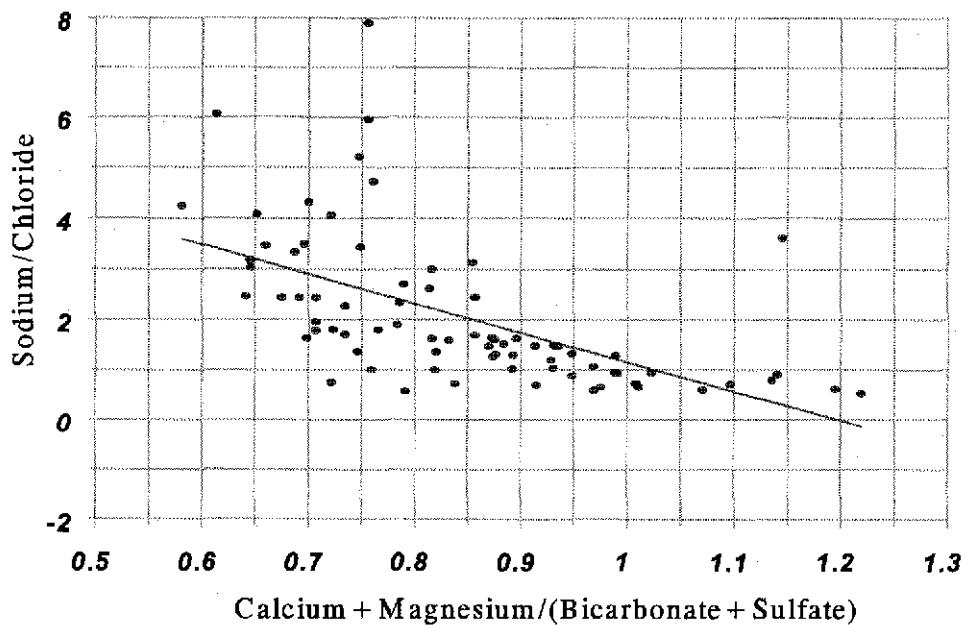


Figure 15. Linear relationship of calcium plus magnesium divided by bicarbonate and sulfate to sodium dived by chloride. Increase in the sodium to chloride ratio is interpreted as an increase in sodium with the depletion of calcium through cation exchange processes.

Ca + Mg/(HCO₃ + SO₄) versus Na/Cl

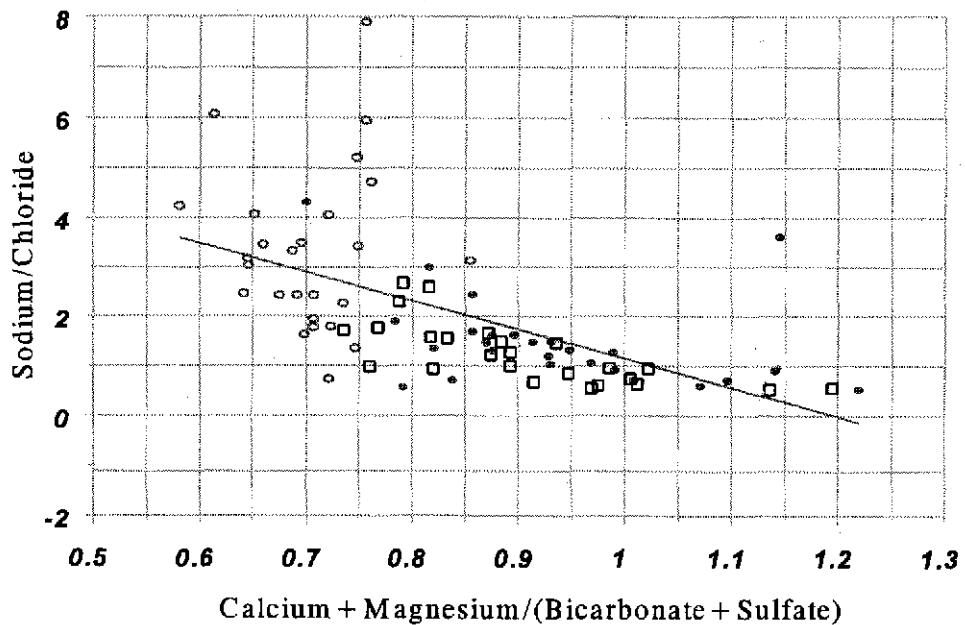


Figure 16. Same relationship as expressed in Figure 15. Symbols have been added to indicate the area represented by each of the sample locations. Area I wells are represented by solid circles; Area II wells by squares; and Area III wells by open circles.

Ca + Mg + Na versus HCO₃ + Cl

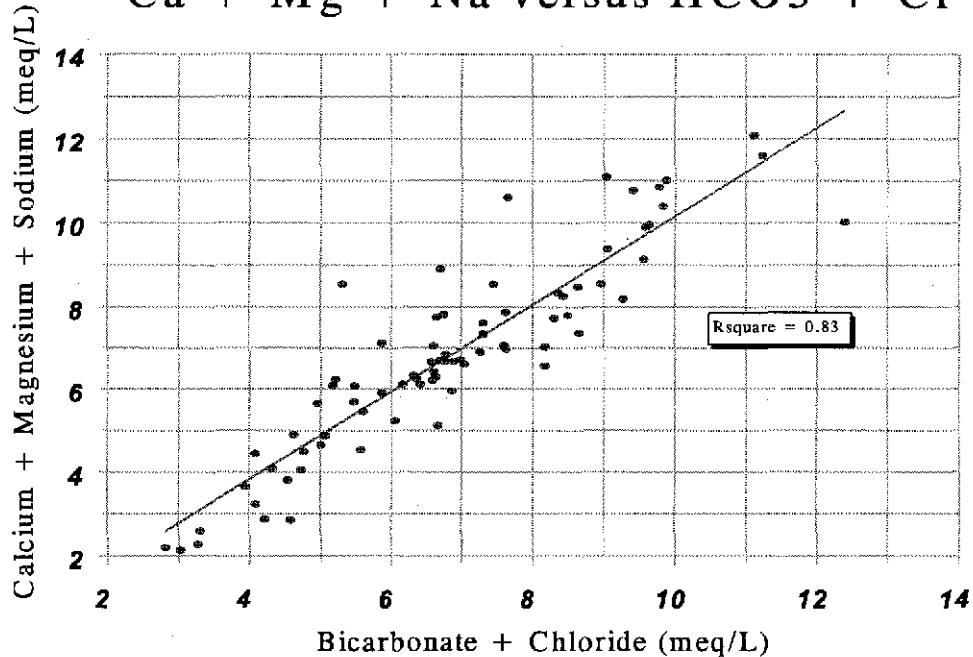


Figure 17. Linear relationship of calcium plus magnesium plus sodium to bicarbonate plus chloride in meq/L. Goodness of fit is represented by r^2 , which was calculated by regression analyses.

Ca + Mg + Na versus HCO₃ + Cl + SO₄

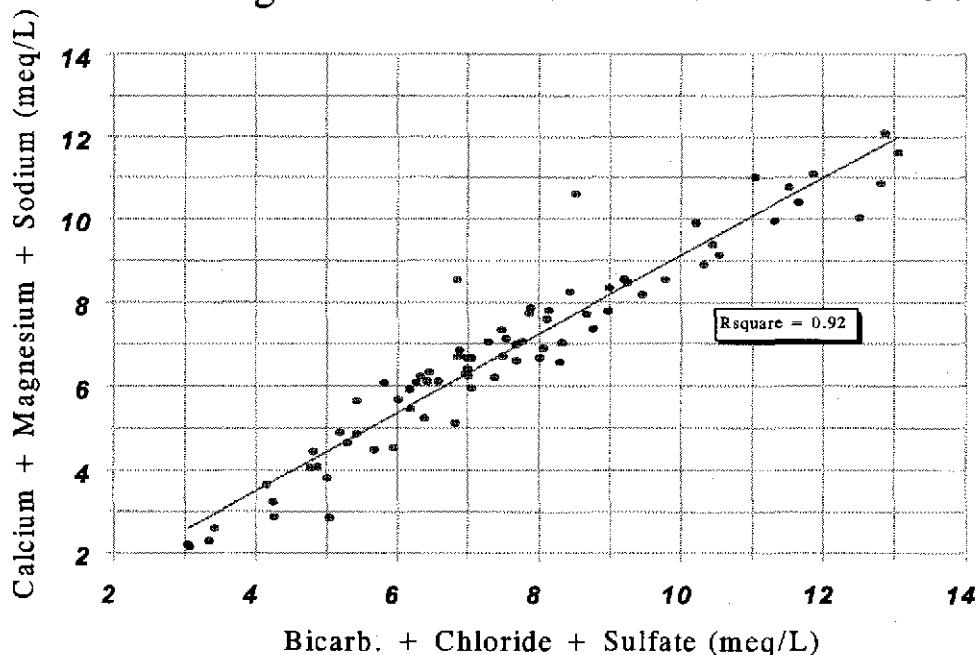


Figure 18. Linear relationship of calcium plus magnesium and sodium to bicarbonate plus chloride and sulfate (meq/L). Goodness of fit is represented by r^2 , which was calculated by regression analysis.

Statistical results from the spreadsheet program are in Appendix II and a summary of the results is provided in Table 10. The ratios most applicable to checking analytical accuracy are those involving the measured (residue weight) to calculated TDS ratios and the cation to anion ratios. The other ratios involving conductance are dependent on the accuracy of the field measurements and do not serve as a direct indicator of laboratory accuracy and precision; however, the ratios do provide an excellent means of identifying and/or confirming which analytes are in error as identified by other means.

A review of the ratios of measured to calculated TDS demonstrates that all values were within the recommended range with the exception of three sites: JF04, JF14 and PH07. The excellent correlation between the measured and calculated TDS is graphically displayed in Figure 19, which reveals an r-squared value of 0.98. Because of this relationship and the fact that waters with moderate TDS concentrations (250-1000 mg/L) are electrically neutral, it follows that the cation/anion ratio should also have a close relationship. However, the ratios of total cations (meq/L) to anions (meq/L) display a wider deviation from the accepted ranges (Table 10) than that of the relationship of the TDS values, although the r-squared value of 0.94 for the regression analysis performed on cations versus anions (see Figure 10) demonstrates a very good linear relationship between the two variables. The largest percentage error for the cation/anion ratios are for sites JF04 and PH07, which also display the largest deviation from the measured to calculated TDS ratios. Inspection of both the data and graphs such as Figures 19 and 10 can be used to determine which samples and which ion pairs are unbalanced, but does not indicate which value is in error. For instance, inspection of the data for JF04 reveals that milliequivalent concentrations of calcium and magnesium are approximately equal to sulfate and bicarbonate, whereas sodium (3.1 meq/L) is 3 times higher than chloride (0.86 meq/L), although it is not evident which value is in error.

To determine which analyte values were in error, a combination of techniques was applied including a review of the data and an inspection of graphs for cations versus conductance and anions versus conductance (figures 20 and 21, respectively). The total milliequivalents per liter for either cations or anions multiplied by 100 agrees closely with conductance in microSiemens per centimeter (Hem, 1989; APHA, 1992). The graphs can assist in determining whether the cations or anions are low and used in combination with the analytical results often can isolate an individual analyte that might be in error. For instance, an inspection of Figure 20 shows that JF04 falls very close to the best-fit line for cations versus conductance; whereas, Figure 21 shows that JF04 falls far below the line indicating that the anions are too low or the conductance is too high. Because both graphs have the same value for conductance and the TDS/Conductance ratio (column 2, Table 10) is within the accepted range, it can be assumed that the conductance values are reasonable and the anions are too low. With the information provided in the previous paragraph, the chloride value is probably suspect. The same technique can be used without the aid of the graphs by simply inspecting the ratios for both cations and anions versus conductance (Table 10). These ratios for JF04 were 0.96 for cations versus conductance and 0.74 for anions versus conductance. The latter value falls outside the accepted range of 0.9 to 1.1 and suggests that there is an error in the anion values and also that the anions are low with respect to the ratio.

Table 10. Statistical results of precision control analyses of laboratory data.

Location	TDS (meas)/ TDS (calc)	Cond (meas)/ Cond (calc)	TDS/Cond Calculated	TDS/Cond Measured	Cation/Anion % difference	Cation/Cond	Anion/Cond
	0.9-1.1	0.9-1.1	0.55-0.7	0.55-0.7	<2%	<5%	0.9-1.1
ARO1	0.97	0.8	0.63	0.49	1.18		0.87
ARO2	0.99	0.82	0.64	0.52	0.88		0.89
AR03	1.02	0.84	0.62	0.53	2.38		0.92
JF01	0.99	0.85	0.6	0.5	3.63		1
JF02	1.04	0.83	0.66	0.57		3.79	0.87
JF03	1.01	0.81	0.61	0.5	0.14		0.9
JF04	1.14	0.72	0.63	0.52		12.73	0.96
JF05	0.98	0.82	0.62	0.5		2.31	0.88
JF07	0.99	0.81	0.65	0.52		0.24	0.98
LN01	1	0.82	0.63	0.52	0.41		0.91
LN02	1	0.81	0.64	0.52	2.06		0.9
LN03	1	0.75	0.65	0.48	3.4		0.86
JF06	0.99	0.83	0.62	0.51	1.88		0.89
JF08	0.99	0.81	0.73	0.58	4.63		0.84
JF09	0.97	0.75	0.76	0.56	10.57		0.77
JF10	1	0.8	0.63	0.5	0.65		0.94
JF11	0.97	0.84	0.64	0.52	2.37		0.94
JF12	1	0.82	0.63	0.51	0.85		1
JF13	1	0.82	0.62	0.51	0		0.94
JF14	0.88	0.85	0.68	0.5	7.11		1
JF15	0.9	0.84	0.66	0.5	3.61		0.93
JF16	0.99	0.85	0.63	0.53	1.11		1.01
JF17	0.98	0.88	0.64	0.55	0.2		0.96
JF18	0.97	0.84	0.63	0.5	1.63		0.92
LN04	1	0.81	0.64	0.52	5.17		0.89
LN05	1.03	0.8	0.64	0.53		4.19	0.85
LN06	0.97	0.83	0.64	0.52	5		0.96
DE01	1.03	0.83	0.62	0.53	2.63		0.92
DE02	0.97	0.9	0.65	0.57		6.71	0.92
DE03	1	0.86	0.65	0.56		3.39	0.95
DE04	1	0.85	0.64	0.54		5.08	0.9
DE05	1	0.84	0.65	0.55		3.25	0.91
DE06	1	0.87	0.62	0.54		4.18	0.91
DE07	1	0.84	0.64	0.53	5.38		0.86
LN07	0.98	0.76	0.71	0.53	1.29		0.83
DE08	0.99	0.75	0.69	0.5	2.25		0.81
DE09	1	0.77	0.65	0.5	3.41		0.86
DE10	1	0.87	0.66	0.57	3.73		0.9
DE11	1.02	0.87	0.63	0.56	3.98		0.9
DE12	1	0.81	0.66	0.55	5.44		0.84
DE13	1	0.82	0.65	0.53	4.72		0.89
DE14	1.04	0.8	0.63	0.52	7.15		0.81
DE15	0.99	0.84	0.65	0.55	5.02		0.92
							1.01

Table 10. Statistical results of precision control analyses of laboratory data.

Location	TDS (meas)/ TDS (calc)	Cond (meas)/ Cond (calc)	TDS/Cond Calculated	TDS/Cond Measured	Cation/Anion % difference	Cation/Cond	Anion/Cond
	0.9-1.1	0.9-1.1	0.55-0.7	0.55-0.7	<2% <5%	0.9-1.1	0.9-1.1
DR01	0.97	0.81	0.71	0.55	5.97	0.82	0.93
JF19	1.02	0.88	0.63	0.56	3.46	0.95	1.02
DE20	1.05	0.88	0.64	0.59	3.04	0.96	1.02
DE21	1.06	0.87	0.67	0.62	3.56	0.95	1.02
PH01	1	0.89	0.65	0.58	1.16	1.04	1.06
PH02	1.04	0.89	0.63	0.58	2.23	1.04	1
PH03	0.98	0.89	0.63	0.55	0.18	1.04	1.04
PH04	1	0.85	0.65	0.54	2.49	1	1.05
PH05	1	0.9	0.64	0.58	1.51	1	1.04
PH06	0.96	0.83	0.65	0.51	1.79	0.96	1
PH07	1.13	0.86	0.62	0.6	11.76	1.09	0.86
PH08	0.98	0.92	0.65	0.58	1.78	1.05	1.09
PH09	0.94	0.92	0.65	0.56	2.31	1.08	1.13
PH10	0.99	0.87	0.66	0.56	1.75	1	1.04
PH11	1	0.95	0.66	0.63	4.82	1.03	1.14
PH12	0.97	0.91	0.66	0.58	3.64	1	1.08
PH13	1	0.93	0.65	0.6	0.65	1.04	1.05
PH14	1.02	0.94	0.65	0.62	2.17	1.04	1.08
PH15	1	0.84	0.67	0.57	7	0.99	1.14
PH16	1.01	0.94	0.66	0.63	1.97	1.07	1.16
PH17	0.96	0.77	0.67	0.5	5.53	0.88	0.98
PH18	0.96	0.84	0.7	0.56	7.05	0.92	1.06
PH19	0.93	0.9	0.69	0.57	3.78	0.98	1.05
PH20	1	0.96	0.62	0.59	0.39	1.15	1.16
PH21	0.98	0.94	0.63	0.58	0.53	1.13	1.11
PH22	0.98	0.92	0.64	0.58	1.24	1.1	1.13
PH23	0.96	0.91	0.65	0.57	1.29	1.08	1.1
PH24	0.97	0.95	0.64	0.6	0.96	1.15	1.13
PH25	0.95	1	0.65	0.61	0.46	1.15	1.14
PH26	0.97	0.99	0.65	0.62	0.42	1.2	1.19

TDS versus TDS (Calculated)

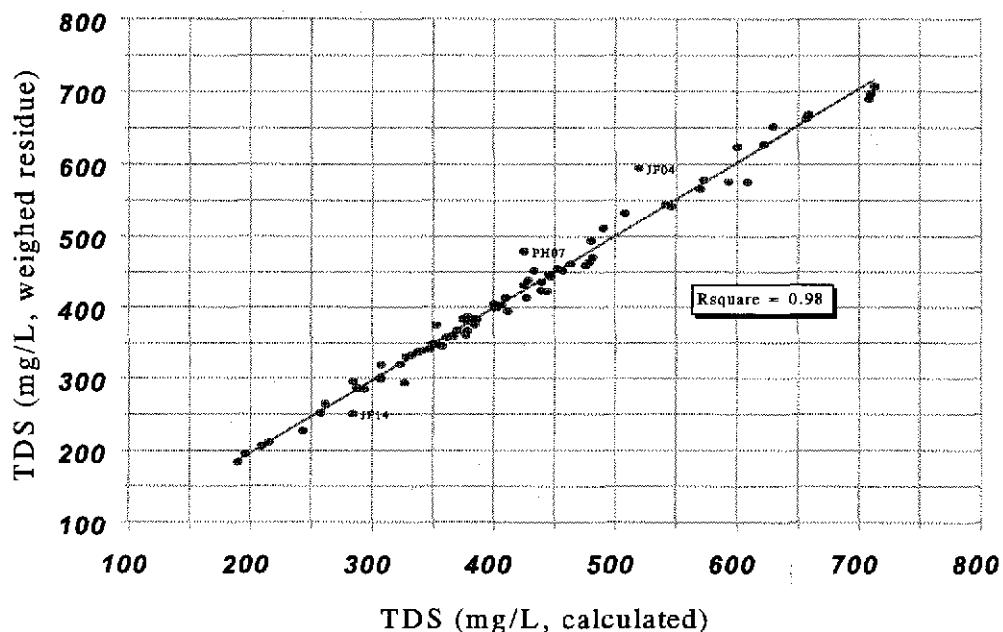


Figure 19. Linear relationship between TDS as measured in the laboratory (weighted residue) to calculated TDS (sum of the individual parameters). Goodness of fit represented by r^2 , which was calculated by regression analysis.

Cations vs. Conductance

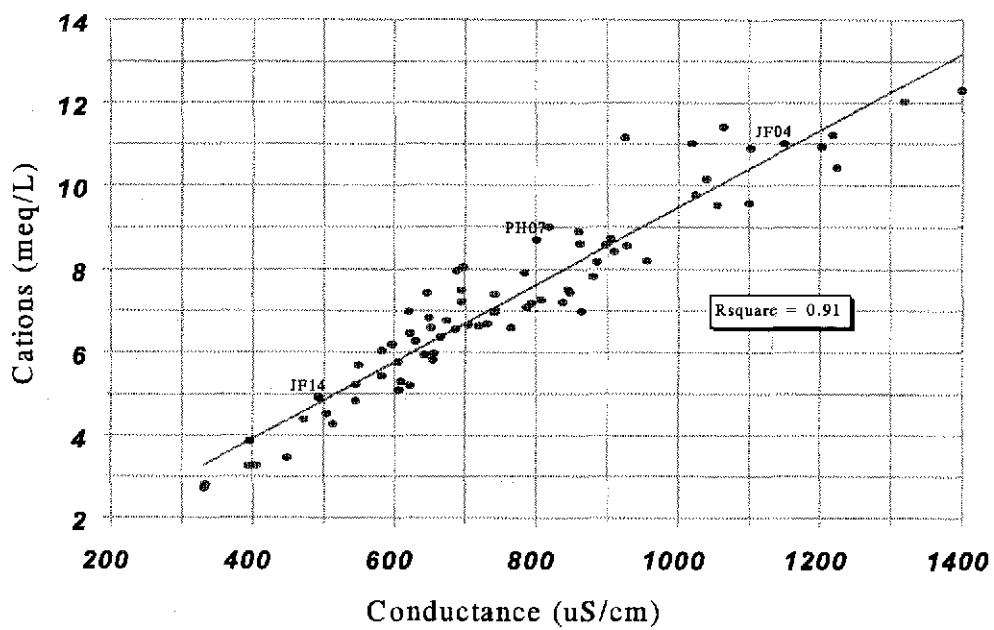


Figure 20. Linear relationship of total cations (meq/L) to conductance ($\mu\text{S}/\text{cm}$). Goodness of fit is represented by r^2 , which was calculated by regression analysis.

Anions vs. Conductance

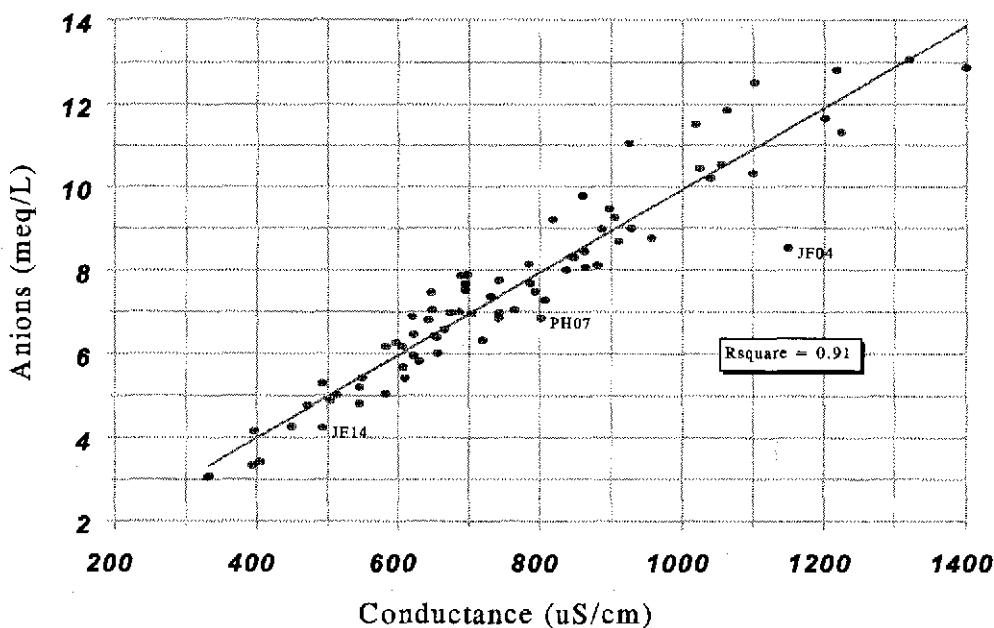


Figure 21. Linear relationship of total anions (meq/L) to conductance ($\mu\text{S}/\text{cm}$). Goodness of fit is represented by r^2 , which was calculated by regression analysis.

The methodology described above was applied to eight samples with cation/anion ratios ≥ 7 and the results are presented in Table 11. The difference in milliequivalent concentrations of major ion pairs will indicate which pair deviates the most from unity. The anions or cations versus conductance can then be used to isolate which ion is in error. The anions and cations versus conductance ratios were effective in potentially identifying seven cases where either the anion and/or cation value was in error (high or low), and, in conjunction with the inspection of individual ion pairs from the data, demonstrated which analyte was in error. The method was less effective in the case where both the cations and anions versus conductance varied from a ratio of unity by the same degree. Potential uses for this method, in addition to identifying possible laboratory errors, is the ability to identify one or two parameters for re-analysis instead of re-analyzing the sample for the entire suite of analytes.

The ratios involving measured versus calculated conductance and measured TDS versus measured conductance (Table 10) demonstrate that measured conductance values were consistently low. This is supported by the fact that measured TDS to calculated conductance values were consistently within the accepted range. Subsequent calibration checks indicated that the conductivity meter was providing low values in reference to the laboratory standard solutions used to calibrate the instrument. Figures 22 and 23 reveal that the error was consistent between all sampling sites, as demonstrated by the high r -squared values and close fit of the data points to the best-fit line.

Table 11. Results of applied methodology for determining possible analytical errors in data set.

Sample	Cations meq/L	Anions meq/L	Calcium - Bicarbonate	Sodium - Chloride	Cation/ Conductivity	Anion/ Conductivity	Potential Error
JF04	11.02	8.53	0.16	2.2	0.96	0.74	Chloride Low
JF09	3.47	4.27	2.15	0.75	0.77	0.95	Calcium Low
JF14	4.91	4.25	1.01	0.03	1.00	0.86	Calcium Low
DE14	6.99	8.06	0.44	0.72	0.81	0.93	Sodium Low
DE17	4.28	5.02	1.43	0.24	0.83	0.98	Calcium Low
PH07	8.7	6.85	1.17	0.53	1.09	0.76	Bicarbonate Low
PH15	10.89	12.52	3.09	0.61	0.99	1.14	Bicarbonate High
PH18	5.94	6.82	2.4	0.71	0.92	1.06	?

Measured vs Calculated Conductance

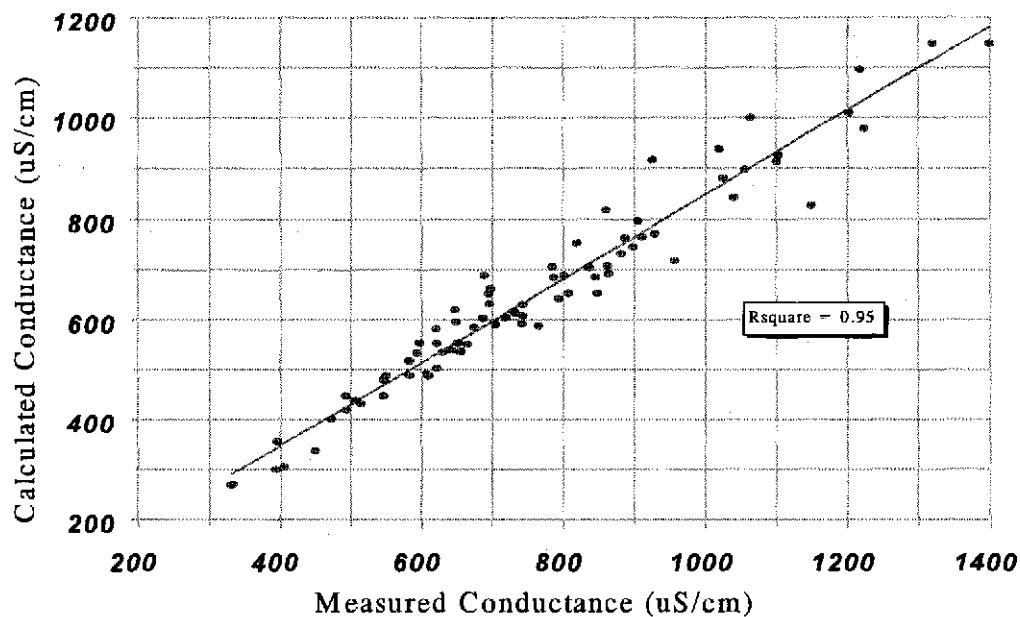


Figure 22. Linear relationship of calculated conductance to field-measured conductance ($\mu\text{S}/\text{cm}$). Goodness of fit represented by r^2 , which was calculated from regression analysis.

TDS versus Conductance

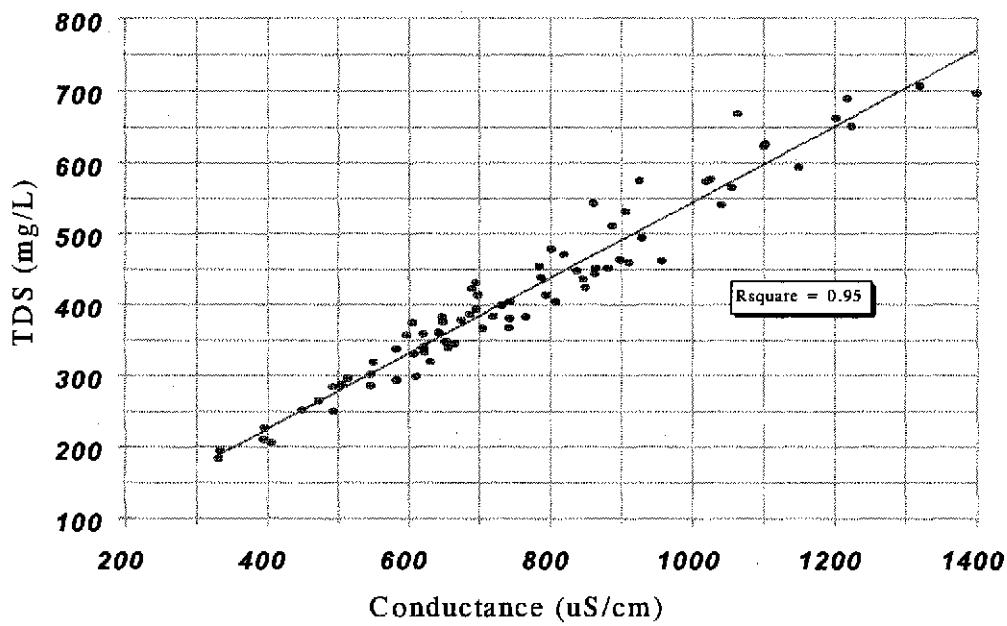


Figure 23. Linear relationship of measured TDS (mg/L) to measured conductance ($\mu\text{S}/\text{cm}$). Goodness of fit represented by r^2 , which was calculated by regression analysis.

SUMMARY AND CONCLUSIONS

A total of 77 wells, all of which were completed in the Mississippi Alluvial Aquifer, were sampled between July 29 and September 23, 1996, in areas within and surrounding Jefferson, Phillips and Desha Counties. Water quality analyses reveal the water to be of very good quality with the exception of high iron concentrations, which ranged from 0.75 to 70.5 mg/L in all of the samples. Areas of elevated chloride were noted in portions of Desha and Jefferson Counties, although all chloride concentrations were below the secondary MCL of 250 mg/L for chloride. Areas of elevated chloride concentrations in Desha County are similar to those first documented over 47 years ago in this area.

Pesticides were detected at 24 of the 77 sites. Pesticide concentrations ranged from 0.007 to 0.609 $\mu\text{g}/\text{L}$ and were below EPA-established MCLs and HALs for all of the detected pesticides. Bentazon was the most frequently-detected pesticide and was detected at 14 of the 24 sites with pesticide detections. The high solubility of bentazon apparently increases the potential for migration to the ground-water table.

Comparisons of pesticides detected in ground water and those detected in surface water revealed two interesting phenomena. First, high-solubility, low-adsorption pesticides were frequently detected in ground water but not in surface water. Secondly, low-solubility, highly sorptive pesticides were detected with a high frequency in surface water, whereas there were no detections in any of the ground water samples. Medium solubility and medium adsorption pesticides were found in both surface and ground waters. Transport of highly sorptive pesticides to surface water is facilitated by movement with eroded sediment, and unfiltered surface water samples commonly revealed positive detections for these pesticides. The absence of detections of highly soluble pesticides in surface water may be explained by rapid subsurface infiltration of the pesticides during the early stages of a rainfall event, leaving little available for transport by runoff water, and by the greater susceptibility of pesticides in surface water to photolysis.

Comparison of the pesticide analyses for the present study with analyses performed by both the AWRC and the USGS demonstrate a wide variation of the types of pesticides detected and the frequency (percentage) of overall detections. All three agencies are currently engaged in a series of meetings, which aim to standardize, where possible, a uniform list of pesticides, analytical techniques and reporting criteria. These efforts represent a commitment by all agencies to gather and report data, which can be compared on a state-wide basis.

Precision of water quality data was analyzed using various ratios involving TDS, conductance, total cations (meq/L) and total anions (meq/L). Overall precision was very good for the data, and efforts associated with the QA/QC process have resulted in the laboratory adopting the procedure for analyses checks prior to releasing the data to Department personnel.

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APPENDIX I

Water Quality Data

Station_ID	Date	pH	Temp Fahrenheit	Latitude	Longitude	Alkalinity mg/L CaCO ₃	Conductivity umhos/cm	Fluoride ug/L	Cd ug/L	Cr ug/L	Pb ug/L	As ug/L	Se ug/L	AI ug/L	Cu ug/L
NSGWAR01	960729	6.87	64.6	34.23	15.3	91.39	30.9	208	610	0.32	<1	<2	<5.0	<5.0	54.2 <2
NSGWAR02	960729	6.91	64.7	34.22	10.8	91.38	10.5	172	546	0.27	<1	<2	<5.0	<5.0	28.5 <2
NSGWAR03	960729	6.94	64.8	34.24	49.4	91.41	01.0	203	720	0.28	<1	<2	11.15	<5.0	52.9 <2
NSGWJF01	960729	6.88	64.7	34.24	39.7	91.48	17.0	245	630	0.27	<1	<2	11.32	<5.0	17.7 <2
NSGWJF02	960805	6.81	64.9	34.28	30.7	91.43	39.7	276	1101	0.21	<1	<2	<5.0	<5.0	<16 <3
NSGWJF03	960805	7	64.2	34.23	44.5	91.44	01.4	249	808	0.27	<1	<2	9.65	<5.0	<16 <2
NSGWJF04	960805	6.92	64.7	34.14	26.8	91.48	47.0	340	1150	0.25	<1	<2	<5.0	<5.0	<16 <2
NSGWJF05	960805	6.81	65.2	34.06	05.8	91.48	27.6	297	1400	0.27	<1	<2	<5.0	<5.0	<16 <2
NSGWJF07	960805	7.07	65.4	34.08	13.8	91.49	45.7	443	1040	0.32	<1	<2	<5.0	<5.0	<16 <2
NSGWLN01	960805	7.03	64.5	34.04	20.0	91.42	23.5	233	656	0.27	<1	<2	<5.0	<5.0	<16 <2
NSGWLN02	960805	6.88	64.1	34.07	36.4	91.43	03.7	297	794	0.21	<1	<2	6.83	<5.0	<16 <2
NSGWLN03	960805	6.95	64.8	34.09	06.0	91.43	18.3	389	957	0.21	<1	<2	5.79	<5.0	<16 <4.2
NSGWLN06	960729	7.05	64.6	34.06	28.0	91.50	06.4	235	882	0.27	<1	<2	<5.0	<5.0	<16 <2
NSGWF08	960812	6.75	65.2	34.10	02.2	91.58	26.6	135	334	0.22	<1	<2	<5.0	<5.0	<16 <2
NSGWF09	960812	6.96	64.5	34.07	37.9	91.57	16.7	200	450	0.33	<1	<2	5.43	<5.0	<16 <2
NSGWF10	960812	6.89	65.3	34.17	14.8	91.58	29.8	314	741	0.31	<1	<2	<5.0	<5.0	<16 <2
NSGWF11	960812	6.81	64.9	34.17	14.5	91.48	20.9	294	705	0.26	<1	<2	<5.0	<5.0	<16 <2
NSGWF12	960812	7.02	65.5	34.11	16.7	91.42	32.3	409	863	0.25	<1	<2	<5.0	<5.0	<16 <2
NSGWF13	960812	6.87	65.3	34.10	06.9	91.36	15.4	285	743	0.31	<1	<2	<5.0	<5.0	<16 <2
NSGWF14	960813	6.76	64.9	34.24	14.0	91.54	54.1	166	494	0.15	<1	<2	14.34	<5.0	<16 <3.6
NSGWF15	960813	6.65	64.7	34.22	37.8	91.55	47.7	157	583	0.13	<1	<2	8.25	<5.0	<16 <3.1
NSGWF16	960813	6.97	65	34.14	01.9	91.42	07.6	305	652	0.3	<1	<2	<5.0	<5.0	<16 <2
NSGWF17	960813	7.15	64.8	34.17	48.3	91.50	07.3	196	546	0.22	<1	<2	<5.0	<5.0	<16 <2
NSGWF18	960813	7.15	64.9	34.26	50.6	91.51	29.4	305	911	0.24	<1	<2	<5.0	<5.0	<16 <2
NSGWLN04	960819	6.8	65.9	34.00	16.2	91.26	42.0	348	846	0.27	<1	<2	<5.0	<5.0	<16 <2
NSGWLN05	960819	6.83	66.2	34.00	30.7	91.30	00.0	295	1224	0.25	<1	<2	<5.0	<5.0	<16 <65.3
NSGWLN06	960819	7.17	65.8	34.03	36.2	91.30	53.6	437	898	0.25	<1	<2	<5.0	<5.0	<16 <2
NSGWDE01	960819	6.93	66.4	33.53	49.4	91.27	55.5	327	929	0.3	<1	<2	<5.0	<5.0	<16 <2
NSGWDE05	960819	6.99	65.6	33.54	28.6	91.23	23.3	312	1218	0.25	<1	<2	<5.0	<5.0	<16 <2
NSGWDE06	960819	6.89	64.5	33.50	35.9	91.25	20.6	307	1320	0.25	<1	<2	<5.0	<5.0	<16 <2
NSGWDE07	960819	6.78	65.7	33.55	48.2	91.30	26.2	247	838	0.2	<1	<2	<5.0	<5.0	<16 <2
NSGWLN07	960826	6.63	66.3	33.49	19.8	91.35	12.6	147	395	0.14	0.08	0.56	<4	<5.0	<16 <2
NSGWDE08	960826	6.67	65.3	33.49	02.9	91.31	49.5	151	406	0.23	<0.4	0.41	<4	<5.0	<16 <2
NSGWDE09	960826	6.98	65.1	33.50	29.3	91.29	58.4	300	765	0.21	<0.4	<4	<5.0	<5.0	<16 <2
NSGWDE10	960826	7	74.5	33.45	57.1	91.25	39.0	175	505	0.2	<0.4	<4	<5.0	<5.0	<16 <2
NSGWLN11	960826	6.89	74.2	33.45	43.4	91.26	06.5	229	787	0.17	<0.4	<4	<5.0	<5.0	<16 <2

Station_ID	Date	pH	Temp Fahrenheit	Latitude	Longitude	Alkalinity mg/L CaCO ₃	Conductivity umhos/cm	Fluoride ug/L	Cd ug/L	Cr ug/L	Pb ug/L	As ug/L	Se ug/L	AI ug/L	Cu ug/L
NSGWDE12	960826	6.92	65.3	33 44 17.6	91 23 46.2	210	607	0.22	<.04	<.4	<.4	<5.0	<16	<2	
NSGWDE13	960826	6.78	64.7	33 42 15.6	91 25 42.7	263	655	0.21	<.04	<.4	<.4	<5.0	<16	<2	
NSGWDE14	960826	6.89	64.6	33 52 38.2	91 32 08.9	259	865	0.2	<.04	<.4	<.4	<5.0	<16	<2	
NSGWDE15	960902	6.74	65.5	33 47 39.0	91 27 41.1	300	732	0.28	<.04	<.4	<.4	<5.0	<16	2.9	
NSGNDR01	960902	7.11	65.7	33 45 11.4	91 33 11.4	124	332	0.33	<.04	<.4	<.4	<5.0	<16	55.6	
NSGMWF19	960903	7.28	64	34 20 58.2	91 46 27.2	274	687	0.29	<.04	<.4	<.4	5.08	<16	<2	
NSGWDE16	960903	7.34	64.9	33 46 52.1	91 23 41.5	236	622	0.24	<.04	<.4	<.4	<5.0	<16	2.7	
NSGWDE17	960903	6.92	65	33 47 07.9	91 24 28.1	184	514	0.17	<.04	<.4	<.4	<5.0	<16	<2	
NSGWDE18	960903	7.23	65.1	33 56 17.4	91 22 15.1	224	473	0.15	<.04	<.4	<.4	<5.0	<16	<2	
NSGWDE19	960903	6.93	65	33 48 14.1	91 19 15.6	310	887	0.2	<.04	<.4	<.4	<5.0	<16	<2	
NSGWDE20	960903	7.59	64.8	33 47 34.4	91 19 16.4	322	906	0.25	<.04	<.4	<.4	<5.0	<16	2.1	
NSGWDE21	960903	7.41	64.5	33 49 38.9	91 19 10.4	261	605	0.23	<.04	<.4	<.4	<5.0	<16	<2	
NSGWP01	960909	6.84	64.3	34 30 00.2	90 53 17.4	285	583	0.27	<.04	<.4	<.4	<5.0	<16	<2	
NSGWP02	960909	7.51	63.9	34 33 06.1	90 53 18.9	230	550	0.29	<.04	<.4	<.4	6.17	<16	<2	
NSGWP03	960909	7.71	63.8	34 36 39.3	90 58 34.0	292	622	0.29	<.04	<.4	<.4	<5.0	<16	<2	
NSGWP04	960909	7.19	66.8	34 24 42.6	90 39 30.1	376	743	0.27	<.04	<.4	<.4	<5.0	<16	<2	
NSGWP05	960910	7.42	65	34 30 31.2	90 58 40.2	308	785	0.29	<.04	<.4	<.4	<5.0	<16	<2	
NSGWP06	960910	7.38	71.2	34 28 10	90 58 08	300	665	0.28	<.04	<.4	<.4	<5.0	<16	<2	
NSGWP07	960910	7.12	66.6	34 25 42	90 57 31	254	801	0.26	<.04	<.4	<.4	<5.0	<16	2.3	
NSGWP08	960910	7.54	65.6	34 27 34	90 47 09	310	649	0.23	<.04	<.4	<.4	<5.0	<16	<2	
NSGWP09	960910	7.15	68.4	34 31 18	90 42 07	464	1020	0.27	<.04	<.4	<.4	9.85	<16	2.3	
NSGWP10	960917	6.63	63.5	34 21 27.9	90 47 20.5	322	674	0.25	<.04	<.4	<.4	9.5	<20	2.5	
NSGWP11	960917	7.32	63.2	34 21 29.1	90 49 16.1	354	861	0.26	<.04	<.4	<.4	5	<16	2.7	
NSGWP12	960917	7.25	62.3	34 23 57.1	90 50 25.1	242	493	0.24	<.04	<.4	<.4	<5.0	<16	4.8	
NSGWP13	960917	7.81	63.9	34 20 12.6	90 58 34.3	250	597	0.18	<.04	<.4	<.4	<5.0	71	2.2	
NSGWP14	960917	7.72	63.9	34 18 39.9	90 59 56.5	282	695	0.12	<.04	0.73	<.4	<5.0	41	2.6	
NSGWP15	960917	7.26	64.8	34 14 49.7	90 54 36.7	614	1103	0.23	<.04	<.4	<.4	14.9	<16	3.9	
NSGWP16	960917	7.57	64.2	34 17 17	90 55 15	416	1064	0.23	<.04	<.4	<.4	5.1	<16	2.5	
NSGWP17	960917	7.34	70.1	34 15 24	90 51 28	402	849	0.18	<.04	<.4	<.4	31.1	<5.0	3.5	
NSGWP18	960917	7.65	63.8	34 25 38	90 43 39	322	643	0.31	<.04	<.4	<.4	<5.0	<16	2.7	
NSGWP19	960917	7.64	67.5	34 25 24	90 46 37	194	397	0.25	<.04	<.4	<.4	7.9	<5.0	16.7	
NSGWP20	960923	7.55	63.5	34 30 55.2	90 38 29.3	358	647	0.32	<.04	<.4	<.4	16.8	<5.0	10.6	
NSGWP21	960923	7.57	63.7	34 29 36.3	90 39 26.0	332	620	0.31	<.04	<.4	<.4	<5.0	<5.0	<2	
NSGWP22	960923	7.55	64	34 27 33.7	90 37 15.2	436	819	0.29	<.04	<.4	<.4	<5.0	<5.0	<2	
NSGWP23	960923	7.47	63.9	34 26 56.1	90 36 22.1	376	696	0.29	<.04	<.4	<.4	5	<5.0	<2	
NSGWP24	960923	7.44	63.8	34 15 25.7	91 02 04.2	376	698	0.18	<.04	0.9	<.4	<5.0	<5.0	<2	
NSGWP25	960923	7.72	63.6	34 13 25.9	90 59 58.7	328	689	0.2	<.04	<.4	<.4	<5.0	<5.0	<2	
NSGWP26	960923	7.64	64.1	34 07 28	91 01 59	488	926	0.18	<.04	<.4	<.4	<5.0	<5.0	<16	39.8

Station_ID	Ca mg/L	Fe ug/L	Fe mg/L	K mg/L	Mg mg/L	Mn ug/L	Na mg/L	Zn ug/L	Ni ug/L	B ug/L	Be ug/L	Ba ug/L	Co ug/L
NSGWAR01	67.7	11900	11.9	0.6	6.1	335	22.7	<2.0	194	<5	31.1	5.6	271.8
NSGWAR02	57.8	10600	10.6	0.6	7	672	22.7	2.9	173	<5	32.7	5.1	238.5
NSGWAR03	87.4	10700	10.7	0.5	12.6	802	19.4	4.4	270	<5	33.2	5.1	313.6
NSGWJF01	83.5	5060	5.06	0.4	13.6	694	18.2	<2.0	265	<5	43.6	2.6	480.1
NSGWJF02	118	17900	17.9	1.4	17	820	37.1	<2.0	365	<5	7136	11.9	444.8
NSGWJF03	81.7	5120	5.12	1.2	17.7	482	35	<2.0	277	<5	7137	3.6	565.8
NSGWJF04	114	10400	10.4	1.5	22.1	339	71.3	<2.0	376	<5	7138	7.2	585.1
NSGWJF05	132	4680	4.69	1.3	33.1	1740	64	2.7	466	<5	7139	3.5	625.8
NSGWJF07	111	6430	6.43	1.2	26.3	1060	50.6	<2.0	385	<5	7141	4.3	492.8
NSGWLND1	70.5	7100	7.1	1	11.5	237	28.2	<2.0	223	<5	7142	4.7	337
NSGWLN02	94.5	12200	12.2	1.2	9	605	29	<2.0	273	<5	7143	7.9	361.7
NSGWLN03	115	22300	22.3	1.2	5.6	1500	26.8	<2.0	310	<5	7144	14.2	499.6
NSGWJF06	83.7	5380	5.38	0.9	19	562	42.8	<2.0	287	<5	7140	3.7	346.8
NSGWJF08	30.2	17900	17.9	1	0.03	479	14.7	<2.0	50	<5	33.1	6.4	160
NSGWJF09	37	15400	15.4	1.7	0.5	616	22.6	3.5	94	<5	72	8.5	292.1
NSGWJF10	96.2	6830	6.83	0.8	14.4	402	16.4	<2.0	300	<5	49.3	2.9	368.9
NSGWJF11	84.3	9270	9.27	0.8	10.6	1080	28.1	2.4	254	<5	55.5	3.8	376
NSGWJF12	128	9260	9.26	0.7	15	1750	14.6	<2.0	381	<5	80	5.7	362.6
NSGWJF13	84	8240	8.24	0.5	21	1580	17.7	<2.0	296	<5	63	5.3	423.1
NSGWJF14	48.7	45600	45.6	1.6	0.03	435	18.5	<2.0	26	<5	<3	15.1	400.1
NSGWJF15	40.5	70500	70.5	2.2	0.03	725	19.1	<2.0	<	<5	<3	29.1	413.8
NSGWJF16	99.3	12800	12.8	0.6	7.8	379	11.9	<2.0	280	<5	51.1	6.2	366.3
NSGWJF17	70.6	8510	8.51	1.1	5.6	542	21	<2.0	199	<5	28.7	4.3	452.6
NSGWJF18	112	18600	18.6	1.1	9.6	729	30.9	2.7	319	<5	46.9	9.5	546.2
NSGWLN04	102	12900	12.9	0.8	8.5	1010	28.7	<2.0	289	<5	51.9	6.9	437.2
NSGWLN05	105	12200	12.2	1.4	15.9	1640	78.4	3.1	328	<5	70	10.1	590.9
NSGWLN06	113	10400	10.4	1	19.7	621	21.2	4.6	363	<5	55.3	5.8	401.4
NSGWDE01	106	5370	5.37	0.8	21.6	720	29.3	<2.0	354	<5	44.2	3.1	553.3
NSGWDE02	113	9120	9.12	1.3	25.2	649	72.3	<2.0	386	<5	43.6	5.1	517.4
NSGWDE03	99.2	10000	10	1.5	20.7	603	62.7	<2.0	333	<5	42.3	5.6	342.1
NSGWDE04	122	9850	9.85	2	18	736	35.9	<2.0	379	<5	39.5	5.3	467.5
NSGWDE05	115	13400	13.4	2	20.6	1120	68.3	<2.0	372	<5	54.3	6.6	803.7
NSGWDE06	127	10400	10.4	1.5	25.6	359	72.9	<2.0	423	<5	50.2	5.2	649.3
NSGWDE07	83.4	14000	14	0.8	8.3	531	42.2	<2.0	242	<5	69	10.4	499.6
NSGWLND7	38.9	26700	26.7	0.9	0.03	538	7.9	<2.0	50	<5	44.9	9.8	258.7
NSGWDE08	42.4	18100	18.1	1	0.03	473	11.1	<2.0	91	<5	38.1	6.8	244.3
NSGWDE09	83.3	16700	16.7	1.4	5.9	509	30.2	2.6	232	<5	48.5	6	485.7
NSGWDE10	51.3	11600	11.6	1.2	3.6	566	28.3	2.4	143	<5	32	4.2	338.1
NSGWDE11	79.9	12300	12.3	1.7	8.2	460	44.8	<2.0	233	<5	40.8	4.5	519.6

Station_ID	Ca mg/L	Fe ug/L	K mg/L	Mg mg/L	Mn ug/L	Na mg/L	Zn ug/L	Ni ug/L	B ug/L	Be ug/L	Ba ug/L	Co ug/L
NSGWDE12	64.9	16300	16.3	0.8	3.9	1250	21.4	<2.0	178	<5	46.4	5.8
NSGWDE13	75.7	15300	15.3	1.1	8	602	18.5	<2.0	222	<5	50.7	5.6
NSGWDE14	89.6	1500	1.5	1.2	13	841	31.3	<2.0	277	<5	70	6.4
NSGWDE15	82.6	12600	12.6	1.1	12.3	598	24.7	5	257	<5	46.7	4.6
NSGWDR01	26.2	13700	13.7	0.9	0.03	294	20.6	2.3	54	<5	45.9	4.8
NSGWJF19	80.7	7960	7.96	0.9	11.2	635	29.9	2.3	248	<5	31.9	2.8
NSGWDE16	68.4	17700	17.7	1	0.1	992	25.8	11.5	171	<5	36.6	5.9
NSGWDE17	50.4	12300	12.3	1.2	2.3	667	25.4	4.4	135	<5	31.5	4.2
NSGWDE18	57.5	8800	8.8	0.9	5.9	372	16.2	5.1	168	<5	41.3	3.1
NSGWDE19	105	9480	9.48	1.5	12.6	1090	34.8	<2.0	314	<5	50.2	3.2
NSGWDE20	110	6050	6.05	1.3	16.7	1350	37.1	8.1	343	<5	46.5	2.3
NSGWDE21	72.9	8300	8.3	0.9	10.9	1010	21.1	2.2	227	<5	31.4	3.2
NSGWP01	67.2	3370	3.37	0.6	24.7	237	12	2.3	270	<5	26.9	2.1
NSGWP02	57	753	0.753	0.5	25.3	834	16.8	3.6	247	<5	23.6	<2
NSGWP03	67.3	2860	2.86	0.5	25.7	295	19.9	5.6	274	<5	30.7	<2
NSGWP04	93.3	8850	8.85	1	25	543	8	142	336	<5	80	7
NSGWP05	86	2310	2.31	0.8	30.9	546	22.5	2.7	342	<5	17.8	<2
NSGWP06	72.4	6540	6.54	0.6	24	315	12.2	9.8	280	<5	24.6	3.8
NSGWP07	97.9	3810	3.81	0.9	35.1	757	17.6	188	389	<5	24.6	2.4
NSGWP08	83.1	3450	3.45	1.4	22.3	300	15.7	3.5	299	<5	27.9	2.3
NSGWP09	121	5970	5.97	1.1	50.5	302	13.3	6.5	510	<5	21.4	3.5
NSGWP10	82.8	7870	7.87	2.3	20.4	277	14.1	<2.0	291	<5	16.8	5.2
NSGWP11	105	8720	8.72	1.9	25.1	488	28.4	<2.0	366	<5	6.8	5.8
NSGWP12	62.5	6790	6.79	1.5	13.5	327	9.8	<2.0	212	<5	17.5	4.8
NSGWP13	71.5	1870	1.87	1.4	18.8	371	22.5	<2.0	256	<5	14.3	<2
NSGWP14	92.8	1420	1.42	1.3	20.3	363	19.1	9.9	315	<5	11.8	<2
NSGWP15	138	21900	21.9	2.9	29.4	1300	16.7	<2.0	466	<5	80.5	17.5
NSGWP16	131	7530	7.53	2	35.3	402	37.9	<2.0	472	<5	23.2	5.3
NSGWP17	109	22200	22.2	2.6	9.7	1050	7.8	<2.0	312	<5	43.5	13.8
NSGWP18	76.4	20500	20.5	3.1	4.5	578	21.6	<2.0	209	<5	22.4	12.4
NSGWP19	48.8	4560	4.56	2.2	11.6	209	6	3.6	170	<5	17	2.9
NSGWP20	86.8	2050	2.05	1	32	746	8.5	2.2	349	<5	13.2	<2
NSGWP21	84.6	2860	2.86	1.1	27.8	252	7.7	<2.0	326	<5	15.1	2.3
NSGWP22	126	10300	10.3	3.4	25.2	801	4.2	<2.0	418	<5	26.7	7.4
NSGWP23	110	12800	12.8	2	16.3	698	3.6	<2.0	342	<5	40.9	8.7
NSGWP24	106	3800	3.8	1.8	24	410	13.9	<2.0	364	<5	19.5	2.7
NSGWP25	110	5110	5.11	1.2	22.1	553	10	<2.0	366	<5	16.8	3.4
NSGWP26	145	2860	2.86	1.4	33.5	474	23.6	17.8	500	<5	24.7	2.1

Station_ID	V	SiO2 mg/L	NH3-N mg/L	CL mg/L	NO3-N mg/L	O-PHOS mg/L	T-PHOS mg/L	SO4 mg/L	TOC mg/L	TSS mg/L	TDS mg/L	HCO3 mg/L
NSGWAR01	<5	23.9	0.409	32.799	0.23	?<0.03	70.775	16.7	2.4	17.5	299	253.76
NSGWAR02	<5	28.6	0.308	23.386	0.217	?<0.03	70.592	34.5	2	9.5	286	209.84
NSGWAR03	<5	28.5	0.422	41.426	0.18	?<0.03	70.377	52.8	4.1	5.5	384	247.66
NSGWJF01	<5	18.7	0.249	21.152	0.478	?<0.03	70.346	15.7	2.1	2.5	320	298.9
NSGWJF02	<5	28	0.454	42.117	0.165	0.051	0.558	174	3	17	624	336.72
NSGWJF03	<5	20.8	0.186	58.061	0.149	?<0.03	0.315	32.2	2.4	7	405	303.78
NSGWJF04	<5	24	0.351	30.388	0.144	?<0.03	0.994	41.9	2.5	25	594	414.8
NSGWJF05	<5	29	0.249	184	0.269	0.085	0.426	83.7	2	2.5	697	362.34
NSGWJF07	<5	29.4	0.269	26.024	0.145	?<0.03	0.649	29.8	2.4	11	541	540.46
NSGWLN01	<5	27.7	0.219	29.276	0.139	?<0.03	0.69	25.6	2.3	9	339	284.26
NSGWLN02	<5	25.5	0.219	37.2	0.149	?<0.03	0.994	23.9	2.2	12.5	414	362.34
NSGWLN03	<5	23.2	0.747	31.472	0.141	?<0.03	0.588	4.9	3.1	32	462	474.58
NSGWLN06	<5	33.2	0.155	92.501	0.146	0.03	0.548	38.7	2.4	6	452	286.7
NSGWJF08	<5	37.4	0.457	11.943	0.082	70.041	0.905	1.9	2.2	12.5	195	164.7
NSGWJF09	<5	51	0.331	8.074	0.039	?<0.03	0.864	1.9	2.7	41.5	252	244
NSGWJF10	<5	25.1	0.43	14.949	0.046	?<0.03	0.682	7.5	2	13.5	368	383.08
NSGWJF11	<5	27.1	0.353	26.898	0.029	?<0.03	0.571	15.6	2.3	34	367	358.68
NSGWJF12	<5	25.1	0.239	9.183	<0.02	?<0.03	0.783	<1.0	2.9	24.5	444	498.98
NSGWJF13	<5	28.3	0.491	37.67	0.03	?<0.03	0.935	10.5	2.8	17.5	381	347.7
NSGWJF14	<5	35.6	0.981	27.471	0.059	0.07	0.571	7.5	4.1	30.5	251	202.52
NSGWJF15	<5	26.7	1.487	51.279	0.06	0.12	0.662	22.2	3.4	25.5	294	191.54
NSGWJF16	<5	24.4	0.249	11.272	0.034	?<0.03	0.53	<1.0	4.5	11.5	348	372.1
NSGWJF17	<5	30.3	0.328	25.293	0.037	?<0.03	0.672	27.1	1.9	7	302	239.12
NSGWJF18	<5	24.9	0.631	78.395	0.031	?<0.03	0.783	17.9	3.5	36	460	372.1
NSGWLN04	<5	27.9	0.631	43.35	0.042	70.076	71.09	6.9	4.6	29	436	424.56
NSGWLN05	<5	26.7	0.534	133	0.048	70.050	70.488	80.7	2.9	21	652	359.9
NSGWLN06	<5	23.2	0.491	19.138	0.047	70.043	70.763	9.2	3.1	23.5	464	533.14
NSGWDE01	<5	26.4	0.283	64.653	0.044	?<0.03	70.468	30.6	2.2	2	494	398.94
NSGWDE02	<5	29.4	0.687	126	0.029	?<0.03	70.834	145.2	2.3	29	690	380.64
NSGWDE03	<5	31.2	0.75	58.77	0.028	?<0.03	70.763	67.1	2.4	15	578	451.4
NSGWDE04	<5	33.8	0.708	94.045	0.032	70.030	70.580	46.8	6.1	22.5	566	422.12
NSGWDE05	<5	38.5	0.566	111.432	<0.02	?<0.03	70.814	87.3	3.5	29	663	408.7
NSGWDE06	<5	23.6	0.479	181	0.022	?<0.03	70.438	87.3	2.1	19.5	707	374.54
NSGWDE07	<5	26	0.646	68.931	0.048	?<0.03	70.885	54.1	2.3	33.5	448	301.34
NSGWLN07	<5	37.4	0.476	12.153	0.063	0.047	1.019	3.3	28.5	211	179.34	
NSGWDE08	<5	31.2	0.255	10.743	0.059	?<0.03	1.019	4.9	6.4	35.5	207	184.22
NSGWDE09	<5	26	0.643	30.693	0.052	?<0.03	0.949	9.1	2.8	30	383	366
NSGWDE10	<5	30.8	0.379	29.52	0.055	?<0.03	0.398	26.5	2.1	13	288	213.5
NSGWDE11	<5	26.4	0.492	87.163	0.052	?<0.03	0.307	30.9	2.4	19	439	279.38

Station_ID	V	SiO2 mg/L	NH3-N mg/L	CL mg/L	NO3-N mg/L	O-PHOS mg/L	T-PHOS mg/L	SO4 mg/L	TOC mg/L	TSS mg/L	TDS mg/L	HCO3 mg/L
	ug/L											
NSGWDE12	<5	31.1	0.574	20.366	0.054	<0.03	0.889	43.5	2.2	14.5	331	256.2
NSGWDE13	<5	28.8	0.696	28.343	0.049	<0.03	0.688	16	2.8	24	348	320.86
NSGWDE14	<5	29.5	0.314	73.789	0.048	<0.03	0.688	38.5	2.2	39	452	315.98
NSGWDE15	<5	30.8	0.792	21.113	0.043	0.059	0.715	37.7	NA	37.5	399	366
NSGWDR01	<5	31.2	0.392	12.143	0.043	0.095	1.101	10.8	NA	16.5	184	151.28
NSGWFJ19	<5	21.7	0.303	31.375	0.164	<0.03	0.5552	31	NA	9	387	334.28
NSGWDE16	<5	30.1	0.36	30.628	0.036	<0.03	0.532	17.4	NA	32	333	287.92
NSGWDE17	<5	29.5	0.402	30.96	0.039	<0.03	0.176	22.3	NA	6.5	296	224.48
NSGWDE18	<5	26.4	0.515	9.221	0.036	<0.03	0.43	2.1	NA	7	265	273.28
NSGWDE19	<5	36.7	0.538	81.607	0.034	<0.03	0.633	23.2	NA	21	512	378.2
NSGWDE20	<5	36	0.467	78.386	0.025	<0.03	0.45	29.3	NA	8.5	532	392.84
NSGWDE21	<5	41.1	0.351	13.915	0.032	0.05	0.745	27.6	NA	6	375	318.42
NSGWPFH01	<5	38.8	0.33	6.091	0.06	0.047	0.309	14.4	1.6	4.5	338	347.7
NSGWPFH02	<5	34	0.296	13.282	0.056	0.067	0.239	22	1.4	8	319	280.6
NSGWPFH03	<5	32	0.397	17.162	0.098	0.033	0.418	6.6	1.8	29.5	341	356.24
NSGWPFH04	<5	34.6	0.219	3.025	0.051	<0.03	0.745	7.8	2.6	27.5	404	458.72
NSGWPFH05	<5	37.9	0.724	21.228	0.05	0.064	0.428	66.5	2.5	4.5	455	375.76
NSGWPFH06	<5	36.1	0.522	5.907	0.079	0.078	0.309	20.1	2	9	345	366
NSGWPFH07	<5	35.4	0.606	8.649	0.068	0.046	0.249	73.5	2.5	22.5	479	309.88
NSGWPFH08	<5	37.2	0.269	13.44	0.304	0.037	0.199	22.9	2.7	5	376	378.2
NSGWPFH09	<5	32.6	1.014	4.828	0.05	<0.03	0.835	101.2	2.3	11.5	575	566.08
NSGWPFH10	<5	38.62	0.273	6.239	0.024	0.045	0.586	18.1	2.9	15	379	392.84
NSGWPFH11	<5	34.57	0.158	13.105	<0.02	0.137	0.344	112.4	1.6	16	544	431.88
NSGWPFH12	<5	34.61	0.09	6.198	0.025	<0.03	0.435	13.6	1.7	7.5	285	295.24
NSGWPFH13	<5	36.58	0.078	6.654	0.023	<0.03	0.223	51.9	1.3	2.5	358	305
NSGWPFH14	<5	33.31	<0.05	8.574	0.02	<0.03	0.192	79.2	1.1	1.5	432	344.04
NSGWPFH15	<5	35.36	1.256	4.239	0.031	<0.03	1.05	5.8	5	24	627	749.08
NSGWPFH16	<5	35.43	0.165	25.682	<0.02	0.047	0.142	135.2	1.6	16.5	669	507.52
NSGWPFH17	<5	36.42	1.271	4.928	0.035	<0.03	1.212	5.8	3.5	43.5	424	490.44
NSGWPFH18	<5	42.59	0.506	8.194	0.023	<0.03	0.949	7.2	3.4	28	361	392.84
NSGWPFH19	<5	41.37	0.267	2.673	0.025	0.082	0.687	9.9	1.8	5.5	227	236.68
NSGWPFH20	<5	28.02	0.337	5.298	0.045	0.048	0.323	8	1.2	15	383	436.76
NSGWPFH21	<5	33.69	0.264	4.862	0.034	0.042	0.353	5.2	1.1	28	359	405.04
NSGWPFH22	<5	31.16	0.382	8.719	0.175	<0.03	0.829	11.7	2.1	27	471	531.92
NSGWPFH23	<5	35.93	0.383	4.062	0.026	<0.03	1.224	2.1	2.4	34	395	458.72
NSGWPFH24	<5	35.79	0.174	3.6	0.028	<0.03	0.191	12.9	1.7	7	414	458.72
NSGWPFH25	<5	37.41	0.069	3.261	0.034	<0.03	0.171	58.7	1.3	14	423	400.16
NSGWPFH26	<5	33.45	0.093	4.607	<0.02	<0.03	0.13	56	2.1	10.5	576	595.36

Station_ID	DCAA-Surr.	3-5-Dichlorobenzoic_Acid ug/L	Dicamba ug/L	Dichlorprop ug/L	2,4-D ug/L	Pentachlorophenol ug/L	Silvex ug/L	2,4-5-T ug/L	Dinoseb ug/L	2,4-DB ug/L	Bentazon ug/L	Picloram ug/L	Acifluorfen ug/L
	% recovery												
NSGWAR01	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
NSGWAR02	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
NSGWAR03	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
NSGWJF01	NA	<.06344	<.02733	<.04198	<.03322	<.05123	<.03553	<.04190	<.07826	<.04042	<.04057	<.07155	<.11140
NSGWJF02	75.572	<.03095	<.02360	<.04376	<.01863	<.06733	<.02466	<.02765	<.10242	<.03849	<.05668	<.04966	<.13955
NSGWJF03	69.592	<.01924	<.02130	<.02615	<.01284	<.01976	<.01749	<.03562	<.07256	<.02356	<.03757	<.03522	<.07621
NSGWJF04	54.593	<.02381	<.03241	<.03560	<.03935	<.03633	<.03548	<.02836	<.09476	<.05306	<.03464	<.07146	<.11963
NSGWJF05	69.204	<.02040	<.02350	<.02896	<.01757	<.02864	<.03087	<.03498	<.04860	<.01597	<.01676	<.06217	<.09287
NSGWJF07	54.301	<.02191	<.02671	<.06283	<.04497	<.06701	<.04927	<.04453	<.09731	<.04431	<.01704	<.09922	<.12389
NSGWLN01	81.01	<.01095	<.02114	<.04296	<.00964	<.03705	<.02999	<.02278	<.05871	<.02049	<.01532	<.06040	<.06589
NSGWLN02	70.933	<.06084	<.02867	<.06082	<.03792	<.03226	<.01632	<.03702	<.09730	<.04135	<.03331	<.03286	<.11849
NSGWLN03	75.603	<.01308	<.01909	<.02510	<.02131	<.02399	<.04337	<.02162	<.06522	<.02509	<.03048	<.08734	<.05549
NSGWFJ06	68.581	<.07448	<.04405	<.05502	<.03183	<.06640	<.04409	<.03110	<.05904	<.05437	<.02219	<.05428	<.12777
NSGWFJ08	112.8	<.07715	<.06497	<.14500	<.11597	<.10908	<.08462	<.08413	<.14046	<.09299	<.04149	<.10418	<.24927
NSGWFJ09	114.59	<.05113	<.05947	<.16395	<.04224	<.08318	<.05467	<.04222	<.11008	<.06977	<.02836	<.06730	<.10006
NSGWFJ10	109.44	<.05209	<.03576	<.05633	<.02967	<.08423	<.06668	<.05183	<.09061	<.03008	<.02014	<.08209	<.12468
NSGWFJ11	116.46	<.06261	<.04720	<.11705	<.07346	<.08272	<.03594	<.05331	<.07984	<.09795	<.02933	<.04424	<.17479
NSGWFJ12	103.68	<.07292	<.02597	<.10493	<.06562	<.05603	<.06837	<.03552	<.09028	<.04726	<.04233	<.08417	<.17811
NSGWFJ13	94.346	<.09749	<.07433	<.10068	<.09535	<.07108	<.04850	<.04850	<.19920	<.11275	<.03406	<.08751	<.13106
NSGWFJ14	91.107	<.10312	<.07345	<.10926	<.03664	<.05278	<.03372	<.04714	<.21285	<.05865	<.02165	<.04152	<.11699
NSGWFJ15	89.319	<.11504	<.06141	<.24325	<.08277	<.15678	<.08592	<.08811	<.17654	<.10445	<.06385	<.10577	<.23735
NSGWFJ16	102.34	<.07672	<.09738	<.18561	<.12923	<.21861	<.08380	<.08185	<.22071	<.05143	<.03816	<.10316	<.15774
NSGWFJ17	100.11	<.08947	<.08155	<.28376	<.03923	<.11564	<.07846	<.09458	<.21584	<.10144	<.09034	<.09660	<.34320
NSGWFJ18	95.057	<.04474	<.02576	<.04024	<.06329	<.03769	<.04201	<.03165	<.05626	<.02580	<.02985	<.03453	<.07645
NSGWLN04	85.768	<.09242	<.06905	<.07183	<.03736	<.07721	<.04119	<.03659	<.10364	<.04878	<.02817	<.03386	<.39373
NSGWLN05	92.578	<.02980	<.04233	<.02970	<.03410	<.08888	<.02834	<.05379	<.07112	<.03250	<.01764	<.02329	<.06316
NSGWLN06	91.813	<.02089	<.01624	<.03579	<.02455	<.03693	<.03681	<.02025	<.05291	<.02864	<.01200	<.03025	<.03342
NSGWDE01	93.394	<.05566	<.03547	<.08199	<.02314	<.03717	<.02274	<.03267	<.03134	<.02333	<.08193	<.01749	<.03424
NSGWDE02	90.95	<.04102	<.02865	<.03567	<.05704	<.02874	<.02853	<.03943	<.03075	<.08029	<.01783	<.03484	<.03241
NSGWDE03	101.17	<.03641	<.08478	<.08835	<.13278	<.06284	<.13225	<.04667	<.04891	<.21522	<.02641	<.05427	<.08835
NSGWDE04	94.692	<.02170	<.04361	<.02733	<.03698	<.03096	<.03541	<.03833	<.04345	<.05652	<.02121	<.06582	<.03151
NSGWDE05	104.03	<.01407	<.05404	<.03303	<.02314	<.03717	<.02274	<.04829	<.06965	<.04128	<.05057	<.02221	<.01481
NSGWDE06	97.221	<.03797	<.04302	<.02886	<.01824	<.04856	<.02870	<.02427	<.03176	<.03855	<.10517	<.01711	<.07643
NSGWDE07	101.72	<.02988	<.01755	<.04398	<.02441	<.05702	<.023	<.02494	<.05431	<.01958	<.01905	<.02013	<.14919
NSGWLN07	106.17	<.09458	<.09857	<.13667	<.12235	<.06191	<.10743	<.06118	<.10622	<.02588	<.03648	<.08626	<.13706
NSGWDE08	113.32	<.05404	<.03563	<.06004	<.07746	<.06083	<.05189	<.04979	<.07512	<.02221	<.03193	<.04166	<.25463
NSGWDE09	99.04	<.03797	<.02886	<.01824	<.04856	<.04302	<.03654	<.02886	<.14137	<.01277	<.02119	<.02934	<.12250
NSGWDE10	98.867	<.02988	<.01755	<.04398	<.02441	<.05702	<.023	<.02494	<.05431	<.01958	<.01905	<.02013	<.14919
NSGWDE11	107.86	<.01407	<.02995	<.03066	<.01110	<.03204	<.028485	<.01957	<.03204	<.01030	<.02051	<.01056	<.08258

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Station_ID	DCAA-Surr. % recovery	3-5-Dichlorobenzoic_Acid ug/L	Dicamba ug/L	Dichlorprop ug/L	2,4-D ug/L	Pentachlorophenol ug/L	Silvex ug/L	2,4,5-T ug/L	Dinoseb ug/L	2,4-DB ug/L	Bentazon ug/L	Picloram ug/L	Aclifluorfen ug/L
NSGWDE12	86.787	<.02909	<.02799	<.03745	<.03484	<.05466	<.04401	<.03181	<.06098	<.02947	<.01075	<.03534	<.14855
NSGWDE13	92.667	<.01885	<.03266	<.02453	<.05005	<.02523	<.02890	<.05476	<.08952	<.02527	<.01312	<.02320	<.10921
NSGWDE14	87.597	<.01276	<.02119	<.02485	<.06043	<.04309	<.03226	<.02173	<.07866	<.00923	<.01030	<.02590	<.10928
NSGWDE15	97.455	<.05334	<.07137	<.06240	<.05685	<.06240	<.08071	<.07046	<.05027	<.04871	<.06258	<.02633	<.04986
NSGWDR01	90.643	<.04921	<.06326	<.04716	<.06588	<.07348	<.03375	<.03596	<.07929	<.05176	<.01935	<.02388	<.08280
NSGMJF19	103.14	<.03598	<.02402	<.07416	<.04362	<.07915	<.05579	<.02629	<.06516	<.04444	<.011531	<.05948	<.06906
NSGWDE16	95.879	<.04272	<.04695	<.10170	<.05928	<.09426	<.06721	<.03370	<.05987	<.03282	<.02499	<.04756	<.06501
NSGWDE17	92.345	<.16016	<.16678	<.23701	<.11036	<.32578	<.45828	<.22892	<.33289	<.32814	<.11163	<.32431	<.19145
NSGWDE18	89.77	<.03755	<.04635	<.08036	<.07698	<.08703	<.05130	<.04651	<.12014	<.04393	<.04360	<.03630	<.06123
NSGWDE19	88.769	<.06640	<.02998	<.09476	<.04032	<.08446	<.07583	<.04546	<.06666	<.04022	<.01140	<.03366	<.03151
NSGWDE20	86.452	<.04876	<.04183	<.04742	<.03055	<.05575	<.04194	<.03168	<.06556	<.03020	<.02661	<.02968	<.05215
NSGWDE21	89.925	<.05678	<.05520	<.09830	<.04454	<.05971	<.03848	<.05097	<.06384	<.05085	<.08748	<.02723	<.06393
NSGWP01	101.13	<.01853	<.00986	<.01816	<.01927	<.02092	<.01807	<.01175	<.02498	<.00657	<.00849	<.01555	<.06514
NSGWP02	100.99	<.01723	<.01895	<.03211	<.03338	<.02255	<.01899	<.01915	<.04636	<.01606	<.01568	<.01634	<.10532
NSGWP03	96.407	<.01203	<.02319	<.03573	<.02464	<.02172	<.02508	<.01550	<.02941	<.01172	<.00979	<.02158	<.10299
NSGWP04	110.87	<.02876	<.01758	<.03102	<.02235	<.04850	<.03197	<.02160	<.07593	<.02264	<.04071	<.02751	<.14545
NSGWP05	106.69	<.01496	<.01468	<.03154	<.02649	<.02506	<.02981	<.01420	<.04288	<.01349	<.01245	<.02565	<.12509
NSGWP06	99.044	<.03480	<.01658	<.02785	<.01155	<.02547	<.01784	<.02785	<.04758	<.01417	<.01587	<.01535	<.07446
NSGWP07	100.73	<.02920	<.01871	<.02207	<.03025	<.03231	<.01764	<.02111	<.05834	<.01390	<.01821	<.01518	<.05869
NSGWP08	97.934	<.01934	<.02249	<.05555	<.02358	<.05610	<.02188	<.02918	<.06931	<.02752	<.05019	<.01883	<.18624
NSGWP09	100.71	<.02809	<.01537	<.03020	<.03268	<.03216	<.02040	<.01910	<.02999	<.02159	<.04735	<.01755	<.10222
NSGWP10	89.974	<.02044	<.01588	<.02104	<.01592	<.01852	<.01710	<.01277	<.04430	<.01414	<.01435	<.01356	<.05352
NSGWP11	90.308	<.01881	<.01691	<.02143	<.02942	<.04395	<.02919	<.01590	<.03347	<.03560	<.01696	<.02315	<.05648
NSGWP12	94.76	<.02148	<.01689	<.03788	<.02507	<.03092	<.02809	<.01706	<.02975	<.01556	<.01123	<.02227	<.06358
NSGWP13	95.812	<.01108	<.01239	<.00835	<.01408	<.01730	<.02318	<.02044	<.02840	<.01125	<.00697	<.01838	<.05129
NSGWP14	83.633	<.01855	<.01054	<.02701	<.01939	<.03430	<.02455	<.01649	<.02546	<.02287	<.01342	<.01947	<.05643
NSGWP15	98.147	<.05021	<.02715	<.06463	<.03671	<.05693	<.03365	<.02365	<.05168	<.05283	<.02145	<.01745	<.02669
NSGWP16	81.212	<.01823	<.02419	<.04347	<.02918	<.06204	<.02385	<.03148	<.05891	<.01781	<.01395	<.01892	<.06793
NSGWP17	87.014	<.04048	<.04232	<.04273	<.03053	<.04009	<.02691	<.03753	<.06765	<.02736	<.01791	<.02134	<.10037
NSGWP18	90.234	<.02899	<.03697	<.03918	<.02838	<.05692	<.04537	<.01885	<.05312	<.01957	<.01330	<.03598	<.06555
NSGWP19	90.147	<.02894	<.04203	<.05640	<.02705	<.03867	<.04532	<.02688	<.05678	<.01777	<.01503	<.03594	<.06417
NSGWP20	85.831	<.02175	<.02732	<.04099	<.05913	<.02668	<.03343	<.02444	<.03682	<.03478	<.01416	<.02387	<.01854
NSGWP21	85.859	<.02764	<.01189	<.01906	<.04518	<.03315	<.02005	<.02234	<.03355	<.04177	<.02148	<.01432	<.05026
NSGWP22	90.983	<.03582	<.01248	<.05244	<.04601	<.02875	<.03745	<.02053	<.02500	<.03098	<.05375	<.02675	<.05620
NSGWP23	79.895	<.02870	<.03950	<.06082	<.04753	<.03048	<.03094	<.03191	<.03751	<.04374	<.07965	<.02209	<.07486
NSGWP24	93.503	<.05960	<.01833	<.03586	<.04830	<.03609	<.04351	<.04000	<.03815	<.02534	<.01783	<.03107	<.03232
NSGWP25	80.745	<.02929	<.01655	<.06690	<.04924	<.06118	<.02044	<.03862	<.02661	<.03854	<.01186	<.01460	<.04253
NSGWP26	74.511	<.03358	<.04188	<.03325	<.04194	<.02479	<.02479	<.01979	<.02152	<.01770	<.06943		

Station_ID	Nitrobenzene-d5	2-Fluorobiphenyl	Terphenyl-d14	Molinate ug/L	Propachlor ug/L	Trifluralin ug/L	Alpha-BHC ug/L	Atraton ug/L	Prometon ug/L	Simazine ug/L	Atrazine ug/L	Propazine ug/L	Beta-BHC ug/L
NSGWAR01	92.702	91.141	98.825	<.00540	<.00573	<.00330	<.02026	<.00778	<.01098	<.00589	<.00747	<.02439	
NSGWAR02	92.725	87.222	98.369	0.03484	<.00712	<.00440	<.02600	<.00593	<.01691	<.02199	<.00576	<.00903	<.03129
NSGWAR03	106.47	93.759	103.87	<.00493	<.00619	<.00239	<.01499	<.00775	<.00662	<.02530	<.00843	<.00561	<.01804
NSGMJF01	92.733	94.137	103.21	<.02813	<.01683	<.01901	<.04451	<.02294	<.02176	<.04539	<.01328	<.01676	<.05358
NSGMJF02	90.441	84.604	99.923	0.24612	<.01589	<.01137	<.03403	<.01763	<.01206	<.02234	<.00882	<.01383	<.04096
NSGMJF03	97.773	94.895	101.44	<.00814	<.00877	<.00949	<.02366	<.01362	<.00951	<.04463	<.00785	<.00693	<.02848
NSGMJF04	105.76	107.14	102.63	<.00265	<.00345	<.00378	<.00936	<.00796	<.00590	<.01173	<.00565	<.00318	<.01127
NSGMJF05	106.34	97.216	101.65	0.08647	<.00535	<.00391	<.01556	<.00631	<.00749	<.02083	<.00382	<.00262	<.01874
NSGMJF07	106.33	88.465	92.983	0.01023	<.00484	<.00318	<.01480	<.00642	<.00702	<.00832	<.00401	<.00422	<.01782
NSGVLN01	103.18	93.139	93.382	<.00459	<.00459	<.00251	<.00933	<.00877	<.00936	<.01654	<.00413	<.00410	<.01123
NSGVLN02	96.214	93.585	95.645	<.00507	<.00878	<.00365	<.01904	<.00595	<.00857	<.00774	<.00528	<.00256	<.02292
NSGVLN03	100.66	90.603	88.868	<.00397	<.00490	<.00289	<.00749	<.00719	<.01042	<.01362	<.00517	<.00254	<.00902
NSGMJF06	93.109	87.12	91.044	<.00427	<.00706	<.01137	<.01251	<.0164	<.01354	<.02826	<.00181	<.00567	<.01523
NSGMJF08	94.318	81.844	90.998	<.00721	<.00777	<.00393	<.02496	<.01292	<.00690	<.02821	<.00795	<.01084	<.03040
NSGMJF09	96.555	84.188	93.117	<.00414	<.01349	<.00483	<.01270	<.01112	<.01146	<.02870	<.00703	<.00658	<.01546
NSGMJF10	93.79	79.86	90.745	<.00434	<.00710	<.00329	<.01346	<.00436	<.01043	<.01613	<.00726	<.00691	<.01639
NSGMJF11	94.735	86.221	91.512	<.00593	<.00657	<.00157	<.01133	<.01403	<.01668	<.01744	<.00866	<.00863	<.01380
NSGMJF12	67.074	77.602	<.00799	<.00729	<.00287	<.01490	<.00966	<.01860	<.01146	<.02870	<.00411	<.00520	<.01815
NSGMJF13	65.065	82.757	<.00327	<.00955	<.00233	<.01410	<.01141	<.02115	<.02132	<.00918	<.00549	<.01717	
NSGMJF14	84.009	66.13	68.303	<.00563	<.00543	<.00241	<.01512	<.01431	<.01083	<.01423	<.00587	<.00584	<.01841
NSGMJF15	74.189	62.138	72.675	<.00435	<.00389	<.00261	<.01631	<.00929	<.01022	<.01520	<.00604	<.00417	<.01986
NSGMJF16	76.989	64.76	72.401	<.01194	<.01653	<.00479	<.01591	<.01345	<.01044	<.02840	<.01102	<.01285	<.01938
NSGMJF17	73.671	61.154	71.363	<.00873	<.00773	<.00885	<.02019	<.01296	<.02011	<.03759	<.01836	<.01009	<.02458
NSGMJF18	99.732	100.68	<.00560	<.01562	<.00492	<.02234	<.01558	<.01360	<.02412	<.00664	<.01268	<.02668	
NSGVLN04	100.83	99.829	101.84	<.02104	<.02146	<.01760	<.02657	<.02207	<.02120	<.06679	<.02263	<.01203	<.02926
NSGVLN05	105.48	103.33	104.92	<.00780	<.01060	<.00639	<.01006	<.00833	<.01428	<.01348	<.00736	<.00972	<.01107
NSGVLN06	104.41	100.35	104.3	<.01605	<.00625	<.00625	<.01506	<.01254	<.01592	<.02512	<.01141	<.00387	<.01659
NSGWDE01	109.81	102.7	100.86	0.09336	<.00649	<.00381	<.01127	<.01588	<.01901	<.01777	<.01420	<.00346	<.01241
NSGWDE02	105.01	99.875	97.848	<.00619	<.00683	<.00326	<.01952	<.01176	<.01935	<.02720	<.00459	<.00546	<.02150
NSGWDE03	104.07	103.33	104.92	<.00775	<.00307	<.00327	<.01321	<.01479	<.01486	<.01892	<.00853	<.00498	<.01455
NSGWDE04	99.4	90.038	98.649	<.00877	<.00664	<.00462	<.02161	<.00861	<.01944	<.02512	<.00851	<.00663	<.02379
NSGWDE05	100.16	104.49	97.835	<.00644	<.00797	<.00401	<.01238	<.00765	<.01576	<.01301	<.00700	<.00482	<.01363
NSGWDE06	107.1	96.538	95.661	0.17074	<.00585	<.00293	<.01726	<.00658	<.01262	<.02023	<.00206	<.00414	<.01901
NSGWDE07	103.47	97.623	98.233	0.01605	<.00382	<.00462	<.02161	<.00861	<.01944	<.02512	<.01141	<.00387	
NSGWLN07	99.841	99.729	105.59	<.01907	<.01687	<.00670	<.02739	<.01764	<.03240	<.05116	<.01215	<.00778	<.03433
NSGWDE08	106.15	105.88	112.27	<.01179	<.00988	<.00749	<.02166	<.00935	<.01521	<.02822	<.00823	<.01076	<.02715
NSGWDE09	108.46	106.19	107.48	<.00895	<.00766	<.00551	<.02419	<.01496	<.01552	<.03413	<.00774	<.01173	<.03032
NSGWDE10	115.5	99.166	109.6	<.00704	<.00837	<.00268	<.01909	<.00972	<.01261	<.01992	<.00579	<.00799	<.02393
NSGWDE11	120.2	97.813	109.78	<.00527	<.00704	<.00176	<.01337	<.01221	<.00993	<.01859	<.00563	<.00534	<.01676

Station_ID	Nitrobenzene-d5	2-Fluorobiphenyl	Terphenyl-d14	Molinate	Propachlor	Trifluralin	Alpha-BHC	Atraton	Prometon	Simazine	Atrazine	Propazine	Beta-BHC
	ug/L	% recovery	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L
NSGWDE12	108.62	108.09	106.02	<0.0495	<0.0567	<0.0415	<0.1729	<0.1575	<0.1281	<0.2565	<0.1055	<0.0702	<0.2167
NSGWDE13	106.43	104.11	98.866	<0.0797	<0.0918	<0.0487	<0.1171	<0.1169	<0.0978	<0.2324	<0.0777	<0.1253	<0.1468
NSGWDE14	110.66	98.99	106.31	<0.0451	<0.0727	<0.0457	<0.1042	<0.1082	<0.0895	<0.02095	<0.0649	<0.0240	<0.1306
NSGWDE15	96.985	97.011	97.264	<0.0330	<0.0684	<0.0099	<0.1595	<0.1265	<0.0586	<0.0989	<0.0549	<0.0396	<0.1918
NSGWDR01	95.478	95.877	98.391	<0.0341	<0.0495	<0.0295	<0.1064	<0.0682	<0.0688	<0.0651	<0.0798	<0.0520	<0.1280
NSGMJF19	84.321	92.685	95.369	0.03861	<0.0662	<0.0426	<0.1005	<0.1058	<0.0599	<0.0799	<0.0866	<0.0532	<0.1209
NSGWDE16	97.977	94.834	98.92	<0.0862	<0.0459	<0.0251	<0.1191	<0.1320	<0.1137	<0.0407	<0.0418	<0.0639	
NSGWDE17	95.107	94.063	94.632	<0.0476	<0.0553	<0.0194	<0.1056	<0.0362	<0.0431	<0.1486	<0.0876	<0.0344	<0.1270
NSGWDE18	86.156	106.6	95.865	<0.0789	<0.0758	<0.0480	<0.1245	<0.1167	<0.1114	<0.1972	<0.0691	<0.0588	<0.1498
NSGWDE19	100.34	102.15	97.973	<0.0659	<0.0646	<0.0315	<0.1349	<0.0420	<0.0920	<0.0686	<0.0619	<0.0374	<0.1623
NSGWDE20	91.638	107.69	104.59	<0.0617	<0.0949	<0.0233	<0.0740	<0.0840	<0.0535	<0.0706	<0.0435	<0.0541	<0.0890
NSGWDE21	90.586	99.193	96.15	<0.0396	<0.0559	<0.0168	<0.0796	<0.0769	<0.0572	<0.1314	<0.0283	<0.0317	<0.0958
NSGMPH01	109.69	104.86	100.42	<0.0449	<0.0221	<0.0081	<0.0567	<0.0407	<0.0752	<0.0826	<0.0348	<0.0247	<0.0650
NSGMPH02	92.717	90.915	88.405	<0.1524	<0.0945	<0.0183	<0.1636	<0.1743	<0.0234	<0.1969	<0.1467	<0.0695	<0.1876
NSGMPH03	105.76	91.502	94.542	<0.0265	<0.0338	<0.0178	<0.0759	<0.0438	<0.0700	<0.1232	<0.0469	<0.0557	<0.0871
NSGMPH04	94.268	89.124	92.087	<0.0324	<0.0440	<0.0250	<0.0683	<0.0737	<0.0680	<0.0997	<0.0270	<0.0263	<0.0784
NSGMPH05	97.138	91.538	93.183	<0.0283	<0.0181	<0.0211	<0.0851	<0.0564	<0.0281	<0.0601	<0.0174	<0.0457	<0.0976
51													
NSGMPH06	101.87	86.003	94.526	<0.0314	<0.0411	<0.0185	<0.0569	<0.0368	<0.0486	<0.0841	<0.0270	<0.0237	<0.0653
NSGMPH07	97.32	90.571	90.491	<0.0210	<0.0297	<0.0278	<0.0791	<0.0696	<0.0562	<0.0919	<0.0404	<0.0297	<0.0907
NSGMPH08	92.691	90.53	94.504	<0.0295	<0.0385	<0.0130	<0.0749	<0.0393	<0.0338	<0.0174	<0.0409	<0.0334	<0.0859
NSGMPH09	96.676	87.924	88.504	<0.0340	<0.0303	<0.0137	<0.0767	<0.0460	<0.0529	<0.0685	<0.0294	<0.0193	<0.0879
NSGMPH10	89.886	100.31	92.143	<0.0295	<0.0241	<0.0179	<0.0586	<0.0421	<0.0474	<0.0738	<0.0299	<0.0179	<0.0706
NSGMPH11	97.734	106.3	98.616	<0.0267	<0.0160	<0.0266	<0.0858	<0.0507	<0.0774	<0.01087	<0.0482	<0.0279	<0.1034
NSGMPH12	104.74	115.12	101.36	<0.0281	<0.0270	<0.0175	<0.1010	<0.0435	<0.0531	<0.1207	<0.0235	<0.0184	<0.1217
NSGMPH13	100.04	109.08	97.248	<0.0214	<0.0399	<0.0132	<0.0775	<0.0490	<0.0552	<0.0681	<0.0244	<0.0328	<0.0934
NSGMPH14	103.11	104.27	101.26	<0.0378	<0.0262	<0.0150	<0.0295	<0.0245	<0.0937	<0.0613	<0.0199	<0.0127	<0.0355
NSGMPH15	88.447	107.5	96.013	<0.0179	<0.0222	<0.0206	<0.0620	<0.0538	<0.0463	<0.1308	<0.0242	<0.0401	<0.0747
NSGMPH16	108.84	106.18	97.173	<0.0240	<0.0348	<0.0220	<0.0521	<0.0612	<0.0487	<0.0708	<0.0223	<0.0206	<0.0628
NSGMPH17	97.003	95.576	98.085	<0.0231	<0.0562	<0.0206	<0.0403	<0.0515	<0.0325	<0.1367	<0.0287	<0.0181	<0.0486
NSGMPH18	101.4	106.92	100.19	<0.0334	<0.0701	<0.0304	<0.0484	<0.0774	<0.0710	<0.1504	<0.0480	<0.0482	<0.0583
NSGMPH19	96.402	99.719	95.08	<0.0364	<0.0270	<0.0224	<0.0660	<0.0382	<0.0292	<0.0663	<0.0355	<0.0248	<0.0796
NSGMPH20	99.378	97.246	99.074	<0.0236	<0.0336	<0.0218	<0.0378	<0.0522	<0.0619	<0.0130	<0.0131	<0.0479	
NSGMPH21	99.507	90.594	96.488	<0.0264	<0.0268	<0.0167	<0.0690	<0.0356	<0.0353	<0.0308	<0.0370	<0.0227	<0.0874
NSGMPH22	96.344	94.122	95.514	<0.0274	<0.0224	<0.0148	<0.0366	<0.0317	<0.0505	<0.0830	<0.0198	<0.0300	<0.0463
NSGMPH23	102.05	95.25	101.41	<0.0155	<0.0301	<0.0126	<0.0247	<0.0525	<0.0654	<0.0673	<0.0182	<0.0272	<0.0313
NSGMPH24	100.56	98.438	102.66	<0.0360	<0.0384	<0.0249	<0.0421	<0.0630	<0.0418	<0.0880	<0.0393	<0.0272	<0.0534
NSGMPH25	92.59	89.61	98.365	<0.0185	<0.0379	<0.0225	<0.0554	<0.0393	<0.0597	<0.1020	<0.0266	<0.0231	<0.0703
NSGMPH26	107.61	91.475	96.509	<0.0263	<0.0301	<0.0232	<0.0728	<0.0407	<0.0351	<0.0686	<0.0240	<0.0240	<0.0923

Station_ID	Gamma-BHC	Bterbutylazine	Diazinon	Fluchloralin	Fonofos	Delta-BHC	Cyprazine	Metrubuzin	Methyl-Parathion	Atrachlor	Ametryn	Prometryn	Heptachlor	Terbutryn
	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L
NSGWAR01	<.02122	<.01139	<.01196	<.00571	<.00509	<.02574	<.00536	<.00714	<.00409	<.00837	<.00491	<.00333	<.00604	<.00467
NSGWAR02	<.02723	<.01016	<.02220	<.00778	<.00491	<.03303	<.00400	<.00741	<.00692	<.00750	<.00687	<.00688	<.00743	<.00844
NSGWAR03	<.01570	<.01617	<.01938	<.00406	<.00904	<.01905	<.00696	<.00669	<.00679	<.01020	<.00500	<.01296	<.00827	<.00422
NSGWJF01	<.04662	<.04637	<.03065	<.03217	<.02930	<.05655	<.01429	<.01953	<.02598	<.01323	<.02438	<.02446	<.01495	<.00924
NSGWJF02	<.03564	<.02664	<.02331	<.01838	<.02365	<.04323	<.01284	<.01645	<.01652	<.01701	<.01141	<.00941	<.01101	<.00789
NSGWJF03	<.02478	<.01369	<.02222	<.01550	<.01573	<.03006	<.00766	<.00767	<.01212	<.00628	<.00623	<.00734	<.01185	<.00800
NSGWJF04	<.00981	<.01324	<.01455	<.00619	<.00319	<.01190	<.00411	<.00608	<.00517	<.00859	<.00350	<.00674	<.00432	<.00192
NSGWJF05	<.01630	<.01051	<.00983	<.00645	<.00550	<.01978	<.00600	<.00398	<.00575	<.00345	<.00412	<.00306	<.00796	<.00646
NSGWJF07	<.01550	<.00779	<.01305	<.00541	<.00348	<.01880	<.00447	<.00596	<.00532	<.00578	<.00624	<.00570	<.00643	<.00401
NSGWLN01	<.00977	<.00862	<.01307	<.00438	<.00296	<.01185	<.00381	<.00684	<.00449	<.00532	<.00403	<.00420	<.00688	<.00699
NSGWLN02	<.01994	<.00790	<.01396	<.00650	<.00549	<.02419	<.00652	<.00719	<.00650	<.00940	<.00654	<.00774	<.01215	<.00360
NSGWLN03	<.00785	<.00869	<.01349	<.00490	<.00321	<.00952	<.00721	<.00543	<.00646	<.00725	<.00588	<.00646	<.00395	<.00148
NSGWJF06	<.01184	<.00615	<.01063	<.00482	<.00291	<.01437	<.00552	<.00547	<.00893	<.00537	<.00539	<.00509	<.00517	<.00510
NSGWJF08	<.01335	<.02183	<.01710	<.00605	<.00641	<.01541	<.00911	<.00773	<.00390	<.01082	<.00820	<.00769	<.00613	<.00549
NSGWJF09	<.02666	<.02594	<.03379	<.00729	<.00637	<.03076	<.00797	<.00594	<.01327	<.01001	<.01430	<.00971	<.01302	<.01162
NSGWJF10	<.01356	<.01784	<.01192	<.00658	<.01259	<.01565	<.00996	<.01212	<.01036	<.01234	<.00823	<.00929	<.01092	<.00769
NSGWJF11	<.01437	<.01770	<.01298	<.00578	<.00600	<.01659	<.00482	<.00704	<.00584	<.00862	<.00556	<.00527	<.01399	<.00422
NSGWJF12	<.01210	<.01710	<.02015	<.00309	<.00746	<.01397	<.00627	<.00870	<.00592	<.01162	<.00858	<.00753	<.01375	<.00561
NSGWJF13	<.01591	<.01271	<.01722	<.00541	<.00356	<.01836	<.00723	<.00531	<.00936	<.00907	<.00534	<.00985	<.01228	<.00828
NSGWJF14	<.01506	<.02227	<.01172	<.00429	<.00405	<.01737	<.00446	<.00635	<.00487	<.01090	<.00437	<.00314	<.00743	<.00503
NSGWJF15	<.01614	<.01390	<.01090	<.00438	<.00770	<.01863	<.00638	<.00466	<.00494	<.00807	<.00561	<.01346	<.00737	<.00502
NSGWJF16	<.01742	<.00801	<.01057	<.00488	<.00373	<.02010	<.00616	<.00871	<.00442	<.00935	<.00840	<.00629	<.00599	<.00513
NSGWJF17	<.01699	<.02047	<.02148	<.00904	<.02366	<.01961	<.00823	<.01028	<.01458	<.01086	<.01619	<.01778	<.01095	<.00878
NSGWJF18	<.02156	<.02426	<.01647	<.01691	<.01517	<.02488	<.00801	<.00878	<.01142	<.02060	<.01498	<.01849	<.01251	<.00835
NSGWLN04	<.02331	<.01451	<.01256	<.00901	<.00783	<.02863	<.00837	<.00613	<.00793	<.00742	<.00608	<.01059	<.00951	<.00743
NSGWLN05	<.02667	<.05148	<.03962	<.03150	<.02386	<.03229	<.01437	<.02078	<.02785	<.01669	<.01632	<.02034	<.03297	<.01157
NSGWLN06	<.01009	<.01131	<.01854	<.01102	<.01102	<.01222	<.01019	<.00656	<.00764	<.01062	<.00716	<.00890	<.01627	<.00667
NSGWDE01	<.01131	<.01703	<.01996	<.00694	<.00516	<.01369	<.00706	<.00916	<.00837	<.00721	<.00598	<.00498	<.01190	<.00781
NSGWDE02	<.01959	<.00907	<.01562	<.00567	<.00729	<.02372	<.00858	<.00778	<.00867	<.01103	<.00647	<.00604	<.00972	<.00828
NSGWDE03	<.01567	<.01663	<.00598	<.00348	<.01606	<.00714	<.00896	<.00983	<.00479	<.00567	<.00664	<.00793	<.00567	<.00525
NSGWDE04	<.01512	<.01321	<.01662	<.01156	<.00374	<.01830	<.00772	<.00439	<.00938	<.01424	<.00842	<.00692	<.01465	<.00664
NSGWDE05	<.01242	<.01680	<.01828	<.00584	<.00710	<.01504	<.01188	<.00439	<.01100	<.00599	<.00829	<.00760	<.01111	<.00472
NSGWDE06	<.01733	<.01327	<.01613	<.00884	<.00836	<.00664	<.02626	<.00811	<.00872	<.00551	<.00530	<.00712	<.00526	<.00885
NSGWDE07	<.02169	<.01715	<.00884	<.01320	<.00307	<.03496	<.01894	<.01630	<.01701	<.02263	<.01055	<.01918	<.03416	<.01403
NSGWLN07	<.02949	<.05633	<.03823	<.01390	<.01144	<.02764	<.01436	<.01648	<.01688	<.00993	<.01579	<.01493	<.02354	<.01129
NSGWDE08	<.02332	<.02757	<.02402	<.01390	<.01046	<.00610	<.03087	<.01015	<.01321	<.01114	<.01058	<.01088	<.00877	<.01235
NSGWDE09	<.02605	<.02034	<.02174	<.01617	<.00524	<.00981	<.02437	<.00637	<.00696	<.00704	<.00670	<.01077	<.00800	<.01453
NSGWDE10	<.02056	<.00816	<.02801	<.01340	<.00342	<.00983	<.01706	<.00840	<.00558	<.00971	<.00646	<.00748	<.00794	<.00279

Station_ID	Gamma-BHC	Terbutylazine	Diazinon	Fluchloralin	Fonofos	Delta-BHC	Cyprazine	Metrubizin	Methyl-Parathion	Aalachlor	Ametryn	Prometryn	Heptachlor	Terbutryn
	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L
NSGWDE12	<.01862	<.01673	<.01498	<.00788	<.00847	<.02207	<.00648	<.01142	<.00546	<.01764	<.01079	<.00678	<.01112	<.00443
NSGWDE13	<.01261	<.02076	<.02348	<.00897	<.00381	<.01494	<.00815	<.00788	<.00830	<.01078	<.00637	<.01728	<.01805	<.00950
NSGWDE14	<.01122	<.02521	<.02444	<.00924	<.00855	<.01330	<.00916	<.00853	<.01093	<.00970	<.00711	<.01226	<.01132	<.00863
NSGWDE15	<.01717	<.01712	<.00926	<.00174	<.00387	<.02076	<.00527	<.00547	<.00541	<.00608	<.00699	<.01092	<.00417	
NSGWDRO1	<.01145	<.00667	<.01314	<.00511	<.00176	<.01385	<.00346	<.00727	<.00614	<.00521	<.00610	<.00624	<.00972	<.00443
NSGWJF19	<.01082	<.01289	<.00914	<.00713	<.00773	<.01308	<.00617	<.00735	<.00502	<.00890	<.00589	<.00567	<.00483	
NSGWDE16	<.00572	<.01775	<.01587	<.00421	<.00592	<.00691	<.00595	<.00625	<.00265	<.00545	<.00636	<.00559	<.00884	<.00561
NSGWDE17	<.01137	<.01229	<.01326	<.00330	<.00672	<.01374	<.00458	<.00571	<.00832	<.00960	<.00518	<.00766	<.01132	
NSGWDE18	<.01340	<.01809	<.01438	<.00763	<.00734	<.01621	<.00764	<.01147	<.00740	<.00830	<.00816	<.01255	<.01499	<.00477
NSGWDE19	<.01453	<.01309	<.01171	<.00532	<.00656	<.01757	<.00534	<.00681	<.00726	<.00404	<.00725	<.00513	<.01178	<.00343
NSGWDE20	<.00797	<.01220	<.01294	<.00394	<.00424	<.00963	<.00492	<.00233	<.00564	<.00233	<.00523	<.00376	<.01267	<.00336
NSGWDE21	<.00857	<.01226	<.00890	<.00273	<.00626	<.01037	<.00388	<.00564	<.00377	<.00653	<.00454	<.00458	<.00878	<.00254
NSGWPH01	<.00584	<.00565	<.00833	<.00144	<.00339	<.00701	<.00463	<.00503	<.00356	<.00432	<.00340	<.00340	<.00521	<.00488
NSGWPH02	<.01687	<.02240	<.02793	<.00376	<.00722	<.02026	<.00915	<.01391	<.01089	<.00874	<.01326	<.00914	<.01711	<.00569
NSGWPH03	<.00783	<.00494	<.00511	<.00321	<.00496	<.00940	<.00462	<.00283	<.00279	<.00859	<.00566	<.00712	<.00639	<.00462
NSGWPH04	<.00705	<.00887	<.01131	<.00458	<.00325	<.00846	<.00754	<.00308	<.00360	<.00391	<.00414	<.00308	<.00521	<.00442
NSGWPH05	<.00878	<.00648	<.00708	<.00399	<.00236	<.01054	<.00275	<.00396	<.00548	<.00461	<.00365	<.00205	<.00407	<.00294
NSGWPH06	<.00587	<.00753	<.00516	<.00349	<.00269	<.00705	<.00318	<.00536	<.00246	<.00369	<.00308	<.00520	<.00422	<.00579
NSGWPH07	<.00815	<.00373	<.01308	<.00511	<.00611	<.00979	<.00640	<.00559	<.00625	<.00377	<.00380	<.01205	<.00409	<.00238
NSGWPH08	<.00772	<.00439	<.00720	<.00237	<.00345	<.00928	<.00335	<.00226	<.00395	<.00434	<.00193	<.00269	<.00473	<.00249
NSGWPH09	<.00790	<.00658	<.00939	<.00249	<.00160	<.00949	<.00328	<.00328	<.00371	<.00372	<.00290	<.00312	<.00506	<.00186
NSGWPH10	<.00642	<.00356	<.00466	<.00282	<.00179	<.00766	<.00246	<.00235	<.00460	<.00354	<.00421	<.07306	<.00472	<.00352
NSGWPH11	<.00940	<.00528	<.00788	<.00428	<.00236	<.01122	<.00426	<.00302	<.00558	<.00195	<.00182	<.00274	<.00417	<.00254
NSGWPH12	<.01107	<.00565	<.00788	<.00284	<.00258	<.01321	<.00665	<.00199	<.00376	<.00367	<.00331	<.00339	<.00340	
NSGWPH13	<.00850	<.00474	<.00838	<.00202	<.00188	<.01014	<.00248	<.00500	<.00291	<.00510	<.00460	<.00406	<.00481	
NSGWPH14	<.00323	<.00338	<.00473	<.00240	<.00192	<.00385	<.00308	<.00169	<.00404	<.00323	<.00388	<.00316	<.00418	<.00203
NSGWPH15	<.00630	<.00540	<.00807	<.00329	<.00278	<.00811	<.00249	<.00505	<.00214	<.00443	<.00308	<.00353	<.00549	<.00201
NSGWPH16	<.00571	<.00234	<.01163	<.00349	<.00220	<.00681	<.00321	<.00366	<.00132	<.00312	<.00309	<.00377	<.00760	<.00158
NSGWPH17	<.00442	<.00642	<.00785	<.00329	<.00417	<.00528	<.00381	<.00308	<.00329	<.00337	<.00260	<.00235	<.00667	<.00185
NSGWPH18	<.00530	<.00994	<.01140	<.00485	<.00209	<.00632	<.00435	<.00600	<.00617	<.00553	<.00474	<.00457	<.00758	<.00320
NSGWPH19	<.00724	<.00521	<.00671	<.00351	<.00251	<.00864	<.00248	<.00418	<.00269	<.00270	<.00265	<.00512	<.00636	<.00346
NSGWPH20	<.00405	<.00346	<.00355	<.00381	<.00240	<.00522	<.00329	<.00367	<.00496	<.00411	<.00177	<.00331	<.00454	<.00306
NSGWPH21	<.00739	<.00438	<.00588	<.00264	<.00270	<.00951	<.00245	<.00310	<.00247	<.00204	<.00244	<.00250	<.00221	<.00190
NSGWPH22	<.00392	<.00548	<.00823	<.00228	<.00172	<.00504	<.00417	<.00248	<.00346	<.00248	<.00265	<.00512	<.00636	<.00333
NSGWPH23	<.00265	<.00432	<.00649	<.00202	<.00182	<.00341	<.00362	<.00233	<.00486	<.00226	<.00187	<.00454	<.00262	
NSGWPH24	<.00451	<.00490	<.00684	<.00393	<.00259	<.00580	<.00228	<.00326	<.00583	<.00303	<.00439	<.00539	<.00319	
NSGWPH25	<.00594	<.00381	<.00468	<.00389	<.00205	<.00765	<.00384	<.00403	<.00455	<.00395	<.00284	<.00335	<.00433	<.00252
NSGWPH26	<.00731	<.00463	<.00721	<.00371	<.00192	<.00453	<.01004	<.00405	<.00386	<.00405	<.00192	<.00405	<.00374	<.00261

Station_ID	Metolachlor	Malathion	Dipropetyn	Chlorpyrifos	Cyanazine	Aldrin	Pendimethalin	Heptachlor-Epoxide	Endosulfan-I	p,p'-DDE	Dieldrin	Endrin	Endosulfan-II	p,p'-DDD
	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L
NSGWAR01	<.00305	<.00599	<.00630	<.01268	<.01190	<.01219	<.00508	<.00489	<.18287	<.00321	<.02112	<.04693	<.05499	<.00485
NSGWAR02	<.00223	<.00816	<.00910	<.01846	<.02292	<.02063	<.00595	<.00688	<.13808	<.00907	<.04375	<.08146	<.00487	<.00363
NSGWAR03	<.00542	<.01902	<.00950	<.02654	<.01649	<.02024	<.00914	<.01183	<.15731	<.00607	<.02525	<.09553	<.08418	<.00715
NSGWJF01	<.00735	<.09514	<.02052	<.03109	<.04113	<.07747	<.01410	<.02213	<.38494	<.01698	<.08235	<.11937	<.25307	<.00795
NSGWJF02	<.00983	<.05096	<.01477	<.01954	<.03153	<.04927	<.00809	<.01529	<.15694	<.01774	<.06730	<.07468	<.12282	<.00418
NSGWJF03	<.00293	<.02023	<.00839	<.00921	<.02787	<.03615	<.00370	<.01815	<.15855	<.00986	<.02802	<.06499	<.08312	<.00492
NSGWJF04	<.00258	<.00998	<.00564	<.01098	<.00811	<.01541	<.00339	<.01025	<.06723	<.00470	<.01789	<.05765	<.07021	<.00492
NSGWJF05	.00407	<.01143	<.00473	<.00982	<.01146	<.01714	<.00379	<.00479	<.09350	<.00186	<.03320	<.03513	<.09029	<.00497
NSGWJF07	<.00250	<.00797	<.00539	<.00613	<.00980	<.01588	<.00376	<.01258	<.12202	<.00273	<.02886	<.01820	<.08020	<.00201
NSGWLN01	<.00266	<.00696	<.00931	<.00733	<.02047	<.01340	<.00412	<.00580	<.12722	<.00376	<.01689	<.02960	<.04859	<.00399
NSGWLN02	<.00280	<.01081	<.00772	<.00904	<.00987	<.01044	<.00331	<.00970	<.06693	<.00181	<.01739	<.03193	<.03397	<.00245
NSGWLN03	<.00312	<.01077	<.00582	<.00813	<.01315	<.01925	<.00218	<.00937	<.08963	<.00430	<.02715	<.05798	<.04101	<.00465
NSGWJF06	<.00356	<.01248	<.00723	<.00825	<.00638	<.02664	<.00497	<.00719	<.07976	<.00428	<.02848	<.02716	<.04227	<.00434
NSGWJF08	.01055	<.00559	<.01038	<.01045	<.01711	<.01072	<.00654	<.00864	<.15635	<.00409	<.03012	<.05522	<.05005	<.00575
NSGWJF09	<.00577	<.01848	<.01458	<.01692	<.02569	<.03651	<.00679	<.01073	<.30333	<.00832	<.04302	<.08347	<.11403	<.00817
NSGWJF10	<.00506	<.01187	<.01185	<.01993	<.02675	<.02851	<.00964	<.01925	<.16209	<.00834	<.03387	<.06856	<.07964	<.00696
NSGWJF11	<.00375	<.01374	<.00876	<.01307	<.01027	<.01606	<.00699	<.01101	<.24406	<.00560	<.02235	<.03693	<.11828	<.00290
NSGWJF12	<.00323	<.00963	<.00704	<.01434	<.01057	<.01628	<.00502	<.01908	<.13450	<.00851	<.03964	<.04311	<.11315	<.00567
NSGWJF13	<.00440	<.01041	<.00802	<.01198	<.01639	<.02574	<.00390	<.01006	<.12630	<.00548	<.02779	<.03426	<12809	<.00559
NSGWJF14	<.00326	<.01405	<.01091	<.01014	<.01815	<.01341	<.00470	<.00449	<.14601	<.00389	<.02790	<.03323	<.08398	<.00356
NSGWJF15	<.00361	<.00719	<.00417	<.00836	<.01813	<.01359	<.00367	<.00452	<.16607	<.00514	<.02315	<.03292	<.06027	<.00405
NSGWJF16	<.00159	<.00672	<.00924	<.00734	<.01067	<.01216	<.00562	<.01309	<.08953	<.00302	<.03274	<.03499	<.05448	<.00521
NSGWJF17	<.00438	<.02704	<.00838	<.01948	<.02327	<.04010	<.00656	<.02839	<.23103	<.01277	<.07168	<.10312	<.13467	<.01064
NSGWJF18	<.00545	<.02977	<.00652	<.01110	<.02606	<.03141	<.00481	<.01643	<.17934	<.00733	<.04962	<.09027	<.05413	<.00856
NSGWLN04	<.00550	<.01014	<.00772	<.00935	<.01756	<.01981	<.00846	<.01664	<.10110	<.00499	<.02770	<.06371	<.07411	<.00336
NSGWLN05	<.00778	<.03725	<.01851	<.03248	<.02011	<.07763	<.00690	<.02593	<.26601	<.01495	<.08885	<.17806	<.13712	<.00657
NSGWLN06	<.00417	<.01032	<.00904	<.01569	<.02262	<.02129	<.00757	<.01104	<.27177	<.00824	<.03862	<.06450	<.09539	<.00643
NSGWDE01	<.00466	<.00799	<.00482	<.01103	<.02332	<.02378	<.00473	<.01093	<.21181	<.00468	<.02198	<.03760	<.06715	<.00513
NSGWDE02	<.00581	<.01042	<.00508	<.01759	<.03598	<.02417	<.00333	<.01042	<.24558	<.00328	<.02192	<.03942	<.12737	<.00649
NSGWDE03	<.00519	<.00969	<.00787	<.01285	<.02772	<.02740	<.00587	<.01166	<.18906	<.00694	<.01225	<.02332	<.10439	<.00561
NSGWDE04	<.00396	<.01094	<.00690	<.01712	<.01666	<.02614	<.00438	<.02154	<.24712	<.00753	<.03935	<.06843	<.08606	<.00447
NSGWDE05	<.00473	<.01576	<.01024	<.00755	<.02273	<.03067	<.00588	<.01202	<.31708	<.00298	<.02127	<.05914	<.11347	<.00440
NSGWDE06	<.00389	<.01093	<.00536	<.00672	<.01015	<.01537	<.00317	<.01173	<.17570	<.00464	<.04877	<.04280	<.08469	<.00438
NSGWDE07	<.00564	<.01308	<.00478	<.01497	<.02458	<.02396	<.00663	<.01147	<.13677	<.00819	<.02858	<.04452	<.10482	<.00315
NSGMLN07	<.00783	<.05723	<.01966	<.02804	<.03332	<.05306	<.01416	<.03286	<.28676	<.01311	<.13667	<.14970	<.09880	<.01274
NSGWDE08	<.00670	<.03247	<.01940	<.02163	<.03765	<.05266	<.00744	<.01875	<.23762	<.00986	<.07791	<.08143	<.17219	<.00851
NSGWDE09	<.00394	<.01452	<.00490	<.01294	<.01898	<.04122	<.00411	<.01796	<.19833	<.00437	<.05900	<.02239	<.13736	<.00682
NSGWDE10	<.00397	<.02061	<.00684	<.00908	<.02195	<.00540	<.01079	<.22059	<.00338	<.05486	<.07269	<.07925	<.00479	<.00499
NSGWDE11	<.00561	<.00347	<.00364	<.01474	<.00853	<.03029	<.00663	<.02176	<.25293	<.00611	<.02675	<.03470	<.10684	<.00499

Station_ID	Metolachlor	Malathion	Dipropetryn	Chlorpyrifos	Cyanazine	Aldrin	Pendimethalin	Heptachlor-Epoxide	Endosulfan-I	p,p'-DDE	Dieldrin	Endrin	Endosulfan-II	p,p'-DDD
	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L
NSGWDE12	<.00371	<.01022	<.00958	<.01672	<.01736	<.01704	<.00645	<.01001	<.11022	<.01036	<.03973	<.05828	<.07599	<.00480
NSGWDE13	<.00545	<.01854	<.00941	<.01336	<.01278	<.02590	<.00957	<.00896	<.24702	<.00820	<.05932	<.03707	<.14668	<.00344
NSGWDE14	<.00715	<.01229	<.01011	<.01627	<.01491	<.00724	<.01142	<.29203	<.00579	<.02891	<.04055	<.10065	<.00685	<.00328
NSGWDE15	<.00203	<.01256	<.00896	<.00925	<.01676	<.01364	<.00458	<.00489	<.09966	<.00402	<.02277	<.03728	<.06351	<.00346
NSGWDR01	<.00179	<.00583	<.00554	<.01269	<.01680	<.01547	<.00497	<.00499	<.11093	<.00552	<.03390	<.02920	<.08328	<.00446
NSGWJF19	<.00284	<.01302	<.00890	<.01477	<.00992	<.01266	<.00607	<.01087	<.16005	<.00606	<.02449	<.04681	<.09987	<.00446
NSGWDE16	<.00366	<.00950	<.00664	<.01089	<.00749	<.00669	<.00525	<.00689	<.07843	<.00374	<.02996	<.0192	<.08131	<.00381
NSGWDE17	<.00318	<.00520	<.00674	<.00673	<.00548	<.02097	<.00551	<.00838	<.17428	<.00584	<.04455	<.03359	<.05407	<.00302
NSGWDE18	<.00451	<.01229	<.00865	<.01321	<.02143	<.01865	<.00925	<.01318	<.10031	<.00806	<.03467	<.10871	<.04272	<.00518
NSGWDE19	<.00289	<.01090	<.00757	<.01527	<.01236	<.01828	<.00420	<.01213	<.10483	<.00564	<.02249	<.01883	<.08576	<.00386
NSGWDE20	<.00334	<.01091	<.00614	<.00559	<.01123	<.01422	<.00662	<.00832	<.10473	<.00298	<.03208	<.03404	<.10258	<.00250
NSGWDE21	0.01808	<.00562	<.00778	<.00440	<.00895	<.00950	<.00321	<.00572	<.10004	<.00657	<.01682	<.02227	<.06194	<.00178
NSGMPH01	<.00181	<.00702	<.00505	<.00509	<.00886	<.00894	<.00588	<.00402	<.14951	<.00301	<.01792	<.01733	<.05086	<.00365
NSGMPH02	<.00870	<.02162	<.01561	<.01690	<.03300	<.02732	<.01567	<.00941	<.25632	<.01105	<.05446	<.09275	<.10586	<.00401
NSGMPH03	<.00233	<.01184	<.00511	<.00423	<.00663	<.00700	<.00527	<.00837	<.06484	<.00295	<.01726	<.02145	<.06374	<.00165
NSGMPH04	<.00221	<.00893	<.00346	<.00584	<.01176	<.00904	<.00406	<.00510	<.10938	<.00369	<.02405	<.02513	<.02993	<.00239
NSGMPH05	<.00292	<.00645	<.00357	<.00789	<.01082	<.01376	<.00405	<.00386	<.07632	<.00541	<.01675	<.02260	<.05300	<.00136
NSGMPH06	<.00221	<.00552	<.00444	<.00741	<.00614	<.00618	<.00336	<.00546	<.07625	<.00176	<.02673	<.02452	<.03423	<.00225
NSGMPH07	<.00156	<.00537	<.00429	<.00422	<.00557	<.01568	<.00473	<.00474	<.06868	<.00403	<.01194	<.00939	<.05295	<.00237
NSGMPH08	<.00163	<.00637	<.00482	<.00302	<.00969	<.00992	<.00203	<.00528	<.11357	<.00364	<.02316	<.02839	<.04073	<.00360
NSGMPH09	<.00120	<.00409	<.00483	<.00709	<.00589	<.00931	<.00171	<.00713	<.07514	<.00262	<.01703	<.03111	<.02776	<.00230
NSGMPH10	<.00198	<.00393	<.00278	<.00664	<.00390	<.01131	<.00245	<.00300	<.10210	<.00275	<.02122	<.01689	<.03588	<.00177
NSGMPH11	<.00149	<.00647	<.00365	<.00460	<.00638	<.01372	<.00240	<.00312	<.06878	<.00118	<.00960	<.00614	<.03523	<.00334
NSGMPH12	<.00083	<.00815	<.00364	<.00418	<.00803	<.00925	<.00472	<.00567	<.07713	<.00297	<.02693	<.02797	<.05236	<.00207
NSGMPH13	<.00092	<.00935	<.00465	<.00344	<.00357	<.00714	<.00303	<.00389	<.07874	<.00245	<.02269	<.01096	<.03320	<.00180
NSGMPH14	<.00088	<.00218	<.00241	<.00554	<.00525	<.00991	<.00431	<.00259	<.08280	<.00179	<.01547	<.00842	<.04620	<.00196
NSGMPH15	<.00257	<.00697	<.00282	<.00498	<.00819	<.00525	<.00438	<.00608	<.09321	<.00230	<.02721	<.01583	<.03732	<.00246
NSGMPH16	<.00137	<.00698	<.00320	<.00368	<.00706	<.00325	<.00549	<.00269	<.12908	<.00280	<.01470	<.01436	<.03231	<.00214
NSGMPH17	<.00160	<.00613	<.00245	<.00348	<.00394	<.00809	<.00335	<.00472	<.07119	<.00393	<.01438	<.01186	<.03990	<.00153
NSGMPH18	<.00166	<.01168	<.00592	<.00481	<.01052	<.01515	<.00182	<.00420	<.07436	<.00385	<.02469	<.02178	<.06532	<.00161
NSGMPH19	<.00111	<.00530	<.00483	<.00398	<.01053	<.00661	<.00311	<.00324	<.11170	<.00210	<.02550	<.01071	<.03178	<.00187
NSGMPH20	<.00198	<.00515	<.00245	<.00299	<.01015	<.00498	<.00346	<.00354	<.09354	<.00177	<.02577	<.01285	<.03504	<.00266
NSGMPH21	<.00270	<.00469	<.00360	<.00561	<.00643	<.00504	<.00212	<.00172	<.07628	<.00215	<.02273	<.01066	<.02961	<.00170
NSGMPH22	<.00161	<.00601	<.00411	<.00401	<.00625	<.00708	<.00229	<.00334	<.12114	<.00228	<.01986	<.01265	<.02529	<.00204
NSGMPH23	<.00204	<.00399	<.00339	<.00320	<.01155	<.00994	<.00581	<.00296	<.11726	<.00193	<.00992	<.02115	<.02819	<.00149
NSGMPH24	<.00162	<.00358	<.00508	<.00462	<.00762	<.01192	<.00422	<.00385	<.10826	<.00307	<.01797	<.01071	<.02408	<.00181
NSGMPH25	<.00103	<.00424	<.00363	<.00555	<.01004	<.00930	<.00443	<.00618	<.09256	<.00303	<.01561	<.01436	<.03708	<.00182
NSGMPH26	<.00110	<.00563	<.00314	<.00436	<.00559	<.01243	<.00324	<.00909	<.12968	<.00393	<.01463	<.03000	<.04749	<.00232

Station_ID	Endosulfan-Sulfate ug/L	p,p'-DDT ug/L	Hexazinone ug/L	Methoxychlor ug/L	PCB-AR1221 ug/L	PCB-AR1232 ug/L	PCB-AR1242 ug/L	PCB-AR1248 ug/L	PCB-AR1254 ug/L	PCB-AR1260 ug/L	Technical-Chlordane ug/L	Propanil ug/L
NSGWAR01	<.01697	<.00386	<.00679	<.00205	<.01881	<.04223	<.03438	<.08935	<.14854	<.06358	<.04217	<.01519
NSGWAR02	<.01645	<.00387	<.00382	<.00283	<.01507	<.03384	<.07811	<.20298	<.13147	<.05624	<.07706	<.03386
NSGWAR03	<.02493	<.00288	<.01671	<.00212	<.02048	<.04598	<.08168	<.21224	<.16194	<.06927	<.08784	<.03761
NSGWFJ01	<.02526	<.00569	<.01882	<.01054	<.02657	<.05966	<.14699	<.38196	<.59433	<.25422	<.09250	<.07377
NSGWFJ02	<.01238	<.00633	<.01021	<.00386	<.03416	<.07670	<.21565	<.56038	<.16345	<.06992	<.08665	<.05262
NSGWFJ03	<.03538	<.00332	<.01321	<.00260	<.01262	<.02834	<.13736	<.35694	<.17869	<.07643	<.06637	<.05594
NSGWFJ04	<.01622	<.00391	<.00588	<.00147	<.01725	<.03872	<.05616	<.14592	<.06105	<.02611	<.06719	<.01747
NSGWFJ05	<.01206	<.00396	<.00466	<.00177	<.00692	<.01554	<.09484	<.24643	<.15675	<.06705	<.05526	<.01575
NSGWFJ07	<.00745	<.00160	<.00621	<.00264	<.01160	<.02604	<.08016	<.20830	<.07479	<.03199	<.07504	<.01974
NSGWLN01	<.01494	<.00317	<.00247	<.00172	<.01068	<.02399	<.06719	<.17458	<.14522	<.06212	<.04994	<.02227
NSGWLN02	<.01470	<.00195	<.00687	<.00269	<.01888	<.04240	<.05710	<.14837	<.10505	<.04493	<.06569	<.03109
NSGWLN03	<.02046	<.00370	<.00937	<.00255	<.01455	<.03268	<.07334	<.19058	<.11592	<.04959	<.04961	<.02870
NSGWFJ06	<.01308	<.00345	<.00361	<.00228	<.01079	<.02423	<.07342	<.19080	<.11462	<.04903	<.6570	<.02870
NSGWFJ08	<.01873	<.00458	<.00822	<.00288	<.02266	<.05089	<.17874	<.46682	<.14529	<.06215	<.06117	<.04954
NSGWFJ09	<.02908	<.00651	<.01142	<.00624	<.02096	<.04107	<.13353	<.34873	<.16722	<.07153	<.08056	<.04209
NSGWFJ10	<.02196	<.00554	<.00931	<.00357	<.02585	<.05805	<.15063	<.39341	<.13667	<.05803	<.09922	<.03278
NSGWFJ11	<.01557	<.00231	<.00689	<.01244	<.01806	<.04055	<.10956	<.28614	<.23380	<.10001	<.06866	<.03214
NSGWFJ12	<.02343	<.00451	<.00636	<.00380	<.02435	<.05468	<.08636	<.22555	<.17558	<.07510	<.05932	<.02439
NSGWFJ13	<.01900	<.00445	<.00834	<.00219	<.01900	<.04267	<.06288	<.16422	<.14814	<.06337	<.06047	<.03470
NSGWFJ14	<.02644	<.00284	<.00587	<.00191	<.02283	<.05126	<.06925	<.18085	<.06460	<.02763	<.05742	<.03164
NSGWFJ15	<.01311	<.00323	<.00972	<.00241	<.01062	<.02385	<.14246	<.37207	<.11430	<.04889	<.04194	<.04010
NSGWFJ16	<.01283	<.00415	<.00939	<.00362	<.01959	<.04399	<.05705	<.14900	<.14848	<.06351	<.04935	<.03136
NSGWFJ17	<.04510	<.00847	<.01539	<.00724	<.02275	<.05109	<.11157	<.30707	<.30249	<.12939	<.12251	<.07955
NSGWFJ18	<.03005	<.00681	<.01335	<.00683	<.04315	<.09690	<.14002	<.36571	<.28520	<.12200	<.04461	<.03621
NSGWLN04	<.01610	<.00338	<.01204	<.00334	<.01436	<.03225	<.07527	<.19558	<.16800	<.07187	<.03647	<.05023
NSGWLN05	<.04403	<.00563	<.01042	<.00703	<.03292	<.07393	<.25030	<.65041	<.57007	<.24385	<.05844	<.07831
NSGWLN06	<.03094	<.00571	<.00715	<.00313	<.02095	<.04704	<.18933	<.49199	<.19647	<.08404	<.07360	<.03963
NSGWDE01	<.01819	<.00455	<.00686	<.00267	<.01423	<.03195	<.13015	<.33819	<.10597	<.04533	<.04330	<.03723
NSGWDE02	<.02728	<.00575	<.01345	<.00282	<.02176	<.04886	<.07972	<.20714	<.14210	<.06079	<.03421	<.06505
NSGWDE03	<.02231	<.00498	<.01289	<.00250	<.00945	<.02122	<.14285	<.37119	<.19816	<.08477	<.03130	<.04431
NSGWDE04	<.02145	<.00396	<.00635	<.00357	<.02809	<.06308	<.17910	<.46539	<.27558	<.11917	<.05247	<.02964
NSGWDE05	<.02659	<.00390	<.00701	<.00361	<.01181	<.02653	<.11001	<.28586	<.13160	<.05629	<.05520	<.06237
NSGWDE06	<.01066	<.00358	<.01168	<.00316	<.01046	<.02348	<.08932	<.23262	<.12850	<.05497	<.05187	<.03240
NSGWDE07	<.02062	<.00279	<.00822	<.00217	<.01263	<.02837	<.07151	<.18581	<.11686	<.04999	<.07023	<.04141
NSGWDE08	<.06150	<.01144	<.01877	<.00486	<.04649	<.10438	<.36936	<.95980	<.30558	<.13071	<.10991	<.07975
NSGWDE09	<.03654	<.00764	<.01341	<.00730	<.02041	<.04583	<.17545	<.45592	<.14302	<.06118	<.07503	<.07937
NSGWDE10	<.03420	<.00612	<.00922	<.00485	<.02174	<.04881	<.10871	<.28248	<.12211	<.05223	<.04456	<.05974
NSGWDE11	<.01925	<.00430	<.00881	<.00483	<.01376	<.03089	<.12900	<.33521	<.117389	<.07438	<.08087	<.05849
NSGWDE12	<.00948	<.00448	<.01007	<.01548	<.03475	<.11976	<.11121	<.16473	<.07046	<.06365	<.04217	<.06365

Station_ID	Endosulfan-Sulfate	p,p'-DDT	Hexazinone	Methoxychlor	PCB-AR1221	PCB-AR1232	PCB-AR1242	PCB-AR1248	PCB-AR1254	PCB-AR1260	Technical-Chlordane	Propanil
	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L	ug/L
NSGWDE12	<.02723	<.00431	<.00978	<.00504	<.03625	<.08139	<.11005	<.28596	<.19460	<.08324	<.03826	<.06010
NSGWDE13	<.01617	<.00309	<.01194	<.00293	<.02215	<.04973	<.28005	<.72771	<.24198	<.10351	<.05494	<.04333
NSGWDE14	<.03101	<.00615	<.00844	<.00326	<.01992	<.04473	<.10960	<.28479	<.18079	<.07733	<.09369	<.05967
NSGWDE15	<.01757	<.00252	<.01024	<.00278	<.01575	<.03537	<.12188	<.31766	<.16894	<.07226	<.07000	<.02952
NSGWDR01	<.01912	<.00266	<.00979	<.00253	<.01350	<.03032	<.10430	<.27184	<.18616	<.07963	<.07142	<.03203
NSGWFJ19	<.01321	<.00343	<.00999	<.00148	<.02307	<.05180	<.09471	<.24885	<.18232	<.07799	<.07292	<.03973
NSGWDE16	<.00812	<.00292	<.00875	<.00261	<.01412	<.03170	<.08808	<.22259	<.26839	<.11481	<.07900	<.02492
NSGWDE17	<.00885	<.00232	<.00637	<.00204	<.02488	<.05586	<.06767	<.17337	<.15735	<.06731	<.06321	<.03842
NSGWDE18	<.02030	<.00398	<.01606	<.00331	<.02151	<.04830	<.25028	<.65324	<.28679	<.12268	<.10299	<.02895
NSGWDE19	<.01328	<.00297	<.00670	<.00293	<.01047	<.02352	<.10379	<.27053	<.11233	<.04805	<.05355	<.03171
NSGWDE20	<.01108	<.00192	<.00969	<.00212	<.02407	<.05404	<11461	<.29871	<.12227	<.05230	<.06086	<.01552
NSGWDE21	<.00918	<.00137	<.00592	<.00190	<.01692	<.03800	<.10559	<.27521	<.19359	<.08281	<.07114	<.01931
NSGWPH01	<.01585	<.00293	<.00518	<.00144	<.01172	<.02631	<.08273	<.21497	<.07757	<.03318	<.03249	<.01782
NSGWPH02	<.03252	<.00321	<.01040	<.00580	<.02369	<.05319	<.23653	<.61464	<.27495	<.11761	<.11427	<.05126
NSGWPH03	<.01289	<.00133	<.00618	<.00243	<.02329	<.05229	<.09292	<.24145	<.13495	<.05772	<.05725	<.01216
NSGWPH04	<.02171	<.00192	<.00592	<.00177	<.01060	<.02381	<.13099	<.34039	<.15314	<.06550	<.05361	<.01736
NSGWPH05	<.01081	<.00109	<.00387	<.00152	<.01249	<.02806	<.05413	<.14067	<.09844	<.04211	<.04503	<.01156
NSGWPH06	<.00612	<.00181	<.00308	<.00129	<.01001	<.02248	<.03741	<.09722	<.14032	<.06002	<.04495	<.01694
NSGWPH07	<.01014	<.00190	<.00547	<.00158	<.01022	<.02295	<.06820	<.17722	<.05537	<.02368	<.0952	<.02045
NSGWPH08	<.00822	<.00289	<.00566	<.00082	<.01175	<.02639	<.07808	<.20290	<.10418	<.04456	<.02196	<.02137
NSGWPH09	<.01180	<.00184	<.00450	<.00124	<.01009	<.02266	<.04885	<.12695	<.09891	<.04231	<.03668	<.01556
NSGWPH10	<.00849	<.00148	<.00603	<.00181	<.00987	<.02217	<.04735	<.12304	<.10814	<.04626	<.03113	<.01599
NSGWPH11	<.00539	<.00279	<.00453	<.00078	<.00543	<.01220	<.08332	<.21651	<.10619	<.04542	<.04675	<.01395
NSGWPH12	<.00810	<.00173	<.00510	<.00148	<.01215	<.02729	<.11912	<.30954	<.08645	<.03698	<.03393	<.02148
NSGWPH13	<.01114	<.00151	<.00393	<.00065	<.01421	<.03190	<.06365	<.16541	<.09799	<.04192	<.01612	<.01855
NSGWPH14	<.00784	<.00164	<.00322	<.00167	<.00900	<.02022	<.04840	<.12577	<.09979	<.04268	<.05830	<.01775
NSGWPH15	<.00652	<.00205	<.00287	<.00129	<.01235	<.02774	<.03689	<.09586	<.07302	<.03123	<.04029	<.01664
NSGWPH16	<.00786	<.00179	<.00577	<.00128	<.00869	<.01951	<.07545	<.19606	<.09679	<.04140	<.03305	<.01890
NSGWPH17	<.01089	<.00128	<.00480	<.00111	<.00939	<.02108	<.05152	<.13388	<.07284	<.03116	<.02970	<.01790
NSGWPH18	<.01555	<.00135	<.00506	<.00208	<.01542	<.03463	<.10180	<.26455	<.05877	<.02514	<.04834	<.01751
NSGWPH19	<.01047	<.00156	<.00418	<.00111	<.00752	<.01688	<.06287	<.16337	<.12314	<.05267	<.02425	<.00467
NSGWPH20	<.01045	<.00237	<.00531	<.00089	<.01050	<.02360	<.08883	<.23024	<.07975	<.03415	<.02490	<.01489
NSGWPH21	<.00648	<.00152	<.00437	<.00125	<.00522	<.01173	<.07572	<.19626	<.07931	<.03396	<.02889	<.01195
NSGWPH22	<.00572	<.00182	<.00457	<.00128	<.00634	<.01425	<.02453	<.06358	<.07707	<.03301	<.02607	<.01447
NSGWPH23	<.00993	<.00133	<.00500	<.00096	<.00578	<.01300	<.06213	<.16105	<.11360	<.04865	<.01626	<.01788
NSGWPH24	<.00865	<.00161	<.00532	<.00157	<.00584	<.01313	<.02824	<.07320	<.07261	<.03109	<.02309	<.01125
NSGWPH25	<.00833	<.00162	<.00578	<.00133	<.01010	<.02270	<.06045	<.15670	<.13550	<.05803	<.02053	<.02062
NSGWPH26	<.01119	<.00207	<.00958	<.00198	<.01034	<.02323	<.04887	<.12666	<.12183	<.05217	<.03923	<.00605



APPENDIX II

Water Quality Accuracy Checks

Sample Location:

AR01

Alkalinity (mg/L)	208
Measured conductivity (umho/cm)	610
Infinite dilution conductivity (umho/cm)	585.124634
Ionic strength (M)	0.0077
Monovalent ion activity coefficient	0.9137
Calculated conductivity (umho/cm)	488.528239
Measured TDS	299
Calculated TDS	307.402792
Ratio: Meas TDS/Calc TDS	0.97266521 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.80086555 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.62924284 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.49016333 Should be between 0.5 and 0.7

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	22.7	0.6	67.7	6.1	11.9	32.799	16.7	0.23	0	253.76	23.9
Concentration (meq/L)	0.9875	0.0153	3.3782	0.5020	0.4261	0.9253	0.3477	0.0037	0.0000	4.1591	
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.9875	0.0153	1.6891	0.2510	0.2131	0.9253	0.1738	0.0037	0.0000	4.1591	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80			54.4	44.5
Infinite dilution conductivity (umho/cm)	49.471245	1.128078	201.004685	26.6513148	23.01793	70.6689848	27.81552	0.28488686	0	185.08125	
Ionic strength	0.00049373	7.674E-06	0.00337823	0.00050191	0.001426	0.00046263	0.00034769	1.855E-06	0	0.00207956	

Cation sum (meq/L)	5.309136
Anion sum (meq/L)	5.43579009
% Difference	-1.17873353 Should be < 5%
Ion Difference	-0.12665409
Ratio: Cation sum*(100)/Measured conductivity	0.87035016 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.89111313 Should be between 0.9 and 1.1

Sample Location:

AR02

Alkalinity (mg/L)	172
Measured conductivity (umho/cm)	546
Infinite dilution conductivity (umho/cm)	534.460761
Ionic strength (M)	0.0071
Monovalent ion activity coefficient	0.9166
Calculated conductivity (umho/cm)	449.036114
Measured TDS	286
Calculated TDS	288.581528
Ratio: Meas TDS/Calc TDS	0.99105511 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.82241047 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.64266842 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.52380952 Should be between 0.5 and 0.7

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	22.7	0.6	57.8	7	10.6	23.386	34.5	0.217	0	209.84	23.6
Concentration (meq/L)	0.9875	0.0153	2.8842	0.5760	0.3796	0.6597	0.7183	0.0035	0.0000	3.4393	
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.9875	0.0153	1.4421	0.2880	0.1898	0.6597	0.3591	0.0035	0.0000	3.4393	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80			54.4	44.5
Infinite dilution conductivity (umho/cm)	49.471245	1.128078	171.61109	30.583476	20.50337	50.4025362	57.4832	0.24991499	0	153.047853	
Ionic strength	0.00049373	7.674E-06	0.00288422	0.00057596	0.000338	0.00032886	0.00071829	1.7501E-06	0	0.00171964	

Cation sum (meq/L)	4.842634
Anion sum (meq/L)	4.82078587
% Difference	0.2260807 Should be < 2%
Ion Difference	0.02184133 Should be between 0.9 and 1.1
Ratio: Cation sum*(100)/Measured conductivity	0.8869293 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.882928 Should be between 0.9 and 1.1

Sample Location:

AR03

Alkalinity (mg/L)	203
Measured conductivity (umho/cm)	720
Infinite dilution conductivity (umho/cm)	736.527658
Ionic strength (M)	0.0099
Monovalent ion activity coefficient	0.9040
Calculated conductivity (umho/cm)	601.954658
Measured TDS	384
Calculated TDS	375.280422
Ratio: Meas TDS/Calc TDS	1.02323483
Ratio: Calc cond/Meas cond	0.83606203
Ratio: Meas TDS/Calc cond	0.62342601
Ratio: Meas TDS/Meas cond	0.53333333

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	19.4	0.5	87.4	12.6	10.7	41.426	52.8	0.18	0	247.66	28.5
Concentration (meq/L)	0.8439	0.0128	4.3613	1.0369	0.3832	1.1686	1.0993	0.0029	0.0000	4.0591	
Molecular weight (mg/mM)	22.9598	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.3984	6.0171	
Concentration (mM)	0.8439	0.0128	2.1806	0.5184	0.1916	1.1686	0.5496	0.0029	0.0000	4.0591	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	42.27939	0.940065	259.49497	55.0502568	20.686795	89.2831379	87.94368	0.20730276	0	186.6321	
Ionic strength	0.00042195	6.395E-06	0.00436126	0.00103673	0.00038327	0.00058431	0.0010993	1.4517E-06	0	0.00203	

Cation sum (meq/L)

6.32997426

2.375062

Should be < 5%

0.30799874

0.92164042

Should be between 0.9 and 1.1

0.87916509

Should be between 0.9 and 1.1

JF01

Sample Location:

AR03

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	245										
Concentration (meq/L)	630										
Molecular weight (mg/mM)	647.831033										
Concentration (mM)	0.0089										
Monovalent ion activity coefficient	0.9081										
Calculated conductivity (umho/cm)	534.236343										
Measured TDS	320										
Calculated TDS	323.75913										
Ratio: Meas TDS/Calc TDS	0.98838512										
Ratio: Calc cond/Meas cond	0.8479842										
Ratio: Meas TDS/Calc cond	0.60602229										
Ratio: Meas TDS/Meas cond	0.50793651										

Cation sum (meq/L)

6.2689246

5.83025306

3.62563103

Should be < 5%

0.48867154

0.9950674

Should be between 0.9 and 1.1

0.92543699

Should be between 0.9 and 1.1

0.002449

0.002449

0.002449

0.002449

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	18.2	0.4	83.5	13.6	5.06	21.152	15.7	0.478	0	288.9	18.7
Concentration (meq/L)	0.7917	0.0102	4.1667	1.1191	0.1812	0.5967	0.3269	0.0077	0.0000	4.8990	
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0635	62.0049	18.3984	6.0171	
Concentration (mM)	0.7917	0.0102	2.0833	0.5595	0.0906	0.5967	0.1634	0.0077	0.0000	4.8990	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	39.66417	0.732052	247.915675	59.4193248	9.7874568	45.5677211	26.14992	0.550504	0	218.0042	
Ionic strength	0.00039585	5.116E-06	0.00416665	0.00111901	0.00018125	0.000295836	0.00032687	3.8551E-06	0	0.002449	

Sample Location:

JF02

Alkalinity (mg/L)	276
Measured conductivity (umho/cm)	1101
Infinite dilution conductivity (umho/cm)	1169.09699
Ionic strength (M)	0.0157
Monovalent ion activity coefficient	0.8844
Calculated conductivity (umho/cm)	914.378377
Measured TDS	624
Calculated TDS	601.247224
Ratio: Meas TDS/Calc TDS	1.03784263
Ratio: Calc cond/Meas cond	0.83049807
Ratio: Calc TDS/Calc cond	0.65754751
Ratio: Meas TDS/Meas cond	0.56675749
Infinite dilution conductivity (umho/cm)	Should be between 0.9 and 1.1
Monovalent ion activity coefficient	Should be between 0.9 and 1.1
Calculated conductivity (umho/cm)	Should be between 0.55 and 0.7
Measured TDS	Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2
Concentration (mg/L)	37.1	1.4	118	17	17.9	42.117	174	0.165	0	336.72	28
Concentration (meq/L)	1.6139	0.0358	5.8892	1.3989	0.6410	3.6227	0.0027	0.0000	5.5188		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	1.6139	0.0358	2.9441	0.6894	0.3206	1.1981	1.8113	0.0027	0.0000	5.5188	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	80.853685	2.632182	360.3479	74.274156	34.623612	90.7724115	289.8144	0.19002753	0	245.5884	
Ionic strength	0.00080653	1.78061E-05	0.0058832	0.00139876	0.000594118	0.00059406	0.00362268	1.3307E-06	0	0.002759	

Cation sum (meq/L)	9.577791
Anion sum (meq/L)	10.3323028
% Difference	-3.7895945
Ion Difference	Should be < 5%
Ratio: Cation sum/(100)/Measured conductivity	-0.75451182
Ratio: Anion sum/(100)/Measured conductivity	0.93644712
Infinite dilution conductivity (umho/cm)	Should be between 0.9 and 1.1
Ionic strength	Should be between 0.9 and 1.1

Sample Location:

JF03

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2
Concentration (mg/L)	3.5	1.2	81.7	17.7	5.12	58.061	32.2	0.149	0	303.78	20.8
Concentration (meq/L)	1.5225	0.0307	4.0768	1.4565	0.1833	1.6379	0.6704	0.0024	0.0000	4.9790	
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	1.5225	0.0307	2.0384	0.7282	0.0917	1.6379	0.3352	0.0024	0.0000	4.9790	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	76.27725	2.256156	242.571385	77.3326036	9.9035136	125.136522	53.63232	0.17160062	0	221.5635	
Ionic strength	0.00076125	1.5348E-05	0.00407683	0.00145636	0.000081895	0.0006704	1.2017E-06	0	0.002489		

Cation sum (meq/L)	7.2699062
Anion sum (meq/L)	7.28966238
% Difference	-0.13569207
Ion Difference	Should be < 5%
Ratio: Cation sum/(100)/Measured conductivity	-0.01975618
Ratio: Anion sum/(100)/Measured conductivity	0.90216594
Infinite dilution conductivity (umho/cm)	Should be between 0.9 and 1.1
Ionic strength	Should be between 0.9 and 1.1

Sample Location:

JF04

Alkalinity (mg/L)	340
Measured conductivity (umho/cm)	1150
Infinite dilution conductivity (umho/cm)	1051.33696
Ionic strength (M)	0.0142
Monovalent ion activity coefficient	0.8891
Calculated conductivity (umho/cm)	831.143735
Measured TDS	594
Calculated TDS	519.68916
Ratio: Meas TDS/Calc TDS	1.14299094 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.72273388 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.62526599 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.51652714 Should be between 0.55 and 0.7

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2	24
Concentration (meq/L)	71.3	1.5	114	22.1	10.4	30.388	41.9	0.144	0	414.8		
Concentration (meq/L)	3.01016	0.0384	5.6886	1.8186	0.3724	0.8572	0.8724	0.0023	0.0000	6.7986		
Molecular weight (mg/mM)	22.36986	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	3.01016	0.0384	2.8443	0.9092	0.1863	0.8572	0.4362	0.0023	0.0000	6.7986		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	155.387855	2.820195	338.4717	96.5564028	20.116512	65.4935547	69.78864	0.16584221	0	302.5365		
Ionic strength	0.00155076	1.9185E-05	0.0056886	0.00181639	0.00037253	0.00042862	0.00087236	1.1614E-06	0	0.003395		

Cation sum (meq/L)
Anion sum (meq/L)
% Difference

Ion Difference
Ratio: Cation sum*(100)/Measured conductivity
Ratio: Anion sum*(100)/Measured conductivity
0.74178245 Should be between 0.9 and 1.1

Sample Location:

JF05

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2	24
Concentration (meq/L)	297											
Concentration (meq/L)	1400											
Molecular weight (mg/mM)	1488.085											
Concentration (mM)	0.0182											
Monovalent ion activity coefficient	0.8778											
Calculated conductivity (umho/cm)	1146.10878											
Measured TDS	697											
Calculated TDS	710.221578											
Ratio: Meas TDS/Calc TDS	0.981363387	Should be between 0.9 and 1.1										
Ratio: Calc cond/Meas cond	0.81864984	Should be between 0.9 and 1.1										
Ratio: Calc TDS/Calc cond	0.61968023	Should be between 0.55 and 0.7										
Ratio: Meas TDS/Meas cond	0.49785114	Should be between 0.55 and 0.7										

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2	24
Concentration (meq/L)	64	1.3	132	33.1	4.69	184	83.7	0.269	0	36.34		
Concentration (meq/L)	2.7840	0.0333	6.5568	2.7238	0.1679	5.1906	1.7426	0.0043	0.0000	5.9388		
Molecular weight (mg/mM)	22.36986	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	2.7840	0.0333	3.2934	1.3617	0.0840	5.1906	0.8713	0.0043	0.0000	5.9388		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	139.4784	2.444169	391.9146	144.616151	9.0717732	396.564896	159.41072	0.30980246	0	264.2745		
Ionic strength	0.001392	1.6627E-05	0.0065868	0.00272347	0.00168	0.00255532	0.00174263	2.1695E-06	0	0.002969		

Cation sum (meq/L)
Anion sum (meq/L)
% Difference
Ion Difference
Ratio: Cation sum*(100)/Measured conductivity
Ratio: Anion sum*(100)/Measured conductivity
0.87827156 Should be between 0.9 and 1.1
0.919744 Should be between 0.9 and 1.1

Sample Location:

JF07

Alkalinity (mg/L)	443
Measured conductivity (umho/cm)	1040
Infinite dilution conductivity (umho/cm)	1069.51687
Ionic strength (M)	0.0145
Monovalent ion activity coefficient	0.8882
Calculated conductivity (umho/cm)	843.661657
Measured TDS	541
Calculated TDS	546.643182
Ratio: Meas TDS/Calc TDS	0.98967686
Ratio: Calc cond/Meas cond	0.81120352
Ratio: Calc TDS/Calc cond	0.64794892
Ratio: Meas TDS/Meas cond	0.52019231
Infinite dilution conductivity (umho/cm)	Should be between 0.9 and 1.1
Ionic strength (M)	Should be between 0.9 and 1.1
Monovalent ion activity coefficient	Should be between 0.55 and 0.7
Calculated conductivity (umho/cm)	Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	50.6	1.2	111	26.3	6.43	26.024	29.8	0.145	0	540.46	29.4
Concentration (meq/L)	2.2011	0.0307	5.5389	2.1642	0.2303	0.7341	0.6204	0.0023	0.0000	8.8881	
Molecular weight (mg/mmL)	22.9898	39.0983	40.0780	24.3050	55.8470	36.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	2.2011	0.0307	2.7695	1.0820	0.1152	0.7341	0.3102	0.0023	0.0000	8.8881	
Charge z (absolute value)			1	1	2	2	1	2	1	1	
Equivalent conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	110.27511	2.256156	329.56455	114.906488	12.4374204	56.0890699	49.63488	0.1669389	0	394.1872	
Ionic strength	0.00110055	1.5348E-05	0.0055389	0.00216396	0.00023032	0.00036707	0.00062044	1.1694E-06	0	0.004429	
Cation sum (meq/L)	10.1651813										
Anion sum (meq/L)	10.2150513										
% Difference	-0.24469785	Should be < 5%									
Ion Difference	-0.04986989										
Ratio: Cation sum*(100)/Measured conductivity	0.97742128	Should be between 0.9 and 1.1									
Ratio: Anion sum*(100)/Measured conductivity	0.98221647	Should be between 0.9 and 1.1									

Sample Location:

LN01

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	233										
Concentration (meq/L)	656										
Molecular weight (mg/mmL)	649.656314										
Concentration (mM)	0.0086										
Monovalent ion activity coefficient	0.9095										
Calculated conductivity (umho/cm)	537.578885										
Measured TDS	339										
Calculated TDS	340.785642										
Ratio: Meas TDS/Calc TDS	0.99476022	Should be between 0.9 and 1.1									
Ratio: Calc cond/Meas cond	0.8194797	Should be between 0.9 and 1.1									
Ratio: Meas TDS/Calc cond	0.633927	Should be between 0.55 and 0.7									
Ratio: Meas TDS/Meas cond	0.51676829	Should be between 0.55 and 0.7									
Cation sum (meq/L)											
Anion sum (meq/L)											
% Difference											
Ion Difference											
Ratio: Cation sum*(100)/Measured conductivity	0.91018537	Should be between 0.9 and 1.1									
Ratio: Anion sum*(100)/Measured conductivity	0.91770296	Should be between 0.9 and 1.1									

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	28.2	1	70.5	11.5	7.1	29.276	25.6	0.139	0	284.26	27.7
Concentration (meq/L)	1.2267	0.0256	3.5180	0.9463	0.2543	0.8259	0.5330	0.0022	0.0000	4.6590	
Molecular weight (mg/mmL)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	1.2267	0.0256	1.7580	0.4731	0.1272	0.8259	0.2685	0.0022	0.0000	4.6590	
Monovalent ion activity coefficient	1	1	2	2	2	2	1	1	1	1	
Calculated conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	61.45767	1.38013	209.318025	50.244282	13.733388	63.0869233	42.63936	0.1600838	0.121E-06	0	207.3265
Ionic strength	0.00061335	1.279E-05	0.00351795	0.00094622	0.00025432	0.00041294	0.00053299	0.0002333	0	0.002333	
Cation sum (meq/L)	5.970816										
Anion sum (meq/L)	6.02013143										
% Difference	-0.41127217	Should be < 5%									
Ion Difference	0.04931543										
Ratio: Cation sum*(100)/Measured conductivity	0.91018537	Should be between 0.9 and 1.1									
Ratio: Anion sum*(100)/Measured conductivity	0.91770296	Should be between 0.9 and 1.1									

Sample Location:

LN02

Alkalinity (mg/l)	297
Measured conductivity (umho/cm)	794
Infinite dilution conductivity (umho/cm)	793.381957
Ionic strength (M)	0.0105
Monovalent ion activity coefficient	0.9017
Calculated conductivity (umho/cm)	645.011727
Measured TDS	414
Calculated TDS	410.811578
Ratio: Meas TDS/Calc TDS	1.00776128 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.81235734 Should be between 0.9 and 1.1
Ratio: Meas TDS/Calc cond	0.6369056 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.52141058 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	F _e	Cl	SO ₄	NO ₃	F	HCO ₃	SIO ₂	
Concentration (mg/L)	29	1.2	94.5	9	12.2	37.2	23.9	0.149	0	362.34	25.5	
Concentration (meq/L)	1.2615	0.0307	4.7156	0.7406	0.4359	1.0494	0.4976	0.0024	0.0000	5.9388		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	16.9984	61.0171		
Concentration (mM)	1.2615	0.0307	2.3578	0.3703	0.2185	1.0494	0.2488	0.0024	0.0000	5.9388		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	63.20115	2.256158	280.575235	39.321612	23.586216	80.1750768	39.80784	0.17160062	0	264.2745		
Ionic strength	0.00063075	1.5348E-05	0.00471555	0.00074052	0.0004477	0.00052471	0.0004976	1.2017E-06	0	0.002969		

Cation sum (meq/L)	7.185238
Anion sum (meq/L)	7.48816997
% Difference	-2.06446864 Should be < 5%
Ion Difference	-0.30292797
Ratio: Cation sum*(100)/Measured conductivity	0.90494181 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.94303995 Should be between 0.9 and 1.1

Sample Location:

LN03

Constituent:	Na	K	Ca	Mg	F _e	Cl	SO ₄	NO ₃	F	HCO ₃	SIO ₂	
Concentration (mg/L)	389	957	891.995673	0.0120	0.8961	716.345997	462	4.9	0.141	0.0023	0.0000	474.58
Concentration (meq/L)	1.1658	0.0307	5.7385	0.4603	0.7385	0.8878	0.1020	0.0023	0.0000	7.7784		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	16.9984	61.0171		
Concentration (mM)	1.1658	0.0307	2.8693	0.2304	0.3994	0.8878	0.0510	0.0023	0.0000	7.7784		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	58.40658	2.256156	341.44075	24.4667808	43.134444	67.8298392	8.16144	0.16238716	0	346.1373		
Ionic strength	0.0005829	1.5348E-05	0.0057385	0.000465077	0.00079879	0.00044391	0.00010202	1.1372E-06	0	0.003889		

Constitution (meq/L)	8.194383
Anion sum (meq/L)	8.77048365
% Difference	-3.39584544 Should be < 5%
Ion Difference	-0.57610065
Ratio: Cation sum*(100)/Measured conductivity	0.85625737 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.91645597 Should be between 0.9 and 1.1

Sample Location:

JF06

Alkalinity (mg/l)	235
Measured conductivity (umho/cm)	882
Infinite dilution conductivity (umho/cm)	909.99167
Ionic strength (M)	0.0113
Monovalent ion activity coefficient	0.8986
Calculated conductivity (umho/cm)	734.872479
Measured TDS	452
Calculated TDS	457.29739
Ratio: Meas TDS/Calc TDS	0.998415388
Ratio: Calc cond/Meas cond	0.85318675
Ratio: Calc TDS/Calc cond	0.67228128
Ratio: Meas TDS/Meas cond	0.51247166
Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm^2/equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	7.8176198
Anion sum (meq/L)	8.11655519
% Difference	-1.87605444
Ion Difference	Should be < 5%
Ratio: Cation sum*(100)/Measured conductivity	-0.29893539
Ratio: Anion sum*(100)/Measured conductivity	0.88635145
Ratio: Cation sum*(100)/Measured conductivity	Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.92024435
Should be between 0.9 and 1.1	

Sample Location:

66

Alkalinity (mg/l)	136
Measured conductivity (umho/cm)	334
Infinite dilution conductivity (umho/cm)	308.09049
Ionic strength (M)	0.0040
Monovalent ion activity coefficient	0.9348
Calculated conductivity (umho/cm)	289.202538
Measured TDS	195
Calculated TDS	196.35799
Ratio: Meas TDS/Calc TDS	0.99308411
Ratio: Calc cond/Meas cond	0.80599562
Ratio: Calc TDS/Calc cond	0.72940616
Ratio: Meas TDS/Meas cond	0.58383234
Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm^2/equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	2.8154777
Anion sum (meq/L)	3.088806449
% Difference	-4.62932984
Ion Difference	Should be < 2%
Ratio: Cation sum*(100)/Measured conductivity	-0.21332879
Ratio: Anion sum*(100)/Measured conductivity	0.8429574
Should be between 0.9 and 1.1	
Ratio: Cation sum*(100)/Measured conductivity	0.92479236
Should be between 0.9 and 1.1	

Sample Location:

JF09

Alkalinity (mg/L)	200
Measured conductivity (umho/cm)	450
Infinite dilution conductivity (umho/cm)	393.795436
Ionic strength (M)	0.0051
Monovalent ion activity coefficient	0.9277
Calculated conductivity (umho/cm)	338.893406
Measured TDS	252
Calculated TDS	258.5178
Ratio: Meas TDS/Calc TDS	0.97478781
Ratio: Calc cond/Meas cond	0.75395946
Ratio: Meas TDS/Calc cond	0.76282924
Ratio: Meas TDS/Meas cond	0.56

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	51
Concentration (mg/L)	22.6	1.7	37	0.5	15.4	8.074	1.9	0.039	0.33	244		
Concentration (meq/L)	0.9831	0.0435	1.8483	0.0411	0.5515	0.2278	0.0395	0.0006	0.0174	3.9992		
Molecular weight (mg/mM)	22.9858	39.0983	40.0780	24.3050	55.8470	35.4527	96.98536	62.0049	18.9884	61.0171		
Concentration (mM)	0.9831	0.0435	0.9232	0.0206	0.2758	0.2278	0.0198	0.0006	0.0174	3.9992		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	49.25331	3.196221	108.85485	2.184534	29.787912	17.4014401	3.16464	0.0449156	0.9449328	177.96262		
Ionic strength	0.00049155	2.1743E-05	0.0018463	4.114E-05	0.00055163	0.00011388	3.9558E-05	3.1454E-07	8.6856E-06	0.001999558		
Cation sum (meq/L)	3.465505											
Anion sum (meq/L)	4.28448381											
% Difference	-10.5675069	Should be < 2%										
Ion Difference	-0.81898081											
Ratio: Cation sum*(100)/Measured conductivity	0.777011222	Should be between 0.9 and 1.1										
Ratio: Anion sum*(100)/Measured conductivity	0.95210796	Should be between 0.9 and 1.1										

Sample Location:

JF10

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	25.1
Concentration (mg/L)	16.4	0.8	96.2	14.4	6.83	14.949	7.5	0.046	0.31	383.08		
Concentration (meq/L)	0.7134	0.0205	4.8004	1.1850	0.2446	0.4217	0.1562	0.0007	0.0163	6.2787		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.7134	0.0205	2.4002	0.5924	0.1223	0.4217	0.0781	0.0007	0.0163	6.2787		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	35.74134	1.504104	285.62261	62.9145792	13.2111324	32.2187426	12.492	0.05297737	0.88772096	279.401313		
Ionic strength	0.0003587	1.0232E-05	0.00480038	0.00118483	0.00024465	0.00021086	0.00015615	3.7098E-07	8.1592E-06	0.00313934		
Cation sum (meq/L)	6.9638023											
Anion sum (meq/L)	6.87360287											
% Difference	0.6518522	Should be < 5%										
Ion Difference	0.05019843											
Ratio: Cation sum*(100)/Measured conductivity	0.93978439	Should be between 0.9 and 1.1										
Ratio: Anion sum*(100)/Measured conductivity	0.92761172	Should be between 0.9 and 1.1										

Sample Location:**JF11**

Alkalinity (mg/L)	294
Measured conductivity (umho/cm)	705
Infinite dilution conductivity (umho/cm)	723.615754
Ionic strength (M)	0.0057
Monovalent ion activity coefficient	0.9050
Calculated conductivity (umho/cm)	592.888933
Measured TDS	367
Calculated TDS	379.319988
Ratio: Meas TDS/Calc TDS	0.96752054 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.84069352 Should be between 0.9 and 1.1
Ratio: Meas TDS/Calc cond	0.63999838 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.52056738 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	28.1	0.8	84.3	10.6	9.27	26.898	15.6	0.039	0.26	358.88	27.1	
Concentration (mg/L)	1.2224	0.0205	4.2066	0.3723	0.3320	0.7588	0.3248	0.0005	0.0137	5.8788		
Molecular weight (mg/mM)	22.9988	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	1.2224	0.0205	2.1033	0.4361	0.1680	0.7588	0.1624	0.0005	0.0137	5.8788		
Charge z (absolute value)	1	1	2	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	61.239735	1.504104	250.290915	46.3121208	17.9307756	57.9717531	25.98336	0.0339878	0.7445016	261.605051		
Ionic strength	0.00061118	1.03232E-05	0.00420657	0.00087217	0.00033205	0.00031794	0.000312479	2.3388E-07	6.8432E-06	0.00293938		

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum*(100)/Measured conductivity

Ratio: Anion sum*(100)/Measured conductivity

Sample Location:

JF12

Alkalinity (mg/L)	49
Measured conductivity (umho/cm)	863
Infinite dilution conductivity (umho/cm)	881.905838
Ionic strength (M)	0.0125
Monovalent ion activity coefficient	0.8945
Calculated conductivity (umho/cm)	705.603874
Measured TDS	444
Calculated TDS	447.951468
Ratio: Meas TDS/Calc TDS	0.99117881 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.81761747 Should be between 0.9 and 1.1
Ratio: Meas TDS/Calc cond	0.63464837 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.51446456 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	14.6	0.7	128	15	9.26	9.183	0.5	0.01	0.25	498.98	25.1	
Concentration (mg/L)	0.6551	0.0179	6.3872	1.2344	0.3316	0.2591	0.0104	0.0002	0.0132	8.1783		
Molecular weight (mg/mM)	22.9988	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.6551	0.0179	3.1936	0.6171	0.1658	0.2591	0.0052	0.0002	0.0132	8.1783		
Charge z (absolute value)	1	1	2	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	31.81851	1.316091	380.0384	65.55602	17.9114328	19.7916057	0.8328	0.01151682	0.715904	363.933568		
Ionic strength	0.00031755	8.953E-06	0.0063872	0.0012342	0.00033169	0.00012853	1.041E-05	8.055E-08	6.58E-06	0.00408914		

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum*(100)/Measured conductivity

Ratio: Anion sum*(100)/Measured conductivity

Sample Location:	JF-13
Alkalinity (mg/l)	285
Measured conductivity (umho/cm)	743
Infinite dilution conductivity (umho/cm)	749.799502
Ionic strength (M)	0.0102
Monovalent ion activity coefficient	0.9029
Calculated conductivity (umho/cm)	611.15499
Measured TDS	381
Calculated TDS	379.21409
Ratio: Meas TDS/Calc TDS	1.004705
Ratio: Calc cond/Meas cond	0.82268505
Ratio: Calc TDS/Calc cond	0.62038608
Ratio: Meas TDS/Meas cond	0.522786

Constituent	Concentration (mg/L)	K	Ca	Mg	Fe	Cl	SC4	NO3	F	HCO3	SiO2	
Na	17.7	0.5	84	21	8.24	37.67	10.5	0.03	0.31	347.7	28.3	
Concentration (meq/L)	0.7700	0.0128	4.1916	1.7281	0.2951	1.0627	0.2168	0.0005	0.0163	5.6988		
Molecular weight (mg/mm)	22.9896	39.9863	40.0780	24.3050	55.8470	35.4527	96.0546	62.0049	18.994	61.0171		
Concentration (mM)	0.7700	0.0128	2.0958	0.8639	0.1476	1.0627	0.1083	0.0005	0.0163	5.6988		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	1	
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	38.574495	0.94065	249.4002	91.750428	15.95384672	81.18904134	17.4888	0.03455046	0.88772096	253.596733		
Ionic strength	0.00088498	6.395E-06	0.0041916	0.00172788	0.00029516	0.00053134	0.00021861	2.4195E-07	8.1592E-06	0.0028494		

Sample location:

Zation sum (meq/L)	Anion sum (meq/L)	% Difference	Ratio: Cation sum*/(100) Measured conductivity	Ratio: Anion sum*/(100) Measured conductivity
6.8975044	6.956886	D.00441891 Should be < 5%	0.0005184	0.94179063 Should be between 0.9 and 1.1
6.956886	D.00441891 Should be < 5%	0.0005184	0.94179063 Should be between 0.9 and 1.1	0.9417074 Should be between 0.9 and 1.1

1514

Sample Location:

JF-15

Alkalinity (mg/L)	157
Measured conductivity (umho/cm)	583
Infinite dilution conductivity (umho/cm)	590.143313
Ionic strength (M)	0.0078
Monovalent ion activity coefficient	0.9135
Calculated conductivity (umho/cm)	492.453492
Measured TDS	294
Calculated TDS	326.879228
Ratio: Meas TDS/Calc TDS	0.89941478 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.844468867 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.863377683 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.50428816 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	19.1	2.2	40.5	0.03	70.5	51.279	22.2	0.06	0.13	191.54	26.7	
Concentration (meq/L)	0.8309	0.0563	2.0210	0.0025	2.5246	1.4466	0.4622	0.0010	0.0058	3.1393		
Molecular weight (mg/mM)	22.9836	39.0983	40.0780	24.3050	55.9470	36.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.8309	0.0563	1.0105	0.0012	1.2627	1.4466	0.2311	0.0010	0.0058	3.1393		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	41.625385	4.136286	120.246525	0.13107204	136.36674	110.518757	36.97632	0.06910092	0.37227008	139.700657		
Ionic strength	0.00041543	2.8138E-05	0.00202095	2.468E-06	0.00252531	0.00072329	0.0004622	4.839E-07	3.4216E-06	0.00156867		

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum*(100)/Measured conductivity

Ratio: Anion sum*(100)/Measured conductivity

Sample Location:

JF-16

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	305											
Concentration (meq/L)	632											
Molecular weight (mg/mM)	678.14447											
Concentration (M)	0.0095											
Monovalent ion activity coefficient	0.9056											
Calculated conductivity (umho/cm)	556.119028											
Measured TDS	348											
Calculated TDS	351.86757											
Ratio: Meas TDS/Calc TDS	0.98960845 Should be between 0.9 and 1.1											
Ratio: Calc cond/Meas cond	0.85294329 Should be between 0.9 and 1.1											
Ratio: Calc TDS/Calc cond	0.63271989 Should be between 0.55 and 0.7											
Ratio: Meas TDS/Meas cond	0.53374233 Should be between 0.55 and 0.7											

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	11.9	0.6	99.3	7.8	12.8	11.272	0.3180	0.6	0.034	0.3	372.1	24.4
Concentration (meq/L)	0.5177	0.0153	4.9551	0.6419	0.4584	0.3180	0.0104	0.0005	0.0158	0.0158	6.0987	
Molecular weight (mg/mM)	22.9898	38.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.5177	0.0153	2.4775	0.3209	0.2292	0.3180	0.0052	0.0005	0.0158	0.0158	6.0987	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	25.934265	1.128078	294.826665	34.0787304	24.758784	24.2939104	0.8328	0.03915719	0.85690848	271.392996		
Ionic strength	0.00025883	7.674E-06	0.00495507	0.00064178	0.00045855	0.00015899	1.041E-05	2.7421E-07	7.896E-06	0.00304936		

Cation sum (meq/L)	6.588298
Anion sum (meq/L)	6.44345254
% Difference	1.1148122 Should be < 5%
Ion Difference	0.14484546
Ratio: Cation sum*(100)/Measured conductivity	1.01047515 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.98825959 Should be between 0.9 and 1.1

Sample Location:

JF17

Alkalinity (mg/l)	196
Measured conductivity (umho/cm)	546
Infinite dilution conductivity (umho/cm)	573.103248
Ionic strength (M)	0.0076
Monovalent ion activity coefficient	0.9140
Calculated conductivity (umho/cm)	478.758381
Measured TDS	302
Calculated TDS	307.335304
Ratio: Meas TDS/Calc TDS	0.98264012 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.87684685 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.6419424 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.55311355 Should be between 0.55 and 0.7

Concentration (mg/L)	305
Concentration (meq/L)	21
Molecular weight (mg/MM)	0.9135
Concentration (mM)	22.9898
Charge z (absolute value)	0.0281
Equivalent conductivity (mho-cm^2/equivalent)	0.9135
Infinite dilution conductivity (umho/cm)	50.1
Ionic strength	0.0045675
Cation sum (meq/L)	5.2301451
Anion sum (meq/L)	5.20909194
% Difference	0.20167336 Should be < 5%
Ion Difference	0.02105316
Ratio: Cation sum*(100)/Measured conductivity	0.957790203 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.95404614 Should be between 0.9 and 1.1

Sample Location:

JF18

Alkalinity (mg/l)	305
Measured conductivity (umho/cm)	911
Infinite dilution conductivity (umho/cm)	950.754878
Ionic strength (M)	0.0123
Monovalent ion activity coefficient	0.8953
Calculated conductivity (umho/cm)	762.161317
Measured TDS	480
Calculated TDS	476.622757
Ratio: Meas TDS/Calc TDS	0.96511412 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.836860937 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.62557131 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.50493963 Should be between 0.55 and 0.7

Sample Location:

JF18

Concentration (mg/L)	305
Concentration (meq/L)	30.9
Molecular weight (mg/MM)	1.3442
Concentration (mM)	22.9898
Charge z (absolute value)	0.0281
Equivalent conductivity (mho-cm^2/equivalent)	1.3442
Infinite dilution conductivity (umho/cm)	50.1
Ionic strength	0.0067208
Cation sum (meq/L)	8.417138
Anion sum (meq/L)	8.69605358
% Difference	-1.62982795 Should be < 5%
Ion Difference	-0.27891558
Ratio: Cation sum*(100)/Measured conductivity	0.9239449 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.95456132 Should be between 0.9 and 1.1

Sample Location:

DE02

Alkalinity (mg/L)	312
Measured conductivity (umho/cm)	1218
Infinite dilution conductivity (umho/cm)	1415.03183
Ionic strength (M)	0.0176
Monovalent ion activity coefficient	0.8793
Calculated conductivity (umho/cm)	1094.05732
Measured TDS	690
Calculated TDS	708.95968
Ratio: Meas TDS/Calc TDS	0.97325703
Ratio: Calc cond/Meas cond	Should be between 0.9 and 1.1
Ratio: Meas TDS/Calc cond	0.89824032
Ratio: Meas TDS/Meas cond	Should be between 0.9 and 1.1
Ratio: Meas TDS/Meas cond	0.64800964
Ratio: Meas TDS/Meas cond	Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.56650246
Ratio: Meas TDS/Meas cond	Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	72.3	1.3	113	25.2	9.12	126	145.2	0.029	0.25	380.64	29.4	
Concentration (meq/L)	3.1450	0.0333	5.6387	2.0737	0.3266	3.5545	3.0231	0.0005	0.0132	6.2387		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.9836	62.0049	18.9884	61.0171		
Concentration (mM)	3.1450	0.0333	2.8194	1.0367	0.1633	3.5545	1.5115	0.0005	0.0132	6.2387		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	241.84512	0.03399878	0.715804	277.621687		
Infinite dilution conductivity (umho/cm)	157.567005	2.444169	335.50265	110.100514	17.6406336	271.560744						
Ionic strength	0.00157252	1.6627E-05	0.0056387	0.00207346	0.00032668	0.00177723	0.00302306	2.3389E-07	6.58E-06	0.00311934		
Cation sum (meq/L)	11.2172992											
Anion sum (meq/L)	12.8288414											
% Difference	-6.70575433	Should be < 5%										
Ion Difference	-1.61254217											
Ratio: Cation sum ⁿ (100)/Measured conductivity	0.92096033	Should be between 0.9 and 1.1										
Ratio: Anion sum ⁿ (100)/Measured conductivity	1.05335315	Should be between 0.9 and 1.1										

Sample Location:

DE03

Alkalinity (mg/L)	370
Measured conductivity (umho/cm)	1025
Infinite dilution conductivity (umho/cm)	1112.03897
Ionic strength (M)	0.0143
Monovalent ion activity coefficient	0.8886
Calculated conductivity (umho/cm)	878.059513
Measured TDS	578
Calculated TDS	573.35138
Ratio: Meas TDS/Calc TDS	1.0081078
Ratio: Calc cond/Meas cond	Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.85664343
Ratio: Meas TDS/Meas cond	Should be between 0.55 and 0.7
Ionic strength	0.65287554
Ratio: Meas TDS/Meas cond	Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.56390244
Ratio: Meas TDS/Meas cond	Should be between 0.55 and 0.7

Sample Location:

DE03

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	62.7	1.5	99.2	20.7	10	58.77	67.1	0.038	0.2	451.4	31.2	
Concentration (meq/L)	2.7275	0.0384	4.9501	1.7034	0.3581	1.6379	1.3970	0.0005	0.0105	7.3984		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.9636	62.0049	18.9884	61.0171		
Concentration (mM)	2.7275	0.0384	2.4750	0.8516	0.1791	1.6379	0.6985	0.0005	0.0105	7.3984		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	136.645245	2.820195	294.52976	90.4397076	19.3428	126.66369	111.76176	0.00082895	0.00139702	2.2582E-07	5.254E-06	0.00369922
Ionic strength	0.00136313	1.9185E-05	0.00495008	0.0017032	0.00035342							
Cation sum (meq/L)	9.777403											
Anion sum (meq/L)	10.4643493											
% Difference	-3.39570984	Should be < 5%										
Ion Difference	-0.68894634											
Ratio: Cation sum ⁿ (100)/Measured conductivity	0.95689298	Should be between 0.9 and 1.1										
Ratio: Anion sum ⁿ (100)/Measured conductivity	1.02091213	Should be between 0.9 and 1.1										

Sample Location:

DE04

Alkalinity (mg/L)	346
Measured conductivity (umho/cm)	1056
Infinite dilution conductivity (umho/cm)	1130.36791
Ionic strength (I)	0.0145
Nonvalent ion activity coefficient	0.8801
Calculated conductivity (umho/cm)	891.955732
Measured TDS	566
Calculated TDS	570.153404
Ratio: Meas TDS/Calc TDS	0.99271529
Ratio: Calc cond/Meas cond	0.84455525
Ratio: Calc TDS/Calc cond	0.63921715
Ratio: Meas TDS/Meas cond	0.53598485
Infinite dilution conductivity (umho/cm)	Should be between 0.9 and 1.1
Nonvalent ion activity coefficient	Should be between 0.9 and 1.1
Calculated conductivity (umho/cm)	Should be between 0.55 and 0.7
Measured TDS	Should be between 0.55 and 0.7

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NOS	F	HCO3	SiO2
Concentration (mg/L)	35.9	2	122	18	9.85	94.045	46.8	0.032	0.17	422.12	33.8
Concentration (meq/L)	1.5616	0.0512	6.0878	1.4812	0.3527	2.6550	0.9744	0.0005	0.0089	6.9185	
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9884	61.0171	
Concentration (mM)	1.5616	0.0512	3.0439	0.7405	0.1764	2.6550	0.4872	0.0005	0.0089	6.9185	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	78.238685	3.76026	362.2241	78.643224	19.052658	202.689322	77.95008	0.03685382	0.48681472	307.875333	
Ionic strength	0.0007802	2.558E-05	0.0060878	0.00148104	0.00035283	0.0013285	0.00097438	2.5808E-07	4.4744E-06	0.00345927	
Cation sum (meq/L)	9.5345585										
Anion sum (meq/L)	10.5553972										
% Difference	-5.08133878										
Ion Difference	-1.0208383871										
Ratio: Cation sum*(100)/Measured conductivity	0.9828938										
Ratio: Anion sum*(100)/Measured conductivity	0.98956413										

Sample Location:	DE05
Alkalinity (mg/L)	335
Measured conductivity (umho/cm)	1202
Infinite dilution conductivity (umho/cm)	1294.2717
Ionic strength (I)	0.0162
Nonvalent ion activity coefficient	0.8831
Calculated conductivity (umho/cm)	1009.40054
Measured TDS	663
Calculated TDS	657.71979
Ratio: Meas TDS/Calc TDS	1.00802865
Ratio: Calc cond/Meas cond	0.8397676
Ratio: Calc TDS/Calc cond	0.85159445
Ratio: Meas TDS/Meas cond	0.5515807
Infinite dilution conductivity (umho/cm)	Should be between 0.9 and 1.1
Nonvalent ion activity coefficient	Should be between 0.9 and 1.1
Calculated conductivity (umho/cm)	Should be between 0.55 and 0.7
Measured TDS	Should be between 0.55 and 0.7

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NOS	F	HCO3	SiO2
Concentration (mg/L)	68.3	2	115	20.6	13.4	111.432	87.3	0.01	0.22	408.7	38.5
Concentration (meq/L)	2.9710	0.0512	5.7385	1.6952	0.4799	3.1435	1.8176	0.0002	0.0116	6.6986	
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9884	61.0171	
Concentration (mM)	2.9710	0.0512	2.8693	0.8475	0.2400	3.1435	0.9088	0.0002	0.0116	6.6986	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	148.849605	3.76026	341.44075	90.0028008	25.919352	240.163149	145.40688	0.01151682	0.62999552	298.087389	
Ionic strength	0.0014852	2.558E-05	0.0057385	0.00169497	0.00047999	0.00157175	0.00181759	8.065E-08	5.7904E-06	0.0033493	
Cation sum (meq/L)	10.935738										
Anion sum (meq/L)	11.6714178										
% Difference	-3.25419007										
Ion Difference	-0.73567382										
Ratio: Cation sum*(100)/Measured conductivity	0.90579517										
Ratio: Anion sum*(100)/Measured conductivity	0.97099582										

Sample Location:

DE06

Alkalinity (mg/L)	307
Measured conductivity (umho/cm)	1320
Infinite dilution conductivity (umho/cm)	1480.14872
Ionic strength (M)	0.0179
Monovalent ion activity coefficient	0.8785
Calculated conductivity (umho/cm)	1142.23795
Measured TDS	707
Calculated TDS	713.733318
Ratio: Meas TDS/Calc TDS	0.99056606
Should be between 0.8 and 1.1	
Ratio: Calc cond/Meas cond	0.86533178
Should be between 0.9 and 1.1	
Ratio: Calc TDS/Calc cond	0.62485519
Should be between 0.55 and 0.7	
Ratio: Meas TDS/Meas cond	0.53560606
Should be between 0.55 and 0.7	

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	72.9	1.5	127	25.6	10.4	181	87.3	0.022	0.25	374.54	23.6	
Concentration (meq/L)	3.1712	0.0384	6.3373	2.1086	0.3724	5.1080	1.8176	0.0004	0.0132	6.1387		
Molecular weight (mg/mM)	22.9838	39.0983	40.0780	24.3050	55.8470	35.4527	96.0836	62.0049	18.9894	61.0171		
Concentration (mM)	3.1712	0.0384	3.1687	1.0532	0.1863	5.1080	0.9088	0.0004	0.0132	6.1387		
Charge z (absolute value)												
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	158.874615	2.820195	377.06935	111.848141	20.116512	390.099164	145.40688	0.025337	0.715904	273.172622		
Ionic strength	0.00158558	1.9155E-05	0.0063373	0.00210637	0.00037253	0.00285301	0.00181759	1.7743E-07	6.58E-06	0.00308936		

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum*(100)/Measured conductivity

Ratio: Anion sum*(100)/Measured conductivity

Sample Location:

DE07

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	247	838										
Measured conductivity (umho/cm)	863.518552											
Infinite dilution conductivity (umho/cm)	0.0108											
Ionic strength (M)	0.9005											
Monovalent ion activity coefficient	700.169133											
Calculated conductivity (umho/cm)	448											
Measured TDS	446.147878											
Calculated TDS	1.00415136											
Ratio: Meas TDS/Calc TDS	0.83551329											
Ratio: Calc cond/Meas cond	0.63720834											
Ratio: Calc TDS/Calc cond	0.53460631											
Ratio: Meas TDS/Meas cond												
Ionic strength												

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	42.2	0.8	83.4	8.3	14	68.331	54.1	0.048	0.2	301.34	26	
Concentration (meq/L)	1.8357	0.0205	4.1617	0.6830	0.5013	1.9445	1.1264	0.0008	0.0105	4.9390		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0836	62.0049	18.9894	61.0171		
Concentration (mM)	1.8357	0.0205	2.0808	0.3415	0.2507	1.9445	0.5632	0.0008	0.0105	4.9390		
Charge z (absolute value)												
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	91.96857	1.504104	247.61877	36.2632644	27.07992	148.563124	90.10896	0.05528074	0.5727232	219.783836		
Ionic strength	0.0091785	1.0232E-05	0.00416166	0.00068292	0.00050148	0.00097227	0.00112636	3.8712E-07	5.284E-06	0.00246948		

Cation sum (meq/L) 7.202171
 Anion sum (meq/L) 8.02117035
 % Difference -5.379892% Should be < 5%
 Ion Difference -0.81899935
 Ratio: Cation sum*(100)/Measured conductivity 0.85944761 Should be between 0.9 and 1.1
 Ratio: Anion sum*(100)/Measured conductivity 0.95718023 Should be between 0.9 and 1.1

Sample Location:

LN07

Alkalinity (mg/L)	147
Measured conductivity (umho/cm)	347
Infinite dilution conductivity (umho/cm)	348.745617
Ionic strength (M)	0.0048
Monovalent ion activity coefficient	0.9297
Calculated conductivity (umho/cm)	301.435237
Measured TDS	211
Calculated TDS	215.527478
Ratio: Meas TDS/Calc TDS	0.9789935
Ratio: Calc cond/Meas cond	0.76312718
Ratio: Meas TDS/Calc cond	0.71504426
Ratio: Meas TDS/Meas cond	0.53417722
Should be between 0.9 and 1.1	
Should be between 0.9 and 1.1	
Should be between 0.55 and 0.7	
Should be between 0.55 and 0.7	

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	7.9	0.9	38.9	0.03	26.7	12.153	3.3	0.063	0	179.34	37.4
Concentration (meq/L)	0.3437	0.0230	1.9411	0.0026	0.9561	0.3428	0.0687	0.0010	0.0000	2.9394	
Molecular weight (mg/mM)	22.9838	39.9983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.3437	0.0230	0.9706	0.0012	0.4782	0.3428	0.0344	0.0010	0.0000	2.9394	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	17.216825	1.692117	115.496045	0.13107204	51.645276	26.1926803	5.49848	0.07256597	0	130.802526	
Ionic strength	0.00017183	1.1511E-05	0.00194111	2.4664E-06	0.00095639	0.00017142	6.8706E-05	5.081E-07	0	0.00146969	
Cation sum (meq/L)	3.2663777										
Anion sum (meq/L)	3.35194092										
% Difference	-1.29282413										
Ion Difference											
Ratio: Cation sum*(100)/Measured conductivity	0.82863106										
Ratio: Anion sum*(100)/Measured conductivity	0.848659264										

Concentration (mg/L)	151
Measured conductivity (umho/cm)	406
Infinite dilution conductivity (umho/cm)	352.845129
Ionic strength (M)	0.0048
Monovalent ion activity coefficient	0.9298
Calculated conductivity (umho/cm)	305.017256
Measured TDS	207
Calculated TDS	210.112874
Ratio: Meas TDS/Calc TDS	0.98518428
Ratio: Calc cond/Meas cond	0.75127403
Ratio: Meas TDS/Calc cond	0.68885802
Ratio: Meas TDS/Meas cond	0.50985222
Should be between 0.9 and 1.1	
Should be between 0.9 and 1.1	
Should be between 0.55 and 0.7	
Should be between 0.55 and 0.7	

Sample Location:

DE08

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	11.1	1	42.4	0.03	18.1	10.743	4.9	0.059	0	184.22	31.2
Concentration (meq/L)	0.4829	0.0256	2.1158	0.0025	0.6482	0.3031	0.1020	0.0010	0.0000	3.0194	
Molecular weight (mg/mM)	22.9898	39.9983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.4829	0.0256	1.0579	0.0012	0.3242	0.3031	0.0510	0.0010	0.0000	3.0194	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	24.190785	1.88013	125.88772	0.13107204	35.010488	23.1537863	8.16144	0.0679424	0	134.361778	
Ionic strength	0.00024143	1.279E-05	0.00211576	2.4684E-06	0.00064834	0.00015153	0.00010202	4.7583E-07	0	0.00150968	
Cation sum (meq/L)	3.2748197										
Anion sum (meq/L)	3.4253955										
% Difference	-2.2432782										
Ion Difference											
Ratio: Cation sum*(100)/Measured conductivity	0.80660584										
Ratio: Anion sum*(100)/Measured conductivity	0.84369347										

Sample Location:

DE09

Alkalinity (mg/L)	300
Measured conductivity (umho/cm)	765
Infinite dilution conductivity (umho/cm)	722.162076
Ionic strength (M)	0.0095
Monovalent ion activity coefficient	0.9056
Calculated conductivity (umho/cm)	592.278182
Measured TDS	383.3072
Calculated TDS	0.98919835
Ratio: Meas TDS/Calc TDS	0.77421985
Ratio: Calc cond/Meas cond	0.64717427
Ratio: Meas TDS/Calc cond	0.50065359
Ratio: Meas TDS/Meas cond	0.50065359
Should be between 0.9 and 1.1	
Should be between 0.9 and 1.1	
Should be between 0.55 and 0.7	
Should be between 0.55 and 0.7	

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	26
Concentration (mg/L)	30.2	1.4	83.3	5.9	16.7	30.693	9.1	0.052	0	366		
Concentration (meq/L)	1.3137	0.0358	4.1567	0.4855	0.5980	0.8658	0.1895	0.0003	0.0000	5.9987		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0536	62.0049	18.9984	61.0171		
Concentration (mM)	1.3137	0.0358	2.0783	0.2427	0.2991	0.8658	0.0947	0.0003	0.0000	5.9987		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	65.81637	2.632182	247.321865	25.7775012	32.302476	66.1509041	15.15696	0.05988746	0	266.94393		
Ionic strength	0.00055685	1.7906E-05	0.00415667	0.00048545	0.00059819	0.00043292	0.00018946	4.1938E-07	0	0.00298937		
Cation sum (meq/L)	6.58972											
Anion sum (meq/L)	7.05489029											
% Difference	-3.40918707	Should be < 5%										
Ion Difference	-0.46517029											
Ratio: Cation sum*(100)/Measured conductivity	0.86140131	Should be between 0.9 and 1.1										
Ratio: Anion sum*(100)/Measured conductivity	0.92220788	Should be between 0.9 and 1.1										

Sample Location:

DE10

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	30.8
Concentration (mg/L)	28.3	1.2	51.3	3.6	11.6	29.52	26.5	0.055	0	213.5		
Concentration (meq/L)	1.2311	0.0307	2.5599	0.2962	0.4154	0.8328	0.5517	0.0009	0.0000	3.4993		
Molecular weight (mg/mM)	22.9988	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	1.2311	0.0307	1.2799	0.1481	0.2078	0.8328	0.2759	0.0009	0.0000	3.4993		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	61.67565	2.256156	152.312285	15.7286448	22.437648	63.6228029	44.1384	0.06334251	0	155.717293		
Ionic strength	0.00061553	1.5548E-05	0.00255987	0.00029621	0.00041551	0.00041638	0.00055173	4.4358E-07	0	0.00174963		
Cation sum (meq/L)	4.533256											
Anion sum (meq/L)	4.88464135											
% Difference	-3.73103822	Should be < 2%										
Ion Difference	-0.35138535											
Ratio: Cation sum*(100)/Measured conductivity	0.89767446	Should be between 0.9 and 1.1										
Ratio: Anion sum*(100)/Measured conductivity	0.96725571	Should be between 0.9 and 1.1										

Sample Location:

DE11

Alkalinity (mg/L)	239
Measured conductivity (umho/cm)	787
Infinite dilution conductivity (umho/cm)	840.827898
Ionic strength (M)	0.0103
Monovalent ion activity coefficient	0.9027
Calculated conductivity (umho/cm)	685.181216
Measured TDS	439
Calculated TDS	428.786146
Ratio: Meas TDS/Calc TDS	1.02362039 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.87062446 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.62579962 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.55781449 Should be between 0.55 and 0.7

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	44.8	1.7	79.9	8.2	12.3	87.163	30.9	0.052	0	279.38	26.4
Concentration (meq/L)	1.9488	0.0435	3.9870	0.6748	0.4405	2.4589	0.6433	0.0008	0.0000	4.5790	
Molecular weight (mg/mM)	22.9858	39.0983	40.0780	24.3050	55.6470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	1.9488	0.0435	1.9935	0.3373	0.2203	2.4589	0.3217	0.0008	0.0000	4.5790	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm ⁻² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	97.63488	3.196221	237.227095	35.8263576	23.791644	187.857533	51.48704	0.05988746	0	203.7672	
Ionic strength	0.0009744	2.1743E-05	0.003986701	0.0006747	0.00044059	0.00122943	0.00064334	4.1938E-07	0	0.00228952	

Cation sum (meq/L)

7.094557

Anion sum (meq/L)

7.68208319

% Difference

-3.97618794 Should be < 5%

Ion Difference

-0.587544619

Ratio: Cation sum*(100)/Measured conductivity

0.90146595 Should be between 0.9 and 1.1

Ratio: Anion sum*(100)/Measured conductivity

0.97612239 Should be between 0.9 and 1.1

DE12

Sample Location:

DE12

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	210										
Concentration (meq/L)	607										
Molecular weight (mg/mM)	592.671909										
Concentration (mM)	0.0079										
Monovalent ion activity coefficient	0.9127										
Calculated conductivity (umho/cm)	493.747358										
Measured TDS	331										
Calculated TDS	328.29354										
Ratio: Meas TDS/Calc TDS	1.00824402 Should be between 0.9 and 1.1										
Ratio: Calc cond/Meas cond	0.811342234 Should be between 0.9 and 1.1										
Ratio: Calc TDS/Calc cond	0.66490187 Should be between 0.55 and 0.7										
Ratio: Meas TDS/Meas cond	0.54530478 Should be between 0.55 and 0.7										

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	5.094508										
Concentration (meq/L)	5.680182388										
Molecular weight (mg/mM)	5.43566244 Should be < 5%										
Concentration (mM)	-5.43566244 Should be < 5%										
Charge z (absolute value)	-0.58567588										
Equivalent conductivity (mho-cm ⁻² /equivalent)	0.83929592 Should be between 0.9 and 1.1										
Infinite dilution conductivity (umho/cm)	0.93577988 Should be between 0.9 and 1.1										
Ionic strength	0.00046545	1.0232E-05	0.00323851	0.00032089	0.00058387	0.00028726	0.00050567	4.3551E-07	0	0.00209956	

Sample Location:

DE13

Alkalinity (mg/L)	263										
Measured conductivity (umho/cm)	656										
Infinite dilution conductivity (umho/cm)	653.503189										
Ionic strength (M)	0.00898										
Monovalent ion activity coefficient	0.9089										
Calculated conductivity (umho/cm)	539.857357										
Measured TDS	348										
Calculated TDS	349.5588812										
Ratio: Meas TDS/Calc TDS	0.98554049	Should be between 0.9 and 1.1									
Ratio: Calc cond/Meas cond	0.82420912	Should be between 0.9 and 1.1									
Ratio: Calc TDS/Calc cond	0.64750225	Should be between 0.55 and 0.7									
Ratio: Meas TDS/Meas cond	0.53129771	Should be between 0.55 and 0.7									
Constituent:											
Na	K	Ca	Mg	F ⁺	Cl	SO ₄	NO ₃	F	HCO ₃	SiO ₂	28.8
18.5	1.1	75.7	8	15.3	28.343	16	0.049	0	320.86		
0.8048	0.0281	3.7774	0.6583	0.5479	0.7986	0.3331	0.0008	0.0000	5.2589		
22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
0.8048	0.0281	1.8887	0.3291	0.2740	0.7986	0.1666	0.0008	0.0000	5.2589		
1	1	2	2	2	1	2	1	1	1	1	
50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
40.317975	2.068143	224.757085	34.952544	29.594484	61.0868087	26.6496	0.05643242	0	234.020845		
0.00040238	1.4069E-05	0.00377743	0.00063824	0.00054805	0.00039978	0.0003312	2.9519E-07	0	0.00292945		
Cation sum (meq/L)	5.816531										
Anion sum (meq/L)	6.39236118										
% Difference	-4.71648866	Should be < 5%									
Ion Difference	-0.5758388										
Ratio: Cation sum*(100)/Measured conductivity	0.88802	Should be between 0.9 and 1.1									
Ratio: Anion sum*(100)/Measured conductivity	0.9759331	Should be between 0.9 and 1.1									

Sample Location:

DE14

Alkalinity (mg/L)	259										
Measured conductivity (umho/cm)	865										
Infinite dilution conductivity (umho/cm)	849.871768										
Ionic strength (M)	0.01017										
Monovalent ion activity coefficient	0.9009										
Calculated conductivity (umho/cm)	689.3205712										
Measured TDS	452										
Calculated TDS	433.8044368										
Ratio: Meas TDS/Calc TDS	1.04194433	Should be between 0.9 and 1.1									
Ratio: Calc cond/Meas cond	0.79748043	Should be between 0.9 and 1.1									
Ratio: Calc TDS/Calc cond	0.62886551	Should be between 0.55 and 0.7									
Ratio: Meas TDS/Meas cond	0.52254335	Should be between 0.55 and 0.7									
Constituent:											
Na	K	Ca	Mg	F ⁺	Cl	SO ₄	NO ₃	F	HCO ₃	SiO ₂	29.5
31.3	1.2	89.6	13	1.5	73.789	38.5	0.048	0	315.98		
1.3616	0.0307	4.4710	1.0698	0.0537	2.0816	0.8016	0.0008	0.0000	5.1789		
22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
1.3616	0.0307	2.2385	0.5348	0.0269	2.0816	0.4008	0.0008	0.0000	5.1789		
1	1	2	2	2	1	2	1	1	1	1	
50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
68.21365	2.256156	266.02688	56.797884	2.90142	159.0333	64.1256	0.05528074	0	230.461593		
0.00068078	1.5348E-05	0.00447104	0.00106564	5.373E-05	0.00104079	0.00080157	3.8712E-07	0	0.00258946		
Cation sum (meq/L)	6.986771										
Anion sum (meq/L)	8.02844113										
% Difference	-7.15017042	Should be < 5%									
Ion Difference	-1.07607313										
Ratio: Cation sum*(100)/Measured conductivity	0.80771919	Should be between 0.9 and 1.1									
Ratio: Anion sum*(100)/Measured conductivity	0.93212011	Should be between 0.9 and 1.1									

Sample Location:

DE15

Alkalinity (mg/L)	300
Measured conductivity (umho/cm)	732
Infinite dilution conductivity (umho/cm)	755.345134
Ionic strength (M)	0.0102
Monovalent ion activity coefficient	0.9028
Calculated conductivity (umho/cm)	615.709189
Measured TDS	399
Calculated TDS	403.1982
Ratio: Meas TDS/Calc TDS	0.98958775
Ratio: Calc cond/Meas cond	0.84113277
Ratio: Calc TDS/Calc cond	0.65485169
Ratio: Meas TDS/Meas cond	0.54508197

Constituent:
Na K Ca Mg Fe Cl SO4 NO3 F HCO3 SiO2

Concentration (mg/L)	24.7	1.1	82.6	12.3	0.4512	21.113	37.7	0.043	0.28	366	30.8
Concentration (meq/L)	1.0745	0.0281	4.1217	1.0122	0.5956	0.7849	0.0007	0.0147	5.9987		
Molecular weight (mg/mM)	22.9845	39.0983	40.0780	24.3050	55.8470	35.4527	96.0536	62.0048	18.9984	61.0171	
Concentration (mM)	1.0745	0.0281	2.0609	0.5050	0.2257	0.5956	0.3525	0.0007	0.0147	5.9987	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	53.829945	2.068143	245.24353	53.7395364	24.371928	45.5036666	62.79312	0.04952233	0.80181248	266.94393	
Ionic strength	0.000053723	1.4069E-05	0.00412174	0.00101204	0.00045133	0.0002978	0.00078491	3.4679E-07	7.3698E-06	0.00299937	

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum^(100)/Measured conductivity

Ratio: Anion sum^(100)/Measured conductivity

Sample Location:

DR01

Constituent:
Na K Ca Mg Fe Cl SO4 NO3 F HCO3 SiO2

Concentration (mg/L)	124	332	306.497493	0.96674006	0.80843544	0.70912807	0.55421667	184	190.330376	0.96674006	0.80843544	0.70912807	0.55421667
Concentration (meq/L)	0.8961	0.0230	1.3074	0.0025	0.4906	0.3426	0.2249	0.0007	0.0174	0.0043	0.33	0.15128	0.312
Molecular weight (mg/mM)	22.9848	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171			
Concentration (mM)	0.8961	0.0230	0.6537	0.0012	0.2454	0.3426	0.1124	0.0007	0.0174	0.0043	0.33	0.15128	0.312
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	1		
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5			
Infinite dilution conductivity (umho/cm)	44.89461	1.692117	77.78911	0.13107204	26.499636	26.171279	17.98848	0.04952233	0.944993828	110.336824			
Ionic strength	0.00044805	1.1511E-05	0.00130738	2.4684E-06	0.00049073	0.00017128	0.00022486	3.4679E-07	8.6856E-06	0.00123974			

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum^(100)/Measured conductivity

Ratio: Anion sum^(100)/Measured conductivity

Sample Location:**JF19**

Alkalinity (mg/L)	274
Measured conductivity (umho/cm)	687
Infinite dilution conductivity (umho/cm)	734.865942
Ionic strength (M)	0.097
Monovalent ion activity coefficient	0.9048
Calculated conductivity (umho/cm)	601.64208
Measured TDS	387
Calculated TDS	379.554476
Ratio: Meas TDS/Calc TDS	1.01961648 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.87575266 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.63085424 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.563331878 Should be between 0.55 and 0.7

Concentration (mg/L)	28.9
Concentration (meq/L)	1.3007
Molecular weight (mg/MM)	22.9898
Concentration (mM)	39.0983
Charge z (absolute value)	1
Equivalent conductivity (mho-cm^-2/equivalent)	50.1
Infinite dilution conductivity (umho/cm)	65.162565
Ionic strength	1.1511E-05
Cation sum (meq/L)	6.5572976
Anion sum (meq/L)	7.02726887
% Difference	-3.459859712 Should be < 5%
Ion Difference	-0.46997127
Ratio: Cation sum/(100) /Measured conductivity	0.96446291 Should be between 0.9 and 1.1
Ratio: Anion sum/(100) /Measured conductivity	1.022859212 Should be between 0.9 and 1.1

Sample Location:**80**

Alkalinity (mg/L)	236
Measured conductivity (umho/cm)	622
Infinite dilution conductivity (umho/cm)	601.580916
Ionic strength (M)	0.0078
Monovalent ion activity coefficient	0.9133
Calculated conductivity (umho/cm)	501.798658
Measured TDS	333
Calculated TDS	332.974264
Ratio: Meas TDS/Calc TDS	1.00007729 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.80675026 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.66356149 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.53556977 Should be between 0.55 and 0.7

Constituent:	Na
Concentration (mg/L)	25.8
Concentration (meq/L)	1.1223
Molecular weight (mg/MM)	22.9898
Concentration (mM)	39.0983
Charge z (absolute value)	1
Equivalent conductivity (mho-cm^-2/equivalent)	50.1
Infinite dilution conductivity (umho/cm)	56.22723
Ionic strength	1.279E-05
Cation sum (meq/L)	5.203106
Anion sum (meq/L)	5.95850636
% Difference	-6.76784785 Should be < 5%
Ion Difference	-0.75540096
Ratio: Cation sum/(100) /Measured conductivity	0.83651222 Should be between 0.9 and 1.1
Ratio: Anion sum/(100) /Measured conductivity	0.95795932 Should be between 0.9 and 1.1

Constituent:	Na
Concentration (mg/L)	28.9
Concentration (meq/L)	1.3007
Molecular weight (mg/MM)	22.9898
Concentration (mM)	39.0983
Charge z (absolute value)	1
Equivalent conductivity (mho-cm^-2/equivalent)	50.1
Infinite dilution conductivity (umho/cm)	56.22723
Ionic strength	1.279E-05
Cation sum (meq/L)	5.203106
Anion sum (meq/L)	5.95850636
% Difference	-6.76784785 Should be < 5%
Ion Difference	-0.75540096
Ratio: Cation sum/(100) /Measured conductivity	0.83651222 Should be between 0.9 and 1.1
Ratio: Anion sum/(100) /Measured conductivity	0.95795932 Should be between 0.9 and 1.1

Sample Location:**DE17**

Alkalinity (mg/l)	184
Measured conductivity (umho/cm)	514
Infinite dilution conductivity (umho/cm)	509.218841
Ionic strength (M)	0.0065
Monovalent ion activity coefficient	0.9200
Calculated conductivity (umho/cm)	430.980851
Measured TDS	296
Calculated TDS	284.945816
Ratio: Meas TDS/Calc TDS	1.03879399 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.83848415 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.66115656 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.57587549 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2	
Concentration (mg/L)	25.4	1.2	50.4	2.3	12.3	30.96	22.3	0.039	0.17	224.48	29.5	
Concentration (meq/L)	1.1049	0.0307	2.5150	0.1893	0.4405	0.8734	0.4643	0.0006	0.0089	3.6792		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	1.1049	0.0307	1.2575	0.0946	0.2203	0.8734	0.2321	0.0006	0.0089	3.6792		
Charge z (absolute value)												
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	2	2	1	2	1	1	1	
Infinite dilution conductivity (umho/cm)	55.35459	2.256156	149.84072	10.04883564	23.791644	66.7263542	37.14288	0.0449156	0.48581472	163.72561		
Ionic strength	0.00055245	1.5348E-05	0.00251496	0.00018824	0.00044059	0.00043669	0.00046429	3.1454E-07	4.4744E-06	0.00183961		

Cation sum (meq/L)
Anion sum (meq/L)
% Difference
Ion Difference
Ratio: Cation sum*(100)/Measured conductivity
Ratio: Anion sum*(100)/Measured conductivity

4.280286
5.02647267
-8.01768582 Should be < 2%
-0.74618667
0.83274047 Should be between 0.9 and 1.1
0.97791297 Should be between 0.9 and 1.1

Sample Location:**DE18**

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2	
Concentration (mg/L)	24											
Concentration (meq/L)	473											
Molecular weight (mg/mM)	473.677529											
Concentration (mM)	0.0065											
Calculated conductivity (umho/cm)	400.931047											
Measured TDS	285											
Calculated TDS	261.578776											
Ratio: Meas TDS/Calc TDS	1.01307913 Should be between 0.9 and 1.1											
Ratio: Calc cond/Meas cond	0.84763435 Should be between 0.9 and 1.1											
Ratio: Calc TDS/Calc cond	0.65242834 Should be between 0.55 and 0.7											
Ratio: Meas TDS/Meas cond	0.5602537 Should be between 0.55 and 0.7											

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2	
Concentration (mg/L)	16.2	0.9	57.5	5.9	8.8	9.221	2.1	0.036	0.15	273.28	26.4	
Concentration (meq/L)	0.7047	0.0230	2.8693	0.4855	0.3151	0.2801	0.0437	0.0006	0.0079	4.4791		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.7047	0.0230	1.4346	0.2427	0.1576	0.2601	0.0219	0.0006	0.0079	4.4791		
Charge z (absolute value)												
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	2	2	1	2	1	1	1	
Infinite dilution conductivity (umho/cm)	35.30547	1.692117	170.720375	25.7775012	17.021664	19.8735049	3.49776	0.04146055	0.4295424	198.318134		
Ionic strength	0.00035295	1.1511E-05	0.00266925	0.00048545	0.00013096	0.00013522	0.00013096	4.3722E-05	2.9034E-07	3.948E-06	0.00223953	

Cation sum (meq/L)
Anion sum (meq/L)
% Difference
Ion Difference
Ratio: Cation sum*(100)/Measured conductivity
Ratio: Anion sum*(100)/Measured conductivity

4.397611
4.79138229
-4.28524961 Should be < 2%
-0.39377129
0.92972748 Should be between 0.9 and 1.1
1.01297723 Should be between 0.9 and 1.1

Sample Location:

DE19

Alkalinity (mg/L)	310
Measured conductivity (umho/cm)	887
Infinite dilution conductivity (umho/cm)	954.777915
Ionic strength (M)	0.0121
Monovalent ion activity coefficient	0.8958
Calculated conductivity (umho/cm)	766.176358
Measured TDS	512
Calculated TDS	491.08194
Ratio: Meas TDS/Calc TDS	1.04256986
Should be between 0.9 and 1.1	
Ratio: Calc cond/Meas cond	0.86378394
Should be between 0.9 and 1.1	
Ratio: Calc TDS/Calc cond	0.64095157
Should be between 0.55 and 0.7	
Ratio: Meas TDS/Meas cond	0.57722661
Should be between 0.55 and 0.7	

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	34.8	1.5	105	12.6	9.48	81.607	23.2	0.034	0.2	378.2	36.7	
Concentration (meq/L)	1.5138	0.0384	5.2395	1.0369	0.3395	2.3021	0.4830	0.0005	0.0105	6.1987		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.9470	35.4527	96.0536	62.0049	18.9984	61.0171		
Concentration (mM)	1.5138	0.0384	2.6198	0.5184	0.1698	2.3021	0.2415	0.0005	0.0105	6.1987		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	75.84138	2.820195	311.75025	55.0302568	18.3363744	175.882997	38.64192	0.03915719	0.5727232	275.842061		
Ionic strength	0.0007589	1.9185E-05	0.0052395	0.00103673	0.00033957	0.00115107	0.00046302	2.7421E-07	5.264E-06	0.00309935		

Cation sum (meq/L)	8.1680028
Anion sum (meq/L)	8.99493189
% Difference	-4.8181101
Ion Difference	-0.82692509
Ratio: Cation sum*(100)/Measured conductivity	0.92085714
Should be between 0.9 and 1.1	
Ratio: Anion sum*(100)/Measured conductivity	1.01408477
Should be between 0.9 and 1.1	

Sample Location:

DE20

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	322	906	999.5666834	999.5666834	0.0128	0.8934	797.749798	512	508.270428	1.04668869	Should be between 0.9 and 1.1	
Concentration (meq/L)	906	39.0983	0.0128	0.8934	797.749798	512	508.270428	1.04668869	Should be between 0.9 and 1.1	0.88051844	Should be between 0.9 and 1.1	
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.9470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	1.6139	0.0333	5.4890	1.3742	0.2167	2.2113	0.3050	0.0004	0.0132	6.4386		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	80.853865	2.444169	326.5955	72.963456	11.702394	168.940956	48.80208	0.02879205	0.715804	286.519818		
Ionic strength	0.00080633	1.6627E-05	0.005489	0.00137408	0.00021671	0.00110533	0.000061003	2.0165E-07	6.58E-06	0.00321932		

Cation sum (meq/L)	8.7269975
Anion sum (meq/L)	9.27350591
% Difference	-3.03607292
Ion Difference	-0.54630841
Ratio: Cation sum*(100)/Measured conductivity	0.96324476
Should be between 0.9 and 1.1	
Ratio: Anion sum*(100)/Measured conductivity	1.02356677
Should be between 0.9 and 1.1	

Sample Location:

DE21

Alkalinity (mg/L)	281
Measured conductivity (umho/cm)	6.05
Infinite dilution conductivity (umho/cm)	636.69497
Ionic strength (M)	0.0087
Monovalent ion activity coefficient	0.9092
Calculated conductivity (umho/cm)	526.34627
Measured TDS	375
Calculated TDS	353.54414
Ratio: Meas TDS/Calc TDS	1.060688
Ratio: Calc cond/Meas cond	0.8899388
Ratio: Calc TDS/Calc cond	0.67169488
Ratio: Meas TDS/Meas cond	0.61983471

Concentration (mg/L)	285
Concentration (meq/L)	0.9
Molecular weight (mg/mM)	72.9
Concentration (mM)	0.8970
Charge z (absolute value)	0.0230
Equivalent conductivity (mho-cm^2/equivalent)	3.6377
Infinite dilution conductivity (umho/cm)	40.0780
Ionic strength	24.3050
Cation sum (meq/L)	5.772766
Anion sum (meq/L)	6.19870131
% Difference	-3.557792067
Ion Difference	Should be < 5%
Ratio: Cation sum*(100)/Measured conductivity	0.9541762
Ratio: Anion sum*(100)/Measured conductivity	0.92457873
Ratio: Meas TDS/Meas cond	Should be between 0.9 and 1.1
Ratio: Meas TDS/Calc TDS	Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	Should be between 0.55 and 0.7
Ratio: Calc TDS/Calc cond	Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	Should be between 0.55 and 0.7

Concentration (mg/L)
Concentration (meq/L)
Molecular weight (mg/mM)
Concentration (mM)
Charge z (absolute value)
Equivalent conductivity (mho-cm^2/equivalent)
Infinite dilution conductivity (umho/cm)
Ionic strength

Cation sum (meq/L)
Anion sum (meq/L)
% Difference
Ion Difference
Ratio: Cation sum*(100)/Measured conductivity
Ratio: Anion sum*(100)/Measured conductivity
Ratio: Meas TDS/Meas cond

83

Sample Location:

PH01

Concentration (mg/L)	285
Concentration (meq/L)	583
Molecular weight (mg/mM)	632.786183
Concentration (mM)	0.0090
Monovalent ion activity coefficient	0.9078
Calculated conductivity (umho/cm)	521.478456
Measured TDS	338.44509
Calculated TDS	338.44509
Ratio: Meas TDS/Calc TDS	0.98865539
Ratio: Calc cond/Meas cond	0.8944742
Ratio: Calc TDS/Calc cond	0.64902986
Ratio: Meas TDS/Meas cond	0.57975986

Concentration (mg/L)
Concentration (meq/L)
Molecular weight (mg/mM)
Concentration (mM)
Charge z (absolute value)
Equivalent conductivity (mho-cm^2/equivalent)
Infinite dilution conductivity (umho/cm)
Ionic strength

Cation sum (meq/L)
Anion sum (meq/L)
% Difference
Ion Difference
Ratio: Cation sum*(100)/Measured conductivity
Ratio: Anion sum*(100)/Measured conductivity
Ratio: Meas TDS/Meas cond

Concentration (mg/L)	285
Concentration (meq/L)	0.6
Molecular weight (mg/mM)	67.2
Concentration (mM)	3.3533
Monovalent ion activity coefficient	0.1207
Calculated conductivity (umho/cm)	24.3050
Measured TDS	55.8470
Calculated TDS	55.4527
Ratio: Meas TDS/Calc TDS	0.96036
Ratio: Calc cond/Meas cond	0.8910092
Ratio: Calc TDS/Calc cond	0.77317632
Ratio: Meas TDS/Meas cond	0.7064E-06

Concentration (mg/L)
Concentration (meq/L)
Molecular weight (mg/mM)
Concentration (mM)
Charge z (absolute value)
Equivalent conductivity (mho-cm^2/equivalent)
Infinite dilution conductivity (umho/cm)
Ionic strength

Cation sum (meq/L)
Anion sum (meq/L)
% Difference
Ion Difference
Ratio: Cation sum*(100)/Measured conductivity
Ratio: Anion sum*(100)/Measured conductivity
Ratio: Meas TDS/Meas cond

Sample Location:

PH02

Alkalinity (mg/L)	230
Measured conductivity (umho/cm)	550
Infinite dilution conductivity (umho/cm)	569.604035
Ionic strength (M)	0.0083
Monovalent ion activity coefficient	0.9111
Calculated conductivity (umho/cm)	489.383275
Measured TDS	318
Calculated TDS	307.95302
Ratio: Meas TDS/Calc TDS	1.03587765 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.88978777 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.62926552 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.58 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2	34
Concentration (mg/L)	16.8	0.5	57	25.3	0.753	13.282	22	0.056	0.29	280.6		
Concentration (meq/L)	0.7308	0.0128	2.8443	2.0819	0.0270	0.3747	0.4580	0.0009	0.0153	4.5990		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.7308	0.0128	1.4222	1.0408	0.0135	0.3747	0.2290	0.0009	0.0153	4.5990		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm)^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	36.61308	0.940065	169.23585	110.53742	1.45651284	28.8255508	36.6432	0.06449419	0.83044864	204.657013		
Ionic strength	0.0003654	6.395E-06	0.00208168	2.6972E-05	0.00018734	0.00045804	4.5164E-07	7.6328E-06	0.00229952			

Cation sum (meq/L)	5.69679193
Anion sum (meq/L)	5.4479281
% Difference	2.23302002 Should be < 5%
Ion Difference	0.24886383
Ratio: Cation sum*(100)/Measured conductivity	1.03578035 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	0.99053238 Should be between 0.9 and 1.1

Sample Location:

PH03

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SIO2	32
Concentration (mg/L)	19.9	0.5	67.3	25.7	2.86	17.162	6.6	0.098	0.29	356.24		
Concentration (meq/L)	0.86557	0.0128	3.3583	2.1149	0.1024	0.4841	0.1374	0.0016	0.0153	5.8388		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.86557	0.0128	1.6791	1.0573	0.0512	0.4841	0.0687	0.0016	0.0153	5.8388		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm)^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	43.369065	0.940065	199.817065	112.285048	5.5320408	36.9882975	10.99296	0.11286484	0.83044864	259.822425		
Ionic strength	0.00044283	6.395E-06	0.000335827	0.0021146	0.000102407	0.00024207	0.00013741	7.99375E-07	7.6328E-06	0.00291939		

Cation sum (meq/L)	6.4539796
Anion sum (meq/L)	6.47717196
% Difference	-0.17935263 Should be < 5%
Ion Difference	-0.02319236
Ratio: Cation sum*(100)/Measured conductivity	1.0376173 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	1.04134597 Should be between 0.9 and 1.1

Sample Location:

PH04

Alkalinity (mg/L)	376
Measured conductivity (umho/cm)	743
Infinite dilution conductivity (umho/cm)	777.585304
Ionic strength (M)	0.0112
Monovalent ion activity coefficient	0.8992
Calculated conductivity (umho/cm)	628.708549
Measured TDS	404
Calculated TDS	407.448624
Ratio: Meas TDS/Calc TDS	0.99155605 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.84617301 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.64807441 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.54374159 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	8	1	93.3	25	8.85	3.025	7.8	0.051	0.27	458.72	34.6
Concentration (meq/L)	0.3480	0.0296	4.6557	2.0573	0.3169	0.0853	0.1624	0.0008	0.0142	7.5184	
Molecular weight (mg/mM)	22.9898	39.0933	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.3480	0.0296	2.3278	1.0285	0.1885	0.0853	0.0812	0.0008	0.0142	7.5184	
Charge z (absolute value)	1	1	2	2	2	2	2	1	1	1	
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	17.4348	1.88013	277.012365	109.2257	17.118378	6.5198131	12.99168	0.05873578	0.77317632	334.569726	
Ionic strength	0.000174	1.279E-05	0.00465567	0.002057	0.00031701	4.2668E-05	0.0001624	4.1131E-07	7.1064E-06	0.00375921	
Cation sum (meq/L)	7.4034185										
Anion sum (meq/L)	7.78119748										
% Difference	-2.48784183	Should be < 5%									

Ion Difference	-0.37776898
Ratio: Cation sum*(100)/Measured conductivity	0.98642241 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	1.04726615 Should be between 0.9 and 1.1

Sample Location:

PH05

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	22.5	0.8	86	30.9	2.31	21.228	66.5	0.05	0.29	375.76	37.9
Concentration (meq/L)	0.9788	0.0205	4.2914	2.5428	0.0827	0.5988	1.3845	0.0006	0.0153	6.1587	
Molecular weight (mg/mM)	22.9898	39.0933	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.9788	0.0205	2.1457	1.2712	0.0414	0.5988	0.6923	0.0008	0.0153	6.1587	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	49.035375	1.504104	255.3383	135.004201	4.4681868	45.7515196	110.7624	0.0575841	0.83044864	274.062435	
Ionic strength	0.0004938	1.0232E-05	0.0042914	0.00254245	8.2744E-05	0.00029894	0.00138453	4.0325E-07	7.9326E-06	0.00307935	
Cation sum (meq/L)	7.91669E-01										
Anion sum (meq/L)	8.15815038										
% Difference	-1.50565149	Should be < 5%									
Ion Difference	-0.24205428										
Ratio: Cation sum*(100)/Measured conductivity	1.00841988 Should be between 0.9 and 1.1										
Ratio: Anion sum*(100)/Measured conductivity	1.03925483 Should be between 0.9 and 1.1										

Sample Location:**PH06**

Alkalinity (mg/L)	300
Measured conductivity (umho/cm)	6.65
Infinite dilution conductivity (umho/cm)	674.229503
Ionic strength (M)	0.0096
Monovalent ion activity coefficient	0.9053
Calculated conductivity (umho/cm)	552.631945
Measured TDS	345
Calculated TDS	355.1682
Ratio: Meas TDS/Calc TDS	0.98323459
Ratio: Calc cond/Meas cond	0.83102548
Ratio: Calc TDS/Calc cond	0.64811346
Ratio: Meas TDS/Meas cond	0.51879699

Concentration (mg/L)	12.2
Concentration (meq/L)	0.0153
Molecular weight (mg/mM)	3.6128
Concentration (mM)	39.0983
Charge z (absolute value)	0.0153
Equivalent conductivity (mho-cm^2/equivalent)	1.8064
Infinite dilution conductivity (umho/cm)	24.3050
Ionic strength	0.5307
Constituent:	
Na	72.4
K	24
Ca	6.54
Mg	0.2342
Fe	0.1666
Cl	5.807
SO4	20.1
NO3	0.079
F	0.28
HCO3	366
SiO2	36.1

Cation sum (meq/L)	6.3679654
Anion sum (meq/L)	6.59987194
% Difference	-1.78832086
Ion Difference	Should be < 5%
Ratio: Cation sum*(100)/Measured conductivity	-0.23190634
Ratio: Anion sum*(100)/Measured conductivity	0.95758378
Ratio: Cation sum*(100)/Measured conductivity	0.99246195
Ratio: Anion sum*(100)/Measured conductivity	0.99246195

Sample Location:**PH07**

Alkalinity (mg/L)	254
Measured conductivity (umho/cm)	801
Infinite dilution conductivity (umho/cm)	859.340253
Ionic strength (M)	0.0125
Monovalent ion activity coefficient	0.8945
Calculated conductivity (umho/cm)	687.613375
Measured TDS	479
Calculated TDS	425.554936
Ratio: Meas TDS/Calc TDS	1.12558895
Ratio: Calc cond/Meas cond	0.85844366
Ratio: Calc TDS/Calc cond	0.61888703
Ratio: Meas TDS/Meas cond	0.590025

Constituent:	
Na	97.9
K	0.9
Ca	4.3852
Mg	36.1
Fe	0.1364
Cl	8.649
SO4	73.5
NO3	0.068
F	0.26
HCO3	309.88
SiO2	35.4

Cation sum (meq/L)	8.6986471
Anion sum (meq/L)	6.86787473
% Difference	11.7602418
Ion Difference	Should be < 5%
Ratio: Cation sum*(100)/Measured conductivity	1.83067237
Ratio: Anion sum*(100)/Measured conductivity	1.08597342
Ratio: Cation sum*(100)/Measured conductivity	0.8574256
Ratio: Anion sum*(100)/Measured conductivity	0.8574256

Sample Location:**PH08**

Alkalinity (mg/L)	310
Measured conductivity (umho/cm)	649
Infinite dilution conductivity (umho/cm)	731.639038
Ionic strength (M)	0.0102
Monovalent ion activity coefficient	0.9028
Calculated conductivity (umho/cm)	596.320731
Measured TDS	376
Calculated TDS	385.98494
Ratio: Meas TDS/Calc TDS	0.97413127 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.91883009 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.64727741 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.57935285 Should be between 0.55 and 0.7

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	15.7	1.4	83.1	22.3	3.45	13.44	22.9	0.304	0.23	378.2	37.2	
Concentration (meq/L)	0.6830	0.0358	4.1467	1.8351	0.1235	0.3791	0.4768	0.0048	0.0121	6.1987		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.6830	0.0358	2.0733	0.9174	0.0618	0.3791	0.2384	0.0049	0.0121	6.1987		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1		
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	34.215795	2.632182	246.729055	97.4302154	6.673266	28.9634794	38.14224	0.35011133	0.65863168	275.842061		
Ionic strength	0.00034148	1.7906E-05	0.00414669	0.00183484	0.00012358	0.00018957	0.00047678	2.4518E-06	6.0536E-06	0.00309935		

Cation sum (meq/L)	6.8240635
Anion sum (meq/L)	7.07162912
% Difference	-1.78158971 Should be < 5%
Ion Difference	-0.24756562
Ratio: Cation sum*(100)/Measured conductivity	1.05147357 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	1.08961928 Should be between 0.9 and 1.1

Sample Location:**PH09**

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	464											
Concentration (meq/L)	1020											
Molecular weight (mg/mM)	1215.16241											
Concentration (mM)	0.0175											
Monovalent ion activity coefficient	0.8794											
Calculated conductivity (umho/cm)	939.653892											
Measured TDS	575											
Calculated TDS	609.159536											
Ratio: Meas TDS/Calc TDS	0.94392335 Should be between 0.9 and 1.1											
Ratio: Calc cond/Meas cond	0.92122931 Should be between 0.9 and 1.1											
Ratio: Calc TDS/Calc cond	0.64828076 Should be between 0.55 and 0.7											
Ratio: Meas TDS/Meas cond	0.56372549 Should be between 0.55 and 0.7											

Constituent	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (mg/L)	13.3	1.1	121	50.5	5.97	4.828	101.2	0.05	0.27	566.08	32.6	
Concentration (meq/L)	0.5786	0.0281	6.0379	4.1556	0.2138	0.1382	2.1070	0.0008	0.0142	9.2781		
Molecular weight (mg/mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171		
Concentration (mM)	0.5786	0.0281	3.0196	2.0776	0.1069	0.1382	1.0835	0.0008	0.0142	9.2781		
Monovalent ion activity coefficient	1	1	2	2	2	1	2	1	1	1		
Calculated conductivity (umho/cm)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Measured TDS	28.985355	2.068143	359.25505	220.637934	11.5476516	10.405518	168.55872	0.0575841	0.77317632	412.873278		
Calculated TDS	0.000328928	1.4098E-05	0.006379	0.0015514	0.00021385	6.8099E-05	0.00210698	4.0325E-07	7.1084E-06	0.00463903		

Cation sum (meq/L)	11.0140187
Anion sum (meq/L)	11.5362524
% Difference	-2.31586431 Should be < 5%
Ion Difference	-0.52223388
Ratio: Cation sum*(100)/Measured conductivity	1.07980575 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	1.13100514 Should be between 0.9 and 1.1

Sample Location:

PH10

Alkalinity (mg/L)	322
Measured conductivity (umho/cm)	674
Infinite dilution conductivity (umho/cm)	716.099535
Ionic strength (M)	0.0101
Monovalent ion activity coefficient	0.9033
Calculated conductivity (umho/cm)	584.264293
Measured TDS	379
Calculated TDS	383.862428
Ratio: Meas TDS/Calc TDS	0.98733289 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.88685097 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.65700135 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.56231454 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	14.1	2.3	82.8	20.4	7.87	6.239	18.1	0.024	0.25	392.84	38.62
Concentration (meq/L)	0.61394	0.09868	4.13117	1.6787	0.2818	0.1760	0.3768	0.0004	0.0132	6.4386	
Molecular weight (mg/mm)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.61394	0.09868	2.0659	0.8393	0.1410	0.1760	0.1684	0.0004	0.0132	6.4386	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1	1	
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	58.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	30.728835	4.324299	245.83734	89.1259872	15.2227836	13.4465673	30.14736	0.02784037	0.715904	286.519818	
Ionic strength	0.00030668	2.9417E-05	0.00413172	0.00167851	0.0002619	8.8001E-05	0.00037684	1.9356E-07	6.59E-06	0.00321932	

Cation sum (meq/L)	6.7644447
Anion sum (meq/L)	7.00503891
% Difference	-1.74730117 Should be < 5%
Ion Difference	-0.24059421
Ratio: Cation sum*(100)/Measured conductivity	1.00362681 Should be between 0.9 and 1.1
Infinite dilution conductivity (umho/cm)	1.03932328 Should be between 0.9 and 1.1
Ionic strength	1.03932328 Should be between 0.9 and 1.1

Sample Location:

PH11

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	354										
Concentration (meq/L)	861										
Molecular weight (mg/mm)	1034.95437										
Concentration (mM)	0.0143										
Monovalent ion activity coefficient	0.8886										
Calculated conductivity (umho/cm)	817.154281										
Measured TDS	544										
Calculated TDS	541.820395										
Ratio: Meas TDS/Calc TDS	1.00402274 Should be between 0.9 and 1.1										
Ratio: Calc cond/Meas cond	0.94908743 Should be between 0.9 and 1.1										
Ratio: Calc TDS/Calc cond	0.66304953 Should be between 0.55 and 0.7										
Ratio: Meas TDS/Meas cond	0.63152236 Should be between 0.55 and 0.7										

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	28.4	1.9	105	25.1	8.72	13.105	112.4	0.01	0.26	431.88	34.57
Concentration (meq/L)	1.2354	0.0486	5.2395	2.0655	0.3123	0.3897	2.3402	0.0002	0.0137	7.0785	
Molecular weight (mg/mm)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	1.2354	0.0486	2.6198	1.0326	0.1562	0.3897	1.1701	0.0002	0.0137	7.0785	
Monovalent ion activity coefficient	1	1	2	2	1	2	1	1	1	1	
Calculated conductivity (umho/cm)	61.89354	3.572247	311.75025	109.663607	16.8663216	28.2444726	187.21344	0.01151682	0.74454016	314.993837	
Infinite dilution conductivity (umho/cm)	61.0006177	2.43015E-05	0.0052395	0.00206523	0.00031235	0.00019485	0.00234017	8.065E-08	6.8432E-06	0.00353926	
Ionic strength											

Cation sum (meq/L)	8.9012442
Anion sum (meq/L)	9.80222085
% Difference	-4.81716486 Should be < 5%
Ion Difference	-0.90057675
Ratio: Cation sum*(100)/Measured conductivity	1.03382627 Should be between 0.9 and 1.1
Ratio: Anion sum*(100)/Measured conductivity	1.1384633 Should be between 0.9 and 1.1

Sample Location:

PH12

Alkalinity [mg/l]	242
Measured conductivity (umho/cm)	493
Infinite dilution conductivity (umho/cm)	533.920822
Ionic strength (M)	0.0075
Monovalent ion activity coefficient	0.9147
Calculated conductivity (umho/cm)	446.703043
Measured TDS	285
Calculated TDS	283.932508
Ratio: Meas TDS/Calc TDS	0.96961034 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.90609136 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.65800427 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.57809331 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	9.8	1.5	62.5	6.79	6.198	13.6	0.24	295.24	34.61		
Concentration (meq/L)	0.4263	0.0384	3.1188	1.1109	0.2431	0.1748	0.2832	0.0004	0.0126		
Molecular weight (mg/MM)	22.9898	39.0983	40.0780	24.3050	55.4470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.4263	0.0384	1.5594	0.5554	0.1216	0.1748	0.1416	0.0004	0.0126		
Charge z (absolute value)	1	1	2	2	2	1	2	1	1		
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	21.35763	2.820195	185.565625	58.982418	13.1337612	13.3562023	22.65216	0.02879205	0.68726784	215.33477	
Ionic strength	0.00021315	1.9185E-05	0.00311875	0.00111078	0.00024322	8.7423E-05	0.00028315	2.0163E-07	6.3168E-06	0.00241949	

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum*(100)/Measured conductivity

Ratio: Anion sum*(100)/Measured conductivity

1.00151824 Should be between 0.9 and 1.1

1.07708276 Should be between 0.9 and 1.1

Sample Location:

PH13

Concentration (mg/L)	250
Measured conductivity (umho/cm)	597
Infinite dilution conductivity (umho/cm)	673.491062
Ionic strength (M)	0.0094
Monovalent ion activity coefficient	0.9083
Calculated conductivity (umho/cm)	563.217552
Measured TDS	358
Calculated TDS	361.3755
Ratio: Meas TDS/Calc TDS	0.9906593 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.92666257 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.65322494 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.59966499 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	22.5	1.4	71.5	18.8	1.87	6.654	51.9	0.023	0.18	305	
Concentration (meq/L)	0.9788	0.0358	3.5679	1.5471	0.0670	0.1877	1.0806	0.0004	0.0095	4.9990	
Molecular weight (mg/MM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.9788	0.0358	1.7839	0.7734	0.0335	0.1877	0.5403	0.0004	0.0095	4.9990	
Charge z (absolute value)	1	1	2	2	2	1	2	1	1		
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	49.035375	2.632182	212.287075	82.1384784	3.6171036	14.3409836	86.44464	0.02648869	0.51545088	222.453275	
Ionic strength	0.00048938	1.7906E-05	0.00356785	0.00154686	6.6983E-05	9.3855E-05	0.00108056	1.8555E-07	4.7376E-06	0.00249948	

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum*(100)/Measured conductivity

Ratio: Anion sum*(100)/Measured conductivity

1.03792776 Should be between 0.9 and 1.1

1.05143443 Should be between 0.9 and 1.1

Sample Location:**PH14**

Alkalinity (mg/L)	282
Measured conductivity (umho/cm)	6365
Infinite dilution conductivity (umho/cm)	812.724896
Ionic strength (M)	0.0114
Monovalent ion activity coefficient	0.8965
Calculated conductivity (umho/cm)	656.108649
Measured TDS	432
Calculated TDS	425.308488
Ratio: Meas TDS/Calc TDS	1.01573366
Ratio: Calc cond/Meas cond	Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.94404122
Ratio: Meas TDS/Meas cond	0.64822872
	Should be between 0.55 and 0.7
	0.62158273
	Should be between 0.55 and 0.7

Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm^2/equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	7.2161612
Anion sum (meq/L)	7.53627154
% Difference	-2.16988171 Should be < 5%
Ion Difference	-0.32011034
Ratio: Cation sum^(100)/Measured conductivity	1.03929638
Ratio: Anion sum^(100)/Measured conductivity	1.08435562
	Should be between 0.9 and 1.1

Sample Location:**PH15**

Alkalinity (mg/L)	614
Measured conductivity (umho/cm)	1103
Infinite dilution conductivity (umho/cm)	1188.22389
Ionic strength (M)	0.0168
Monovalent ion activity coefficient	0.8813
Calculated conductivity (umho/cm)	922.397382
Measured TDS	627
Calculated TDS	622.882836
Ratio: Meas TDS/Calc TDS	1.00661018
Ratio: Calc cond/Meas cond	Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.83671567
Ratio: Meas TDS/Meas cond	0.6749208
	Should be between 0.55 and 0.7
	0.56844968
	Should be between 0.55 and 0.7

Sample Location:**PH16**

Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm^2/equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	10.890397
Anion sum (meq/L)	12.5303666
% Difference	-7.00220388 Should be < 5%
Ion Difference	-1.63896962
Ratio: Cation sum^(100)/Measured conductivity	0.98734334
Ratio: Anion sum^(100)/Measured conductivity	1.13602599
	Should be between 0.9 and 1.1
	1.13602599 Should be between 0.9 and 1.1

Sample Location:**PH16**

Alkalinity (mg/L)	416
Measured conductivity (umho/cm)	1054
Infinite dilution conductivity (umho/cm)	1255.4689
Ionic strength (M)	0.0179
Monovalent ion activity coefficient	0.8784
Calculated conductivity (umho/cm)	999.505145
Measured TDS	889
Calculated TDS	659.829584
Ratio: Meas TDS/Calc TDS	1.01369816
Ratio: Calc cond/Meas cond	0.93938453
Ratio: Meas TDS/Calc cond	0.66015627
Ratio: Calc TDS/Meas cond	0.6267594
Ratio: Meas TDS/Meas cond	0.7

Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm ⁻² /equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	11.4111963
Anion sum (meq/L)	11.86868745
% Difference	-1.97017664
Ion Difference	< 5%
Ratio: Cation sum/(100)/Measured conductivity	1.07248606
Ratio: Anion sum/(100)/Measured conductivity	1.11558971

Sample Location:**PH17**

Concentration (mg/L)	402
Concentration (meq/L)	849
Molecular weight (mg/mM)	809.376842
Concentration (mM)	0.0114
Monovalent ion activity coefficient	0.8982
Calculated conductivity (umho/cm)	653.018001
Measured TDS	424
Calculated TDS	439.812348
Ratio: Meas TDS/Calc TDS	0.96404751
Ratio: Calc cond/Meas cond	0.76916137
Ratio: Meas TDS/Calc cond	0.67350723
Ratio: Calc TDS/Meas cond	0.49941107
Ratio: Meas TDS/Meas cond	0.7

Cation sum (meq/L)	849
Anion sum (meq/L)	809.376842
% Difference	< 5%
Ion Difference	
Ratio: Cation sum/(100)/Measured conductivity	1.07248606
Ratio: Anion sum/(100)/Measured conductivity	1.11558971

Sample Location:**PH17**

Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm ⁻² /equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	7.438103
Anion sum (meq/L)	8.30812623
% Difference	-5.52527985
Ion Difference	Should be < 5%
Ratio: Cation sum/(100)/Measured conductivity	0.87871016
Ratio: Anion sum/(100)/Measured conductivity	0.97857788

Constituent:	
Na	
K	
Ca	
Mg	
Fe	
Cl	
SO4	
NO3	
F	
HCO3	
SiO2	
Concentration (mg/L)	402
Concentration (meq/L)	849
Molecular weight (mg/mM)	809.376842
Concentration (mM)	0.0114
Monovalent ion activity coefficient	0.8982
Calculated conductivity (umho/cm)	653.018001
Measured TDS	424
Calculated TDS	439.812348
Ratio: Meas TDS/Calc TDS	0.96404751
Ratio: Calc cond/Meas cond	0.76916137
Ratio: Meas TDS/Calc cond	0.67350723
Ratio: Calc TDS/Meas cond	0.49941107
Ratio: Meas TDS/Meas cond	0.7
Cation sum (meq/L)	7.438103
Anion sum (meq/L)	8.30812623
% Difference	-5.52527985
Ion Difference	Should be < 5%
Ratio: Cation sum/(100)/Measured conductivity	0.87871016
Ratio: Anion sum/(100)/Measured conductivity	0.97857788

Constituent:	
Na	
K	
Ca	
Mg	
Fe	
Cl	
SO4	
NO3	
F	
HCO3	
SiO2	
Concentration (mg/L)	402
Concentration (meq/L)	849
Molecular weight (mg/mM)	809.376842
Concentration (mM)	0.0114
Monovalent ion activity coefficient	0.8982
Calculated conductivity (umho/cm)	653.018001
Measured TDS	424
Calculated TDS	439.812348
Ratio: Meas TDS/Calc TDS	0.96404751
Ratio: Calc cond/Meas cond	0.76916137
Ratio: Meas TDS/Calc cond	0.67350723
Ratio: Calc TDS/Meas cond	0.49941107
Ratio: Meas TDS/Meas cond	0.7
Cation sum (meq/L)	7.438103
Anion sum (meq/L)	8.30812623
% Difference	-5.52527985
Ion Difference	Should be < 5%
Ratio: Cation sum/(100)/Measured conductivity	0.87871016
Ratio: Anion sum/(100)/Measured conductivity	0.97857788

Sample Location: PH18

Alkalinity (mg/L)	322
Measured conductivity (umho/cm)	643
Infinite dilution conductivity (umho/cm)	656.137746
Ionic strength (M)	0.0089
Monovalent ion activity coefficient	0.9082
Calculated conductivity (umho/cm)	541.224515
Measured TDS	361
Calculated TDS	377.576428
Ratio: Meas TDS/Calc TDS	0.95609762 Should be between 0.9 and 1.1
Ratio: Calc cond/Meas cond	0.84171775 Should be between 0.9 and 1.1
Ratio: Calc TDS/Calc cond	0.69763384 Should be between 0.55 and 0.7
Ratio: Meas TDS/Meas cond	0.56143079 Should be between 0.55 and 0.7

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	21.6	3.1	76.4	4.5	20.5	8.194	7.2	0.023	0.31	392.84	42.59
Concentration (meq/L)	0.9396	0.0793	3.8124	0.3703	0.7341	0.2312	0.1499	0.0004	0.0163	6.4386	
Molecular weight (mg/mM)	22.9836	39.0983	40.0780	24.3050	55.8470	35.4577	36.0636	62.0049	18.9984	61.0171	
Concentration (mM)	0.9396	0.0793	1.9062	0.1851	0.3672	0.2312	0.0750	0.0004	0.0163	6.4386	
Charge z (absolute value)	1	1	2	2	1	2	1	2	1	1	1
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	11.95232	0.026488869	0.88772096	286.519818	
Infinite dilution conductivity (umho/cm)	47.0736	5.828403	226.83542	19.660806	39.65274	17.6600693	0.0001499	1.855E-07	8.1592E-06	0.00321932	
Ionic strength	0.0004698	3.9649E-05	0.00381286	0.00037026	0.000173431	0.00011558	0.0001499				
Cation sum (meq/L)	5.935668										
Anion sum (meq/L)	6.83639373										
% Difference	-7.05231269	Should be < 5%									
Ion Difference	-0.90072573										
Ratio: Cation sum/(100)/Measured conductivity	0.923121	Should be between 0.9 and 1.1									
Ratio: Anion sum/(100)/Measured conductivity	1.06320276	Should be between 0.9 and 1.1									

Sample Location:

PH18

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	194										
Concentration (meq/L)	397										
Molecular weight (mg/mM)	417.222376										
Concentration (mM)	0.0059										
Monovalent ion activity coefficient	0.9230										
Calculated conductivity (umho/cm)	355.464331										
Measured TDS	227										
Calculated TDS	243.75356										
Ratio: Meas TDS/Calc TDS	0.93126847 Should be between 0.9 and 1.1										
Ratio: Calc cond/Meas cond	0.898337615 Should be between 0.9 and 1.1										
Ratio: Calc TDS/Calc cond	0.68573281 Should be between 0.55 and 0.7										
Ratio: Meas TDS/Meas cond	0.57718841 Should be between 0.55 and 0.7										

Constituent:	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2
Concentration (mg/L)	3.8702536										
Concentration (meq/L)	4.17427178										
Molecular weight (mg/mM)	3.7791935 Should be < 2%										
Concentration (mM)	-0.310401818										
Charge z (absolute value)	1	1	2	2	1	2	1	2	1	1	1
Equivalent conductivity (mho-cm^2/equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5	
Infinite dilution conductivity (umho/cm)	13.0761	4.136286	144.88984	50.6811888	8.8203168	5.76096721	16.48944	0.02879205	0.715904	172.623741	
Ionic strength	0.0001305	2.8138E-05	0.00243512	0.00095445	0.00016334	3.7703E-05	0.00020612	2.0163E-07	6.58E-06	0.00193959	
Cation sum (meq/L)											
Anion sum (meq/L)											
% Difference											
Ion Difference											
Ratio: Cation sum/(100)/Measured conductivity	0.97487496 Should be between 0.9 and 1.1										
Ratio: Anion sum/(100)/Measured conductivity	1.05145385 Should be between 0.9 and 1.1										

Sample Location:

	PH20
Alkalinity (mg/L)	358
Measured conductivity (umho/cm)	647
Infinite dilution conductivity (umho/cm)	765.15815
Ionic strength (M)	0.0111
Monovalent ion activity coefficient	0.8996
Calculated conductivity (umho/cm)	620.098325
Measured TDS	353
Calculated TDS	386.797782
Ratio: Meas TDS/Calc TDS	0.9902058
Ratio: Calc cond/Meas cond	0.9584209
Ratio: Meas TDS/Calc cond	0.62375252
Ratio: Meas TDS/Meas cond	0.59196281
Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm ⁻² /equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	7.4333405
Anion sum (meq/L)	7.49208363
% Difference	-0.39357763
Ion Difference	Should be < 5%
Ratio: Cation sum*(100)/Measured conductivity	1.14889343
Ratio: Anion sum*(100)/Measured conductivity	1.15797274
Should be between 0.9 and 1.1	

PH21

	PH21
Alkalinity (mg/L)	332
Measured conductivity (umho/cm)	620
Infinite dilution conductivity (umho/cm)	712.507643
Ionic strength (M)	0.0103
Monovalent ion activity coefficient	0.9026
Calculated conductivity (umho/cm)	580.431567
Measured TDS	359
Calculated TDS	367.314468
Ratio: Meas TDS/Calc TDS	0.97736497
Ratio: Calc cond/Meas cond	0.93617995
Ratio: Meas TDS/Calc cond	0.63282941
Ratio: Meas TDS/Meas cond	0.57903226
Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm ⁻² /equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	6.9747068
Anion sum (meq/L)	6.90089344
% Difference	0.53196373
Ion Difference	Should be < 5%
Ratio: Cation sum*(100)/Measured conductivity	1.12495568
Ratio: Anion sum*(100)/Measured conductivity	1.11304733
Should be between 0.9 and 1.1	

Sample Location:**PH22**

Alkalinity (mg/L)	436
Measured conductivity (umho/cm)	819
Infinite dilution conductivity (umho/cm)	946.939207
Ionic strength (M)	0.0136
Monovalent ion activity coefficient	0.8909
Calculated conductivity (umho/cm)	751.561899
Measured TDS	471
Calculated TDS	482.639064
Ratio: Meas TDS/Calc TDS	0.97578345
Ratio: Calc cond/Meas cond	0.91765793
Ratio: Calc TDS/Calc cond	0.64224793
Ratio: Meas TDS/Meas cond	0.57509153

Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm^2/equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	8.999623
Anion sum (meq/L)	9.22561414
% Difference	-1.24107368
Ion Difference	Should be < 5%
Ratio: Cation sum*(100)/Measured conductivity	-0.22619114
Ratio: Anion sum*(100)/Measured conductivity	1.09885557
Ratio: Meas TDS/Calc TDS	1.2647303

Sample Location:

94

Alkalinity (mg/L)	376
Measured conductivity (umho/cm)	696
Infinite dilution conductivity (umho/cm)	781.858492
Ionic strength (M)	0.0113
Monovalent ion activity coefficient	0.8989
Calculated conductivity (umho/cm)	631.786919
Measured TDS	395
Calculated TDS	412.660824
Ratio: Meas TDS/Calc TDS	0.95720303
Ratio: Calc cond/Meas cond	0.90773984
Ratio: Calc TDS/Calc cond	0.65316423
Ratio: Meas TDS/Meas cond	0.56752874

Sample Location:**PH23**

Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm^2/equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	7.496455
Anion sum (meq/L)	7.6924168
% Difference	-1.29016892
Ion Difference	Should be < 5%
Ratio: Cation sum*(100)/Measured conductivity	-0.1959618
Ratio: Anion sum*(100)/Measured conductivity	1.07707687
Ratio: Meas TDS/Calc TDS	1.1052323

Concentration (mg/L)	
Concentration (meq/L)	
Molecular weight (mg/mM)	
Concentration (mM)	
Charge z (absolute value)	
Equivalent conductivity (mho-cm^2/equivalent)	
Infinite dilution conductivity (umho/cm)	
Ionic strength	
Cation sum (meq/L)	7.496455
Anion sum (meq/L)	7.6924168
% Difference	-0.1959618
Ion Difference	Should be between 0.9 and 1.1
Ratio: Cation sum*(100)/Measured conductivity	1.07707687
Ratio: Anion sum*(100)/Measured conductivity	1.1052323

Sample Location:

PH24

Alkalinity (mg/L)	376
Measured conductivity (umho/cm)	698
Infinite dilution conductivity (umho/cm)	824.966937
Ionic strength (M)	0.0118
Nonvalent ion activity coefficient	0.8969
Calculated conductivity (umho/cm)	663.683683
Measured TDS	414
Calculated TDS	427.550674
Ratio: Meas TDS/Calc TDS	0.96830639
Ratio: Calc cond/Meas cond	0.95083622
Ratio: Calc TDS/Calc cond	0.64420843
Ratio: Meas TDS/Meas cond	0.59312321
Should be between 0.9 and 1.1	
Should be between 0.9 and 1.1	
Should be between 0.55 and 0.7	
Should be between 0.55 and 0.7	

Constituent:

Concentration (mg/L)	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (meq/L)	13.9	1.8	106	24	3.8	3.6	12.9	0.028	0.18	458.72	35.79	
Molecular weight (mg/mM)	0.6047	0.0460	5.2884	1.9750	0.1361	0.1016	0.2686	0.0005	0.0095	7.5184		
Concentration (mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	98.0636	62.0049	18.9884	61.0171		
Charge z (absolute value)	0.6047	0.0460	2.6447	0.9874	0.0681	0.1016	0.1343	0.0005	0.0095	7.5184		
Equivalent conductivity (mho-cm^2/equivalent)	1	1	2	2	1	2	1	1	1	1		
Infinite dilution conductivity (umho/cm)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Ionic strength	30.292965	3.384234	314.7193	104.857632	7.350264	7.7588784	21.48624	0.032471	0.51545088	334.559726		
0.00030233	2.3022E-05	0.0052884	0.00197472	0.00013612	5.0778E-05	0.00028858	2.2582E-07	4.7376E-06	0.00375921			

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum^(100)/Measured conductivity

Ratio: Anion sum^(100)/Measured conductivity

Should be between 0.9 and 1.1

Should be between 0.9 and 1.1

Sample Location:

PH25

Alkalinity (mg/L)	328
Measured conductivity (umho/cm)	689
Infinite dilution conductivity (umho/cm)	854.55277
Ionic strength (M)	0.0123
Nonvalent ion activity coefficient	0.8953
Calculated conductivity (umho/cm)	684.817513
Measured TDS	423
Calculated TDS	444.773872
Ratio: Meas TDS/Calc TDS	0.9510455
Ratio: Calc cond/Meas cond	0.99392963
Ratio: Calc TDS/Calc cond	0.6494776
Ratio: Meas TDS/Meas cond	0.61393324
Should be between 0.9 and 1.1	
Should be between 0.9 and 1.1	
Should be between 0.55 and 0.7	
Should be between 0.55 and 0.7	

Constituent:

Concentration (mg/L)	Na	K	Ca	Mg	Fe	Cl	SO4	NO3	F	HCO3	SiO2	
Concentration (meq/L)	10	1.2	110	22.1	5.11	3.261	58.7	0.034	0.2	400.16	37.41	
Molecular weight (mg/mM)	0.4350	0.0307	5.4880	1.8186	0.1830	0.0920	1.2221	0.0005	0.0105	6.5586		
Concentration (mM)	22.9898	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9884	61.0171		
Charge z (absolute value)	0.4350	0.0307	2.7445	0.9092	0.0815	0.0920	0.6111	0.0005	0.0105	6.5586		
Equivalent conductivity (mho-cm^2/equivalent)	1	1	2	2	1	2	1	1	1	1		
Infinite dilution conductivity (umho/cm)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
0.0002176	2.256156	326.5985	98.5564028	9.8841708	7.02825068	97.77072	0.03915719	0.5727232	291.888697			
0.0002176	1.5348E-05	0.005489	0.00181839	0.00018304	4.5996E-05	0.00122213	2.7421E-07	5.264E-06	0.00327931			

Cation sum (meq/L)

Anion sum (meq/L)

% Difference

Ion Difference

Ratio: Cation sum^(100)/Measured conductivity

Ratio: Anion sum^(100)/Measured conductivity

Should be between 0.9 and 1.1

Should be between 0.9 and 1.1

Sample Location:

	PH26
Alkalinity (mg/L)	488
Measured conductivity (umho/cm)	928
Infinite dilution conductivity (umho/cm)	1174.43148
Ionic strength (I)	0.0167
Monovalent ion activity coefficient	0.8815
Calculated conductivity (umho/cm)	912.816181
Measured TDS	576
Calculated TDS	593.345512
Ratio: Meas TDS/Calc TDS	0.97076659
Ratio: Calc cond/Meas cond	0.98554663
Ratio: Meas TDS/Calc cond	0.65015888
Ratio: Meas TDS/Meas cond	0.62203024

Constituent:	Na	K	Ca	Mg	F _e	Cl	SO ₄	NO ₃	F	HCO ₃	SiO ₂	33.45
Concentration (mg/L)	23.6	1.4	145	33.5	2.86	4.607	56	0.01	0.18	595.36		
Concentration (meq/L)	1.0286	0.0358	7.2355	2.7567	0.1024	0.1300	1.1659	0.0002	0.0095	9.7580		
Molecular weight (mg/mM)	22.9858	39.0983	40.0780	24.3050	55.8470	35.4527	96.0636	62.0049	18.9884	61.0171		
Concentration (mM)	1.0286	0.0358	3.6178	1.3782	0.0512	0.1300	0.5830	0.0002	0.0095	9.7580		
Charge z (absolute value)	1	1	2	2	1	2	1	2	1	1		
Equivalent conductivity (mho-cm ² /equivalent)	50.1	73.5	59.5	53.1	54	76.4	80	71.4	54.4	44.5		
Infinite dilution conductivity (umho/cm)	51.43268	2.632182	430.51225	146.363778	5.5320408	9.92920911	93.2736	0.01151682	0.51545068	434.228793		
Ionic strength	0.0005133	1.7906E-05	0.00723855	0.002756538	0.00010245	6.4982E-05	0.00116592	8.065E-08	4.7376E-06	0.00487893		

Cation sum (meq/L)	11.1570436
Anion sum (meq/L)	11.08334704
% Difference	0.42111191
Ion Difference	0.09357323
Ratio: Cation sum/(100)Measured conductivity	1.20486432
Ratio: Anion sum/(100)Measured conductivity	1.19475922